

Review of Physics 2 - Exam, muster

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For every task is for correct general result 1 point, for correct numerical result 1 point and correct way of solution for 3 points, i.e. maximum of possible points is 5 per task and maximum 20 points for the test. Numerical results estimate with the 1-digit of precision.

Task 1 - Water power plant

Calculate the difference in the river levels before and after the hydroelectric power station, where the current flow rate is $I = 400 \text{ m}^3\text{s}^{-1}$, the output power is $P = 2 \text{ MW}$, and efficiency of the power plant is $\eta = 75 \%$. Assume gravitational acceleration as $g = 10 \text{ m s}^{-2}$.

Solution:

$$\Delta W = mg\Delta h;$$

$$P_{\text{water}} = \frac{\Delta W}{t} = \frac{mg\Delta h}{t} = \rho I g \Delta h;$$

$$P_{\text{out}} = \eta P_{\text{water}};$$

$$\Delta h = \frac{P_{\text{out}}}{I \rho g \eta} = \frac{2 \cdot 10^6}{400 \cdot 10 \cdot 1000 \cdot 0.75} = \frac{2}{3} \text{ m}.$$

Task 2 - Linear harmonic oscillator

The linear harmonic oscillator with the amplitude of displacement $y_0 = 12 \text{ cm}$ has a period of movement $T = 40 \text{ ms}$. The rigidity of the oscillating system is $k = 6 \text{ Nm}^{-1}$. Calculate the total mass, maximal velocity, maximal acceleration, and total energy of the oscillator.

Solution:

$$m = \frac{k}{\omega^2} = \frac{T^2 k}{4\pi^2} = \frac{0.04 \cdot 0.04 \cdot 6}{40} = 24 \cdot 10^{-5} \text{ kg} = 210 \text{ mg};$$

$$v_{\text{max}} = \omega y_0 = \frac{2\pi}{T} y_0 \approx \frac{2 \cdot 3.14}{0.04} \cdot 0.12 = 2 \cdot 3.14 \cdot 3 \approx 19 \text{ m s}^{-1};$$

$$a_{\text{max}} = \omega v_{\text{max}} = \omega^2 y_0 = \frac{4\pi^2}{T^2} y_0 \approx \frac{2 \cdot 3.14}{0.04} \cdot 19 \approx 3000 \text{ m s}^{-2};$$

$$E_{\text{tot}} = E_{\text{k max}} = \frac{1}{2} \frac{k}{\omega^2} (\omega y_0)^2 = \frac{1}{2} k y_0^2 = \frac{1}{2} \cdot 6 \cdot 0.12^2 \approx 3 \cdot 0.0144 \approx 0.04 \text{ J}.$$

Task 3 - Capacitors

Three capacitors with capacities $C_1 = 1 \mu\text{F}$, $C_2 = 3 \mu\text{F}$ and $C_3 = 20 \mu\text{F}$ are connected serial. What is the total capacity? What is the total bound electric charge, if the capacities are charged to voltage $U = 200 \text{ V}$? how is the voltage divided into individual capacities?

Solution:

$$C = \frac{1}{\frac{1}{C_1} + \frac{1}{C_2} + \frac{1}{C_3}} = \frac{1}{\frac{1}{\mu\text{F}} \left(\frac{1}{1} + \frac{1}{3} + \frac{1}{20} \right)} = \frac{1\mu\text{F}}{\frac{60+20+3}{60}} = \frac{60}{83} \mu\text{F};$$

$$Q = CU = \left(\frac{60}{83} \mu\text{F} \right) \cdot 200 \text{ V} \approx 145 \mu\text{C};$$

$$U_1 = \frac{Q}{C_1} = \frac{UC}{C_1} = U \frac{\frac{1}{\frac{1}{C_1} + \frac{1}{C_2} + \frac{1}{C_3}}}{C_1} = U \frac{1}{\frac{C_1}{C_1} + \frac{C_1}{C_2} + \frac{C_1}{C_3}} = 200 \frac{1}{1 + \frac{1}{3} + \frac{1}{20}} = 200 \frac{60}{83} \approx 145 \text{ V};$$

$$U_2 = U \frac{1}{\frac{C_2}{C_1} + \frac{C_2}{C_2} + \frac{C_2}{C_3}} = 200 \frac{1}{3 + 1 + \frac{3}{20}} = 200 \frac{20}{83} \approx 48 \text{ V};$$

$$U_3 = U \frac{1}{\frac{C_3}{C_1} + \frac{C_3}{C_2} + \frac{C_3}{C_3}} = 200 \frac{1}{20 + \frac{20}{3} + 1} = 200 \frac{3}{81} \approx 7 \text{ V};$$

Task 4 - Water vapor

Inside the closed glass tube with a volume of $V = 3 \text{ cm}^3$ is located pure water vapor with the pressure $p = 2000 \text{ Pa}$ and the temperature $\vartheta = 20^\circ \text{C}$. Calculate their total mass, molar mass, density and number of molecules of the vapor. Relative atomic mass let's assume as 1 for hydrogen and 16 for oxygen; the molar gas constant is $R = 8.3 \text{ J K}^{-1} \text{ mol}^{-1}$, Avogadro constant is $N_A = 6.6 \cdot 10^{23} \text{ mol}^{-1}$, let's assume the behavior of the vapor as an ideal gas.

$$M = \frac{m}{n} = \frac{M m_{\text{molecule}}}{\frac{N}{N_A}} = N_A u (2A_H + A_O) = \frac{1 \text{ g}}{1 \text{ mol}} (2 \cdot 1 + 16) = 18 \text{ g mol}^{-1};$$

$$pV = nRT = \frac{m}{M} RT; T = (\vartheta + 273.15) \text{ K} \approx 293 \text{ K};$$

$$m = \frac{pVM}{RT} \approx \frac{2000 \cdot 3 \cdot 10^{-6} \cdot 18 \text{ g mol}^{-1}}{8.3 \cdot 293} \approx \frac{9}{2} \cdot 10^{-5} \text{ g} \approx 45 \text{ ng};$$

$$\rho = \frac{m}{V} = \frac{pM}{RT} \approx \frac{2000 \cdot 18 \text{ g mol}^{-1}}{8.3 \cdot 293} \approx \frac{30}{2} \text{ g m}^{-3} = 15 \text{ g m}^{-3};$$

$$N = nN_A = \frac{pVM}{RT} N_A \approx \frac{2000 \cdot 3 \cdot 10^{-6}}{8.3 \cdot 293} \cdot 6.6 \cdot 10^{23} \approx 1.5 \cdot 10^{18}.$$