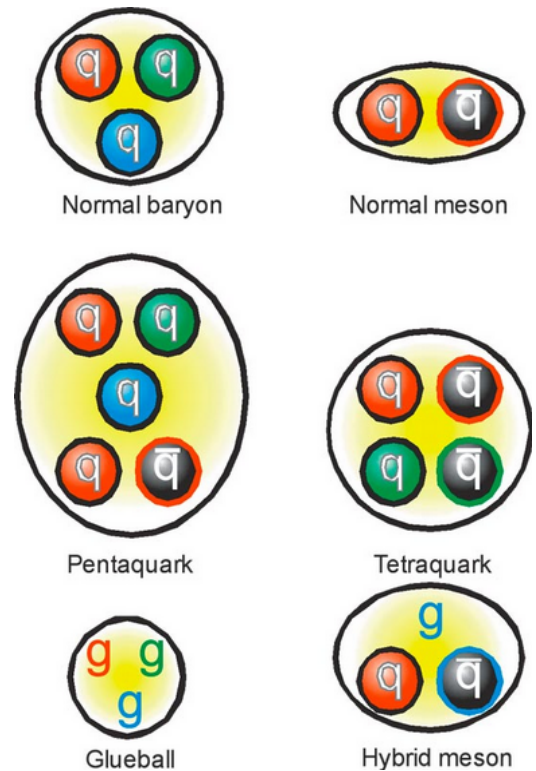
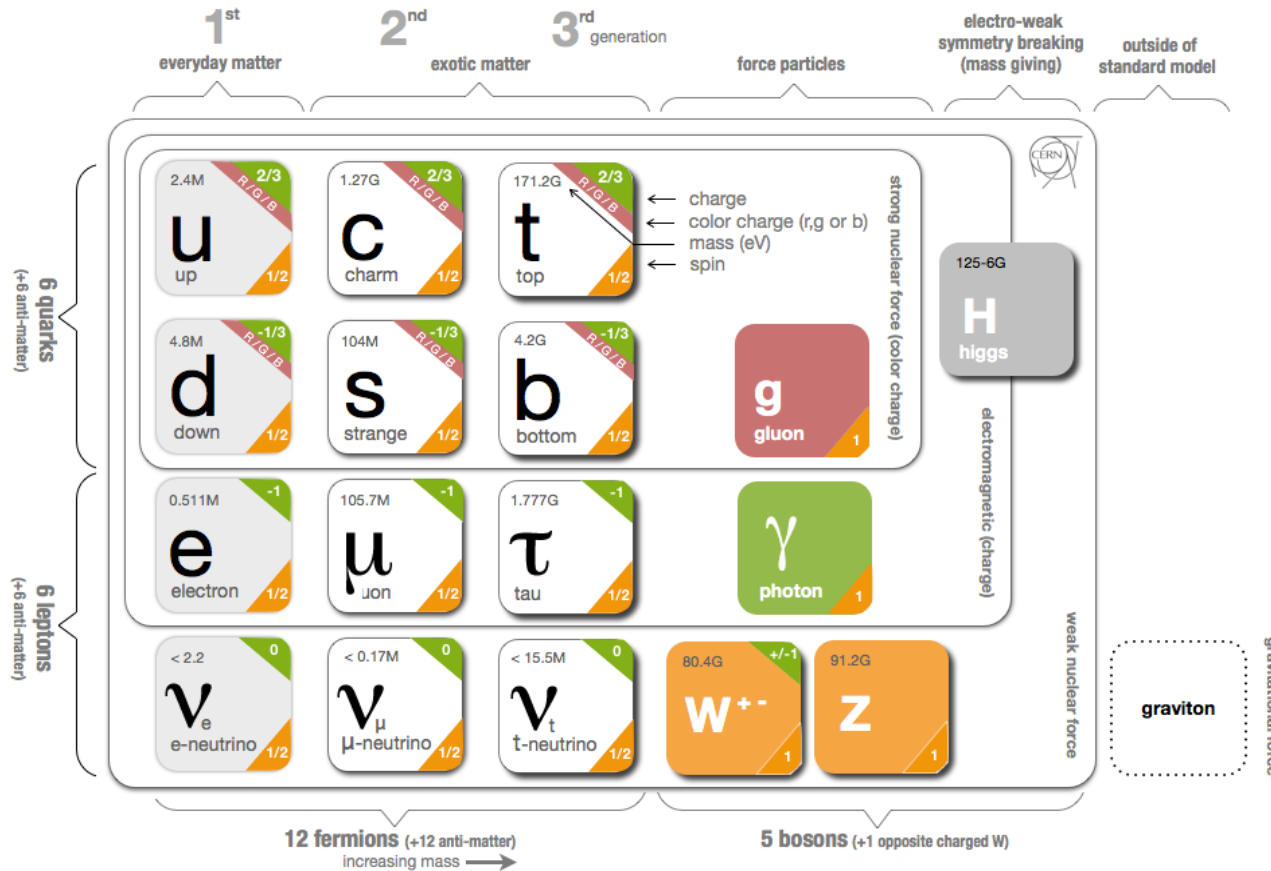


The background of the slide is a reproduction of the painting 'The Starry Night' by Vincent van Gogh. It depicts a night scene with a turbulent, star-filled sky reflected in a dark body of water. A small boat with two figures is in the foreground, and a village with a prominent church spire is visible on the horizon. The text 'The origin of elements' is overlaid in the center.

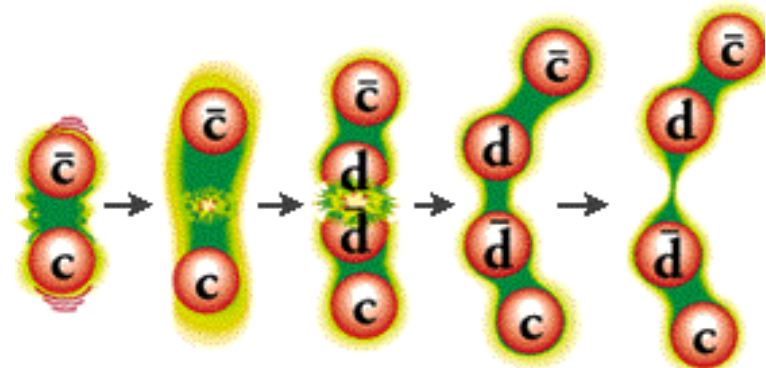
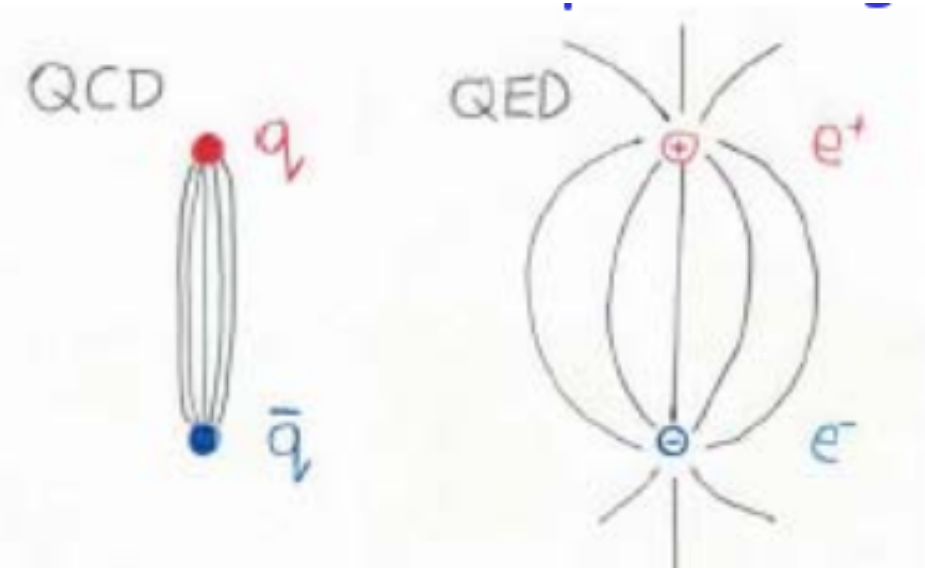
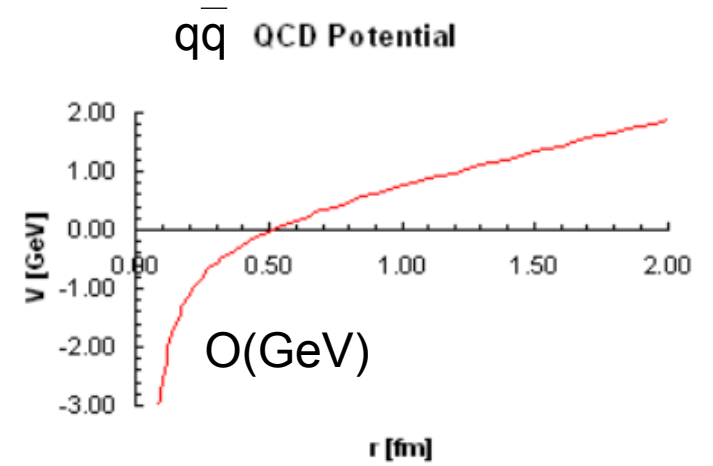
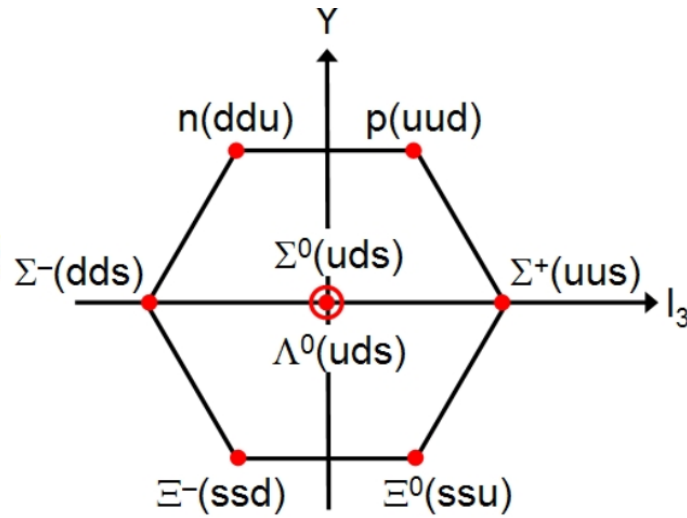
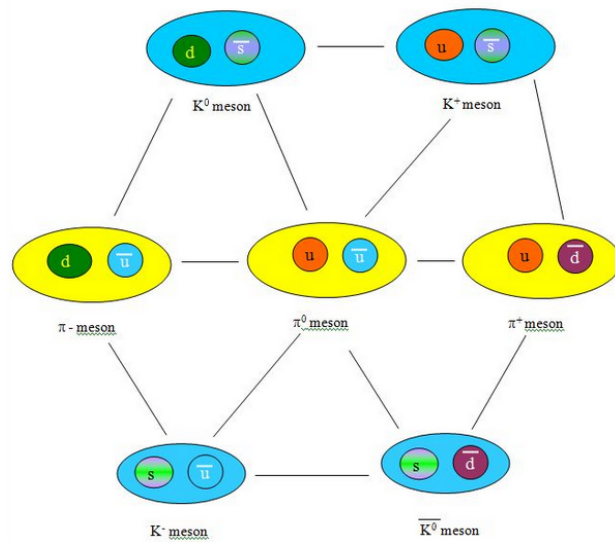
The origin of elements

The Standard Model of Particles and Interactions

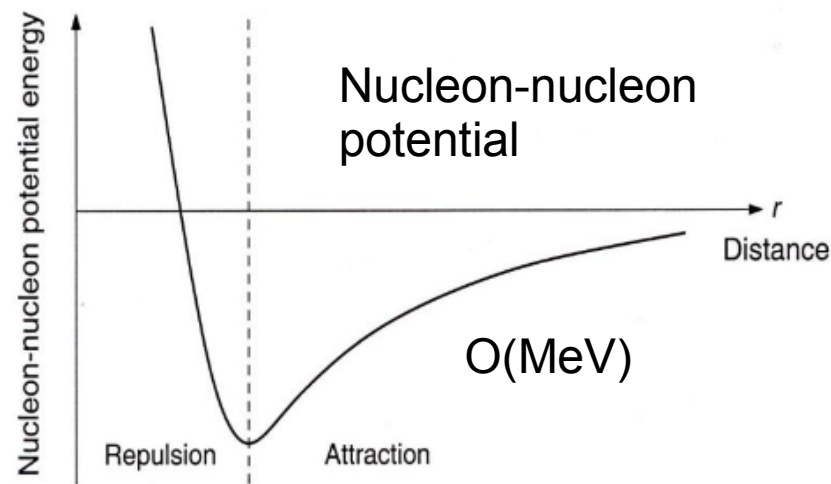
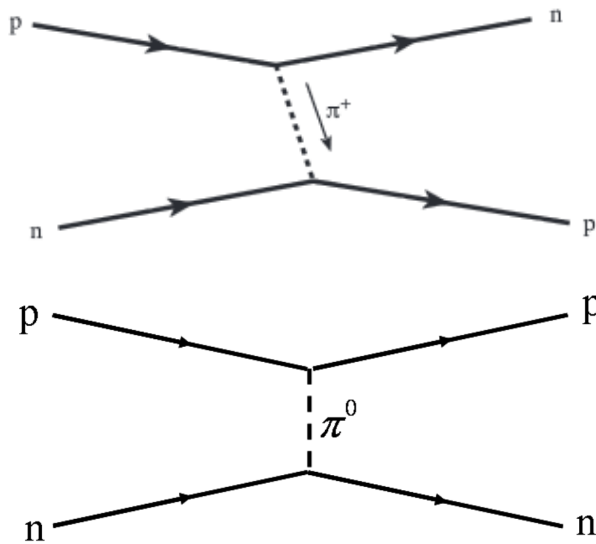


Strong 	Force which holds nucleus together	Strength 1	Range (m) 10^{-15} (diameter of a medium sized nucleus)	Particle gluons, π (nucleons)
Electro-magnetic 		Strength $\frac{1}{137}$	Range (m) Infinite	Particle photon mass = 0 spin = 1
Weak 		Strength 10^{-6}	Range (m) 10^{-18} (0.1% of the diameter of a proton)	Particle Intermediate vector bosons W^+ , W^- , Z_0 , mass > 80 GeV spin = 1
Gravity 		Strength 6×10^{-39}	Range (m) Infinite	Particle graviton ? mass = 0 spin = 2

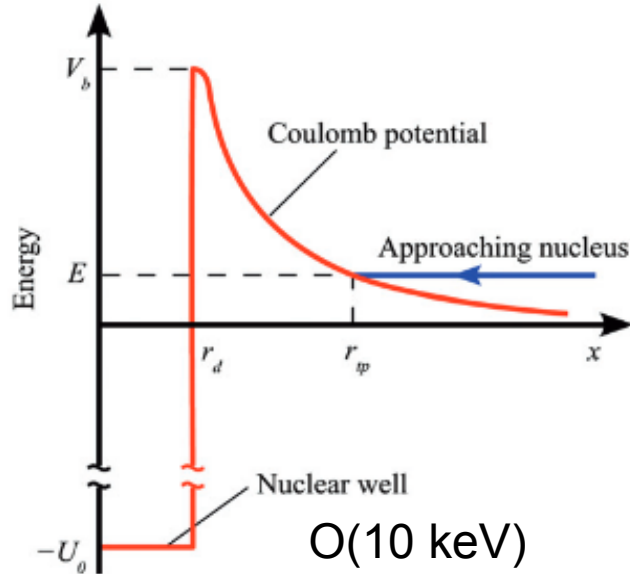
Mesons, interquark potentials



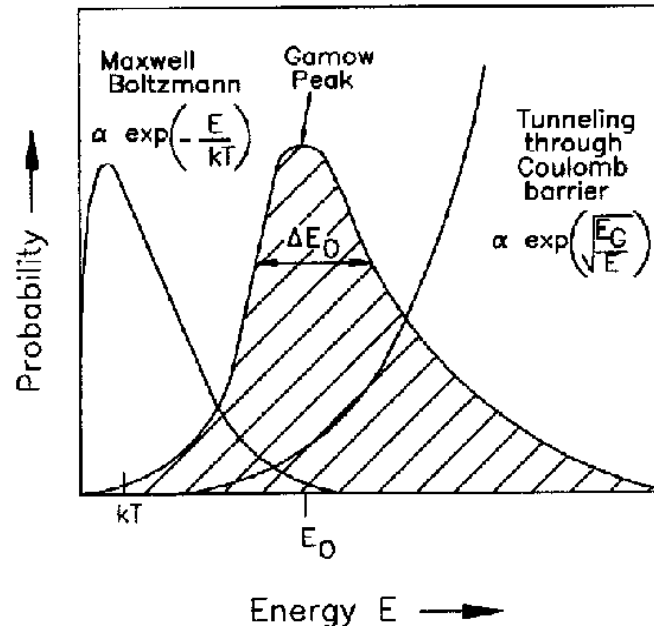
Nuclear forces, fusion



- Strong force > elmag at small distances
 - Charge independence
 - Bound nucleons are stable

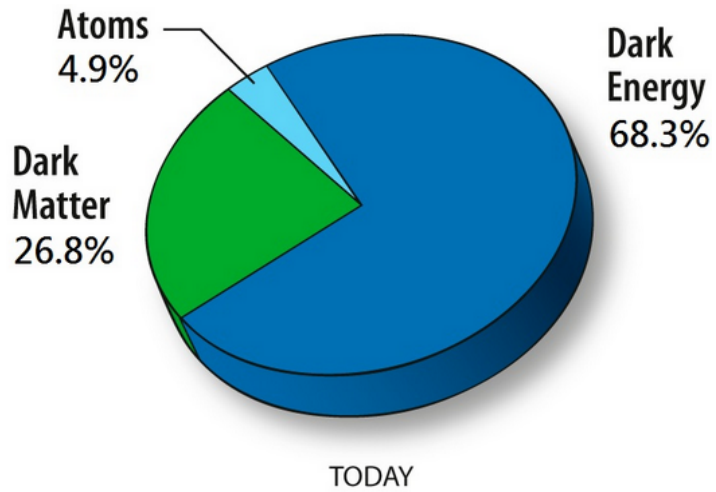


Coulomb force gains relative strength



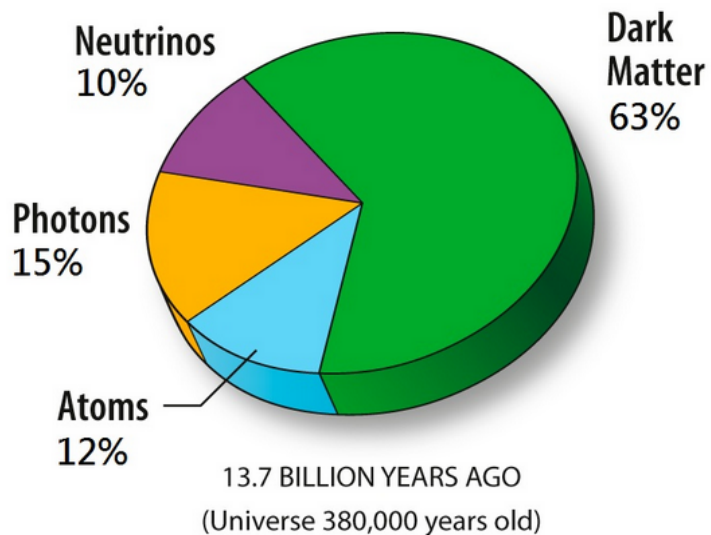
- Gamow peak
- Most reactions occur here

The Standard Model of Particles and Interactions



Troubles:

1. Gravitation
2. Dark matter / Dark Energy
3. Neutrino masses
4. Matter/antimatter symmetry



Periodic Table of the Elements

1
1IA
11A

2
IIA
2A

13
IIIA
3A

14
IVA
4A

15
VA
5A

16
VIA
6A

17
VIIA
7A

18
VIIIA
8A

1
H
Hydrogen
1.0079

3
Li
Lithium
6.941

11
Na
Sodium
22.989768

19
K
Potassium
39.0983

37
Rb
Rubidium
85.4678

55
Cs
Cesium
132.90543

87
Fr
Francium
223.0197

2
He
Helium
4.00260

4
Be
Beryllium
9.01218

12
Mg
Magnesium
24.305

20
Ca
Calcium
40.078

38
Sr
Strontium
87.62

56
Ba
Barium
137.327

88
Ra
Radium
226.0254

5
B
Boron
10.811

13
Al
Aluminum
26.981539

31
Ga
Gallium
69.732

49
In
Indium
114.818

67
Ho
Holmium
164.93032

85
At
Astatine
209.9871

6
C
Carbon
12.011

14
Si
Silicon
28.0855

32
Ge
Germanium
72.64

50
Sn
Tin
118.71

68
Er
Erbium
167.26

86
Rn
Radon
222.0176

7
N
Nitrogen
14.00674

15
P
Phosphorus
30.973762

33
As
Arsenic
74.92159

51
Sb
Antimony
121.760

69
Tm
Thulium
168.93421

87
Lu
Lutetium
174.967

8
O
Oxygen
15.9994

16
S
Sulfur
32.066

34
Se
Selenium
78.96

52
Te
Tellurium
127.6

70
Yb
Ytterbium
173.04

88
No
Nobelium
259.1009

9
F
Fluorine
18.998403

17
Cl
Chlorine
35.4527

35
Br
Bromine
79.904

53
I
Iodine
126.90447

81
Tl
Thallium
204.3833

99
Es
Einsteinium
[254]

10
Ne
Neon
20.1797

18
Ar
Argon
39.948

36
Kr
Krypton
83.80

54
Xe
Xenon
131.29

86
Rn
Radon
222.0176

11
V
Vanadium
50.9415

23
Nb
Niobium
92.90638

41
Ta
Tantalum
180.9479

73
Ta
Tantalum
180.9479

105
Db
Dubnium
[262]

12
Zn
Zinc
65.39

30
Zn
Zinc
65.39

48
Cd
Cadmium
112.411

66
Dy
Dysprosium
162.50

84
Po
Polonium
[208.9824]

112
Cn
Copernicium
[277]

21
Sc
Scandium
44.95591

39
Y
Yttrium
88.90585

57-71
Lanthanide Series

72
Hf
Hafnium
178.49

104
Rf
Rutherfordium
[261]

22
Ti
Titanium
47.88

40
Zr
Zirconium
91.224

72
Hf
Hafnium
178.49

104
Rf
Rutherfordium
[261]

24
Cr
Chromium
51.9961

42
Mo
Molybdenum
95.94

74
W
Tungsten
183.85

106
Sg
Seaborgium
[266]

25
Mn
Manganese
54.938

43
Tc
Technetium
98.9072

75
Re
Rhenium
186.207

107
Bh
Bohrium
[264]

26
Fe
Iron
55.847

44
Ru
Ruthenium
101.07

76
Os
Osmium
190.23

108
Hs
Hassium
[269]

27
Co
Cobalt
58.9332

45
Rh
Rhodium
102.9055

77
Ir
Iridium
192.22

109
Mt
Meitnerium
[268]

28
Ni
Nickel
58.6934

46
Pd
Palladium
106.42

78
Pt
Platinum
195.08

110
Ds
Darmstadtium
[269]

29
Cu
Copper
63.546

47
Ag
Silver
107.8682

79
Au
Gold
196.9665

111
Rg
Roentgenium
[272]

31
Ga
Gallium
69.732

49
In
Indium
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Mercury
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Ba
Barium
137.327

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Po
Polonium
[208.9824]

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[277]

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Yttrium
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Lanthanide Series

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At
Astatine
209.9871

113
Uut
Ununtrium
unknown

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Hf
Hafnium
178.49

86
Rn
Radon
222.0176

114
Uuq
Ununquadium
[289]

41
Nb
Niobium
92.90638

73
Ta
Tantalum
180.9479

87
Lu
Lutetium
174.967

115
Uup
Ununpentium
unknown

42
Mo
Molybdenum
95.94

74
W
Tungsten
183.85

88
No
Nobelium
259.1009

116
Uuh
Ununhexium
[298]

43
Tc
Technetium
98.9072

75
Re
Rhenium
186.207

89
Ac
Actinium
227.0278

117
Uus
Ununseptium
unknown

44
Ru
Ruthenium
101.07

76
Os
Osmium
190.23

90
Th
Thorium
232.0381

118
Uuo
Ununoctium
unknown

45
Rh
Rhodium
102.9055

77
Ir
Iridium
192.22

91
Pa
Protactinium
231.03588

46
Pd
Palladium
106.42

78
Pt
Platinum
195.08

92
U
Uranium
238.0289

47
Ag
Silver
107.8682

79
Au
Gold
196.9665

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Np
Neptunium
237.0482

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Cd
Cadmium
112.411

80
Hg
Mercury
200.59

94
Pu
Plutonium
244.0642

49
In
Indium
114.818

81
Tl
Thallium
204.3833

95
Am
Americium
243.0614

50
Sn
Tin
118.71

82
Pb
Lead
207.2

96
Cm
Curium
247.0703

51
Sb
Antimony
121.760

83
Bi
Bismuth
208.98037

97
Bk
Berkelium
247.0703

52
Te
Tellurium
127.6

84
Po
Polonium
[208.9824]

98
Cf
Californium
251.0796

53
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Iodine
126.90447

85
At
Astatine
209.9871

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Es
Einsteinium
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Xe
Xenon
131.29

86
Rn
Radon
222.0176

100
Fm
Fermium
257.0951

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Cs
Cesium
132.90543

87
Lu
Lutetium
174.967

101
Md
Mendelevium
258.1

56
Ba
Barium
137.327

88
No
Nobelium
259.1009

102
No
Nobelium
259.1009

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Lanthanide Series

89
Ac
Actinium
227.0278

103
Lr
Lawrencium
[262]

58
Ce
Cerium
140.115

90
Th
Thorium
232.0381

104
Rf
Rutherfordium
[261]

59
Pr
Praseodymium
140.90765

91
Pa
Protactinium
231.03588

105
Db
Dubnium
[262]

60
Nd
Neodymium
144.24

92
U
Uranium
238.0289

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Seaborgium
[266]

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Pm
Promethium
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Np
Neptunium
237.0482

107
Bh
Bohrium
[264]

62
Sm
Samarium
150.36

94
Pu
Plutonium
244.0642

108
Hs
Hassium
[269]

63
Eu
Europium
151.9655

95
Am
Americium
243.0614

109
Mt
Meitnerium
[268]

64
Gd
Gadolinium
157.25

96
Cm
Curium
247.0703

110
Ds
Darmstadtium
[269]

65
Tb
Terbium
158.92534

97
Bk
Berkelium
247.0703

111
Rg
Roentgenium
[272]

66
Dy
Dysprosium
162.50

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Copernicium
[277]

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Holmium
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Ununtrium
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Fermium
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Ununquadium
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173.04

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No
Nobelium
259.1009

116
Uuh
Ununhexium
[298]

71
Lu
Lutetium
174.967

103
Lr
Lawrencium
[262]

117
Uus
Ununseptium
unknown

118
Uuo
Ununoctium
unknown

Alkali Metal

Alkaline Earth

Transition Metal

Basic Metal

Semimetals

Nonmetals

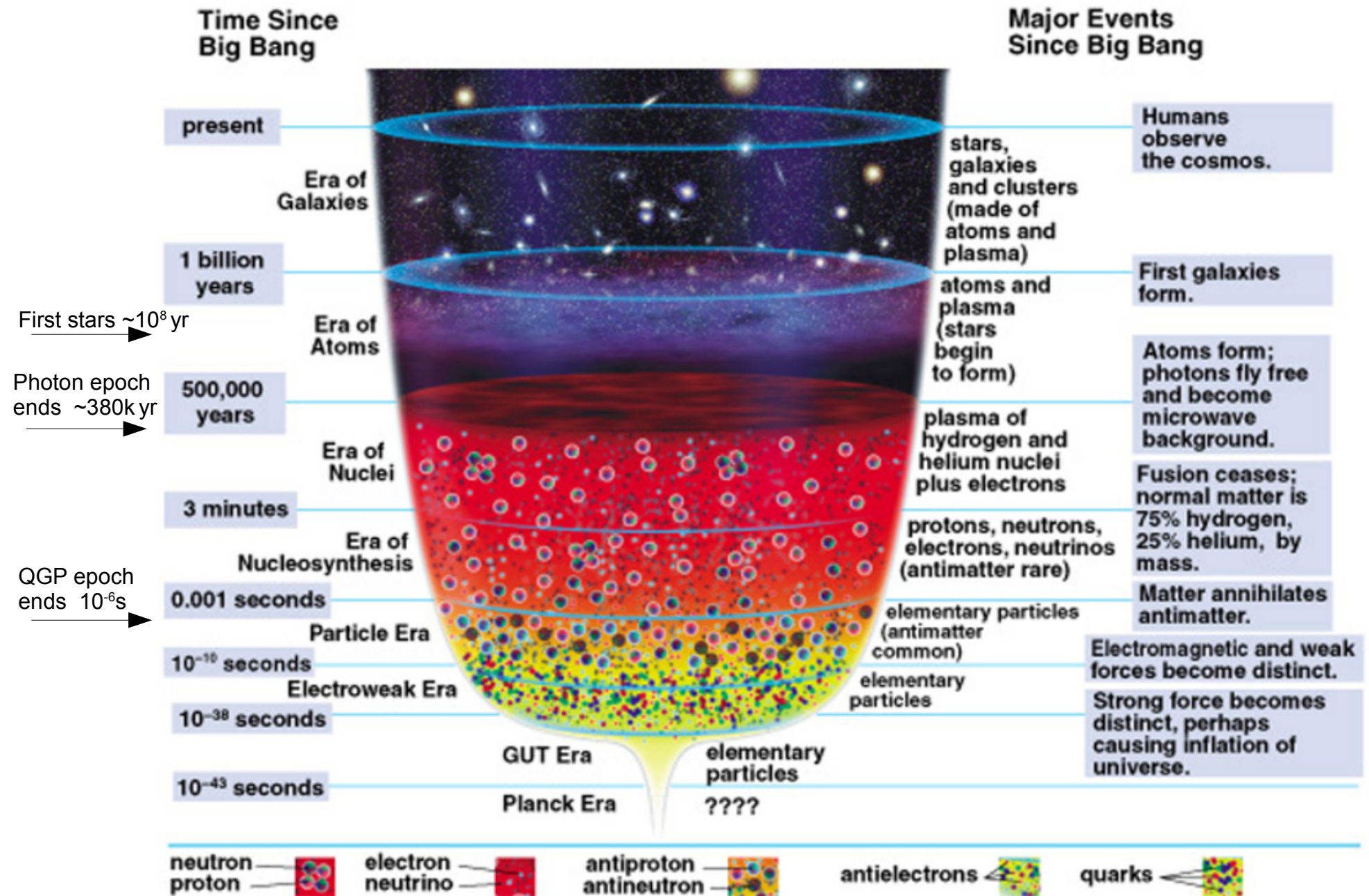
Halogens

Noble Gas

Lanthanides

Actinides

The Big Bang



BBN according to the SM

- $t_U < 10^{-6} \text{ s}$
 - Particle antiparticle pairs annihilate and are produced by photons, $n_p = n_{\bar{p}} \sim n_\gamma$
- $t_U > 10^{-6} \text{ s}$
 - Particle pairs annihilate into photons, but the universe is too cold to produce them ($T < 2m_p$)
- Today
 - All particles and antiparticles have annihilated, there are no galaxies, no stars, no humans

Sakharov's conditions



- Sakharov's conditions
 - Baryon number violation
 - Matter and antimatter are (almost) symmetric, but our
 - part of the universe involves matter only
 - $\frac{n_B - \bar{n}_B}{n_\gamma} = 6.1 \mp 0.3 \times 10^{-10}$
 - Sphalerons convert leptons to baryons, conserving B-L
 - C and CP violation – antimatter behaves differently than matter
 - C is violated by EW
 - CPV is too small in SM (if CKM is the only source, must occur during EW phase)
 - Out of equilibrium B-violating processes
 - Phase transitions
 - Decay of non-SM heavy particles
- 4th Generation of quarks?

Letters to the Editor

PUBLICATION of brief reports of important discoveries in physics may be secured by addressing them to this department. The closing date for this department is five weeks prior to the date of issue. No proof will be sent to the authors. The Board of Editors does not hold itself responsible for the opinions expressed by the correspondents. Communications should not exceed 600 words in length.

The Origin of Chemical Elements

R. A. ALPHER*

*Applied Physics Laboratory, The Johns Hopkins University,
Silver Spring, Maryland*

AND

H. BETHE

Cornell University, Ithaca, New York

AND

G. GAMOW

*The George Washington University, Washington, D. C.
February 18, 1948*

AS pointed out by one of us,¹ various nuclear species must have originated not as the result of an equilibrium corresponding to a certain temperature and density, but rather as a consequence of a continuous building-up process arrested by a rapid expansion and cooling of the primordial matter. According to this picture, we must imagine the early stage of matter as a highly compressed neutron gas (overheated neutral nuclear fluid) which started decaying into protons and electrons when the gas pressure fell down as the result of universal expansion. The radiative capture of the still remaining neutrons by the newly formed protons must have led first to the formation of deuterium nuclei, and the subsequent neutron captures resulted in the building up of heavier and heavier nuclei. It must be remembered that, due to the comparatively short time allowed for this process,² the building up of heavier nuclei must have proceeded just above the upper fringe of the stable elements (short-lived Fermi elements), and the present frequency distribution of various atomic species was attained only somewhat later as the result of adjustment of their electric charges by β -decay.

Thus the observed slope of the abundance curve must not be related to the temperature of the original neutron gas, but rather to the time period permitted by the expansion process. Also, the individual abundances of various nuclear species must depend not so much on their intrinsic stabilities (mass defects) as on the values of their neutron capture cross sections. The equations governing such a building-up process apparently can be written in the form:

$$\frac{dn_i}{dt} = f(t)(\sigma_{i-1}n_{i-1} - \sigma_i n_i) \quad i = 1, 2, \dots, 238, \quad (1)$$

where n_i and σ_i are the relative numbers and capture cross sections for the nuclei of atomic weight i , and where $f(t)$ is a factor characterizing the decrease of the density with time.

We may remark at first that the building-up process was apparently completed when the temperature of the neutron gas was still rather high, since otherwise the observed abundances would have been strongly affected by the resonances in the region of the slow neutrons. According to Hughes,³ the neutron capture cross sections of various elements (for neutron energies of about 1 Mev) increase exponentially with atomic number halfway up the periodic system, remaining approximately constant for heavier elements.

Using these cross sections, one finds by integrating Eqs. (1) as shown in Fig. 1 that the relative abundances of various nuclear species decrease rapidly for the lighter elements and remain approximately constant for the elements heavier than silver. In order to fit the calculated curve with the observed abundances⁴ it is necessary to assume the integral of ρdt during the building-up period is equal to 5×10^4 g. sec./cm³.

On the other hand, according to the relativistic theory of the expanding universe⁵ the density dependence on time is given by $\rho \approx 10^9/P$. Since the integral of this expression diverges at $t=0$, it is necessary to assume that the building-up process began at a certain time t_0 , satisfying the relation:

$$\int_{t_0}^{\infty} (10^9/P) dt \approx 5 \times 10^4, \quad (2)$$

which gives us $t_0 \approx 20$ sec. and $\rho_0 \approx 2.5 \times 10^8$ g. sec./cm³. This result may have two meanings: (a) for the higher densities existing prior to that time the temperature of the neutron gas was so high that no aggregation was taking place, (b) the density of the universe never exceeded the value 2.5×10^8 g. sec./cm³ which can possibly be understood if we

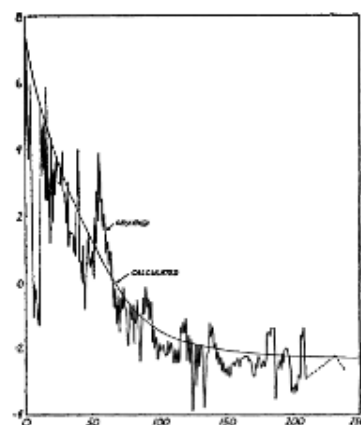


FIG. 1.
Log of relative abundance
Atomic weight

The Origin of Chemical Elements

R. A. ALPHER*

*Applied Physics Laboratory, The Johns Hopkins University,
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AND

H. BETHE

Cornell University, Ithaca, New York

AND

G. GAMOW

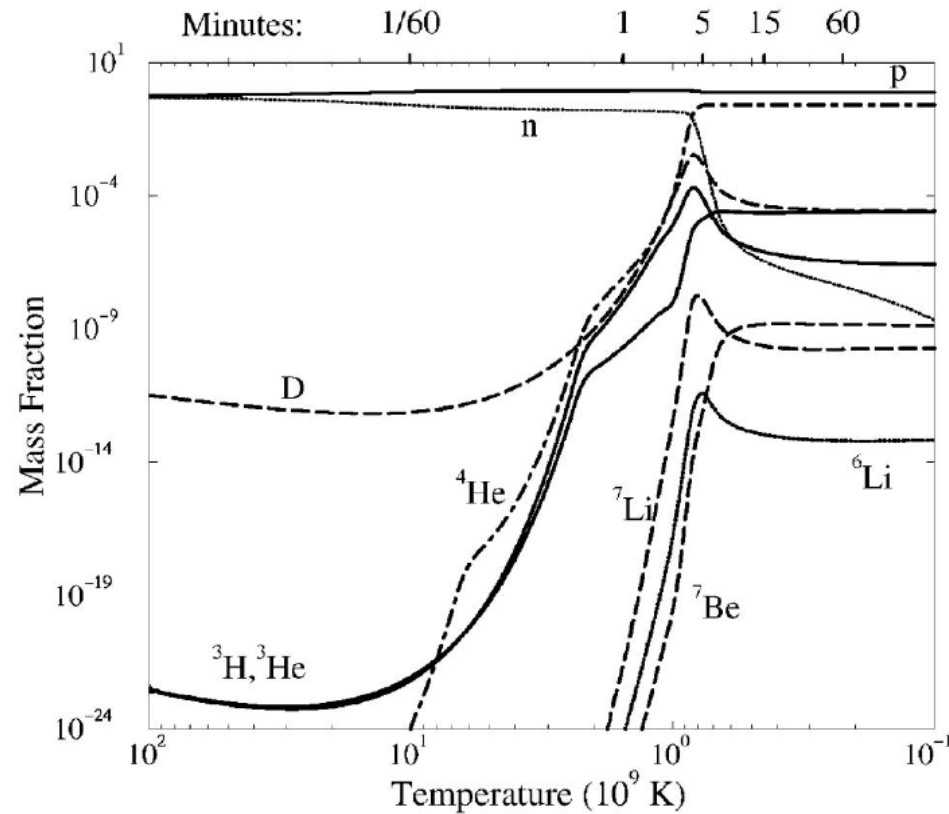
The George Washington University, Washington, D. C.

February 18, 1948

BBN

- $t_U < 10^{-11} \text{ s}$ - CPV interactions distinguish matter over antimatter, $(n_p - n_{\bar{p}})/(n_p + n_{\bar{p}}) \sim 10^{-7}$
- $t_U < 10^{-6} \text{ s}$ - Particle antiparticle pairs annihilate and are produced by photons, $n_p \approx n_{\bar{p}} \approx n_\gamma$
- $t_U > 10^{-6} \text{ s}$ Particle pairs annihilate into γ , but $T < 2 m_p$, $n_n \approx n_p$
 - $\nu_e + n \Leftrightarrow p + e^-$
 - $e^+ + n \Leftrightarrow p + \bar{\nu}_e$
 - $n \Leftrightarrow p + e^- + \bar{\nu}_e$
 - $e^+ + e^- \Leftrightarrow \gamma + \gamma$
- $t_U > 1 \text{ s}$ – Annihilation stops as all antiparticles have annihilated
 - Weak interaction cannot maintain equilibrium between all particles; neutrons decouple
 - $n_n / n_p \sim 1/6$
 - $n \rightarrow p + e^- + \bar{\nu}_e$
 - $p + n \rightarrow D + \gamma$
 -
- Today – Tiny surplus of particles that remained makes galaxies, stars, and humans

Big Bang nucleosynthesis



$$1\text{eV} = 11605\text{ K}$$

$$10^{10}\text{eV} \cong 1\text{ MeV}$$

- p^+ and n^0 soup
- ^2H in equilibrium (photodisintegration, $E_B \sim 2.2\text{ MeV}$),
- ^2H bottleneck - any deuterium that was created was destroyed
- E_B for $^3\text{He} \sim 7.72\text{ MeV}$, $^3\text{H} \sim 8.48\text{ MeV}$

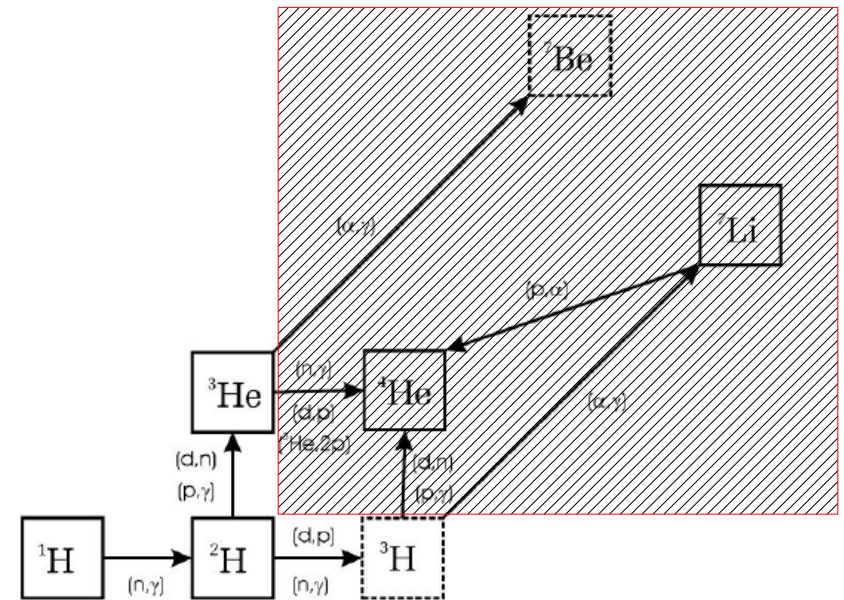
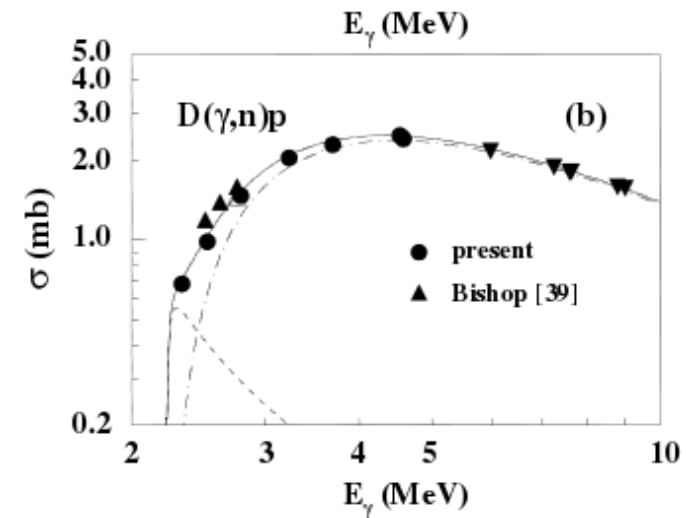


FIGURE 4. The reaction network of standard big bang nucleosynthesis. Unstable nuclear species are marked by dashed boxes. When all reactions are stopped, the unstable ^7Be decays to ^7Li and ^3H decays to ^3He .



Big Bang nucleosynthesis

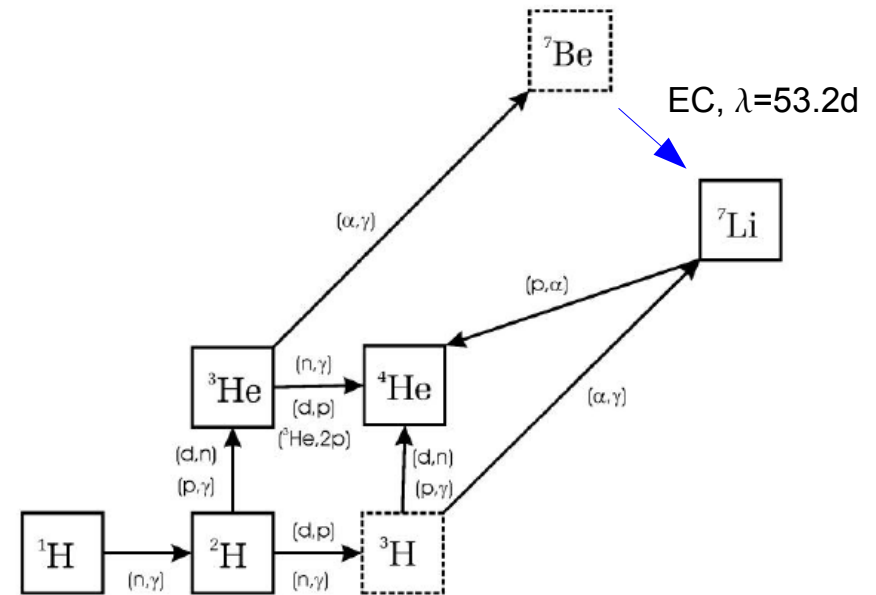
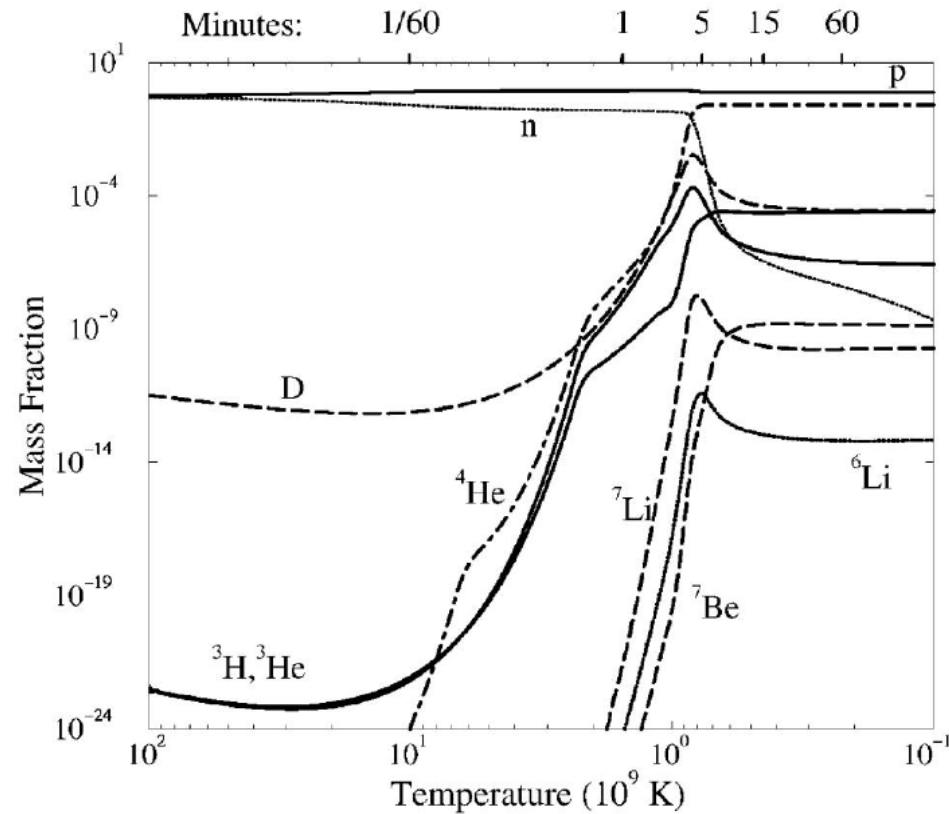
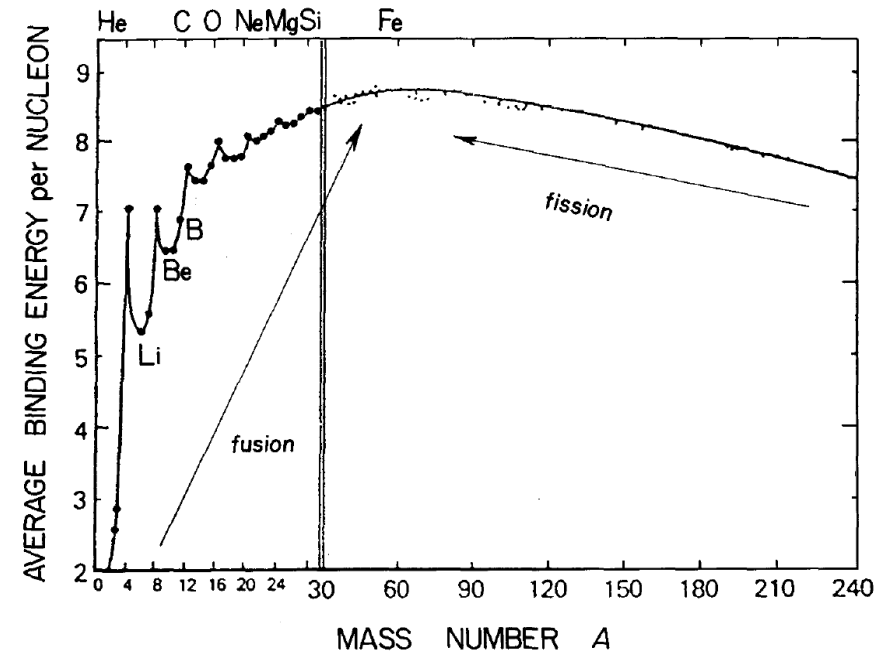


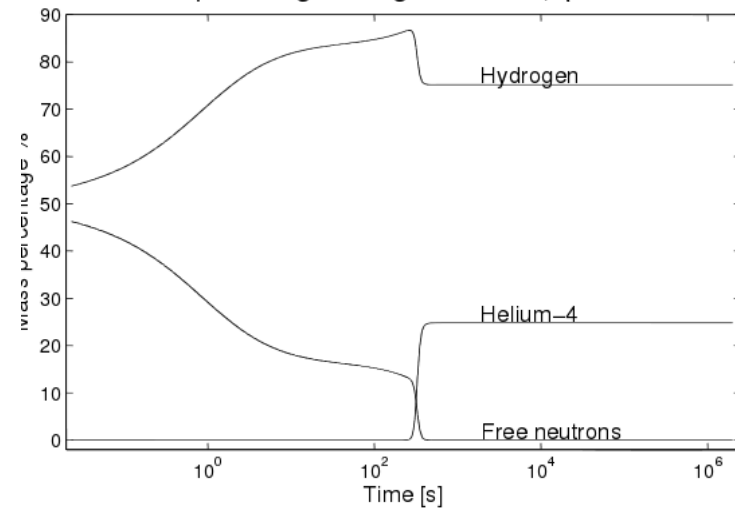
FIGURE 4. The reaction network of standard big bang nucleosynthesis. Unstable nuclear species are marked by dashed boxes. When all reactions are stopped, the unstable ^7Be decays to ^7Li and ^3H decays to ^3He .

- After temperature drop to about 1 MeV (1s), ^2H , ^3H and ^3He abundance increases
- BBN functions during 1s to 3 minutes after BB
- ^4He is stable – very high binding energy (~ 28 MeV)
- Ladder-wise creation of would follow ...
- There are no *stable* elements with $N=5,8$
 - Cannot proceed to higher N

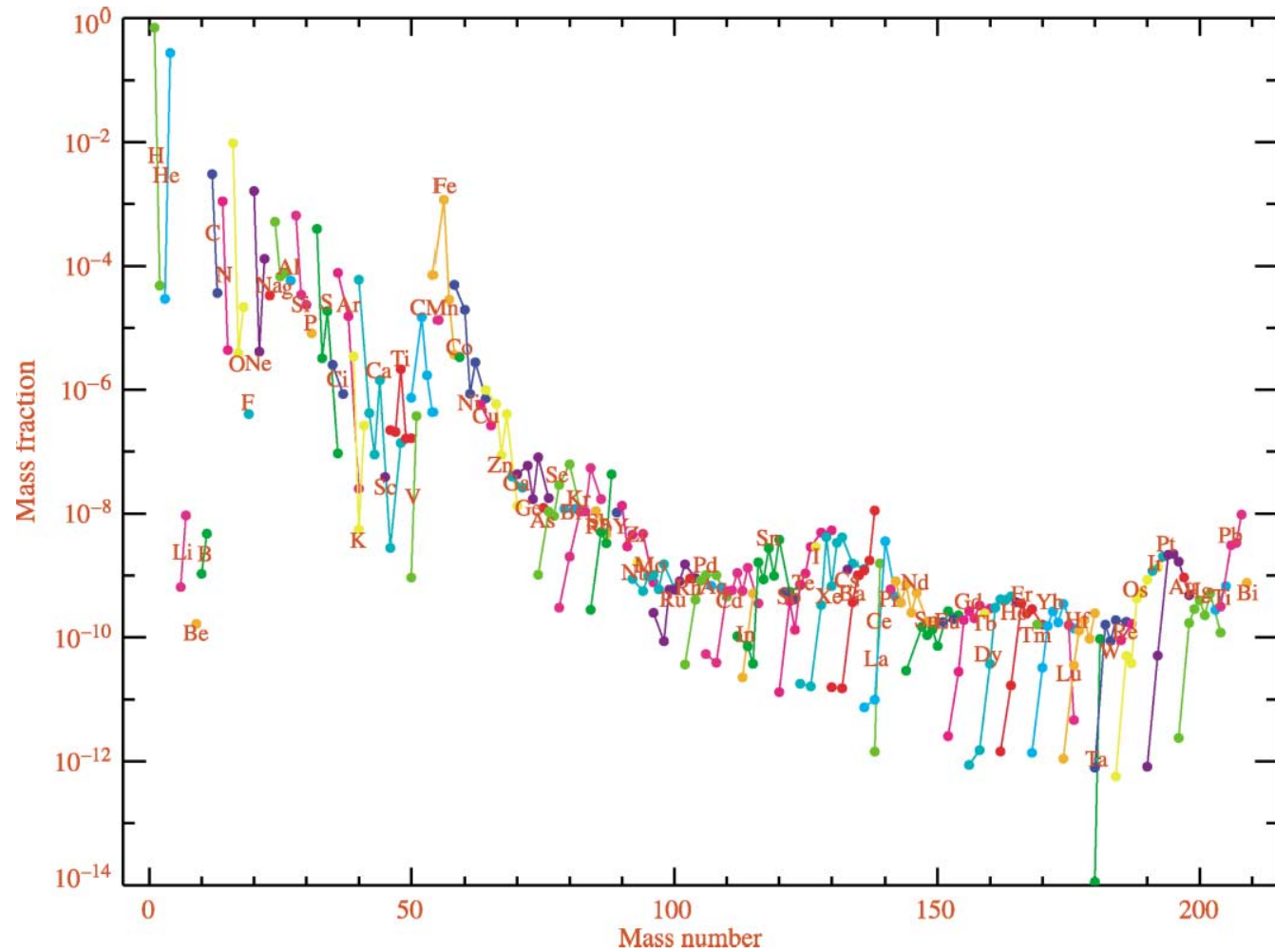


Elemental abundances in the solar system

Mass percentages for light elements, $\eta=6.1\text{E-}10$



Element	Mass percentage	Particles per hydrogen
H	75.2	1
n	0	0
d	3.90E-03	2.58E-05
t	0	0
³ He	2.40E-03	1.04E-05
⁴ He	24.8	0.0825
⁶ Li	5.06E-12	1.12E-14
⁷ Li	1.50E-08	2.85E-11
⁸ Li	7.40E-13	1.23E-15
⁷ Be	2.21E-07	4.20E-10



Stellar nucleosynthesis

REVIEWS OF MODERN PHYSICS

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Synthesis of the Elements in Stars*

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"It is the stars, The stars above us, govern our conditions";
(*King Lear*, Act IV, Scene 3)

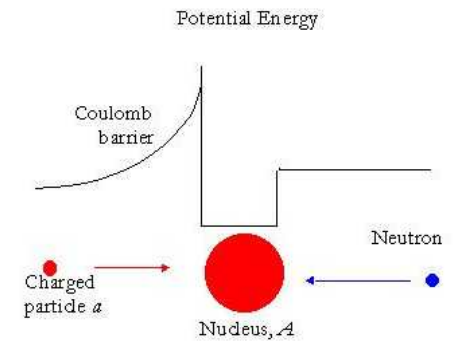
but perhaps

"The fault, dear Brutus, is not in our stars, But in ourselves,"
(*Julius Caesar*, Act I, Scene 2)

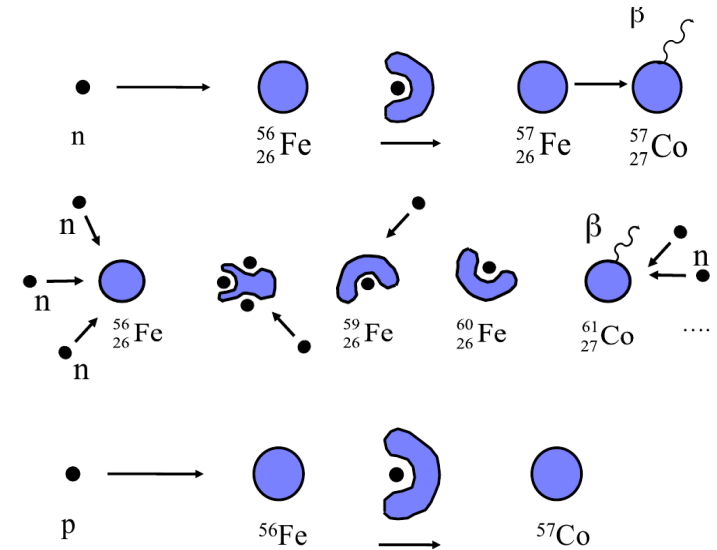
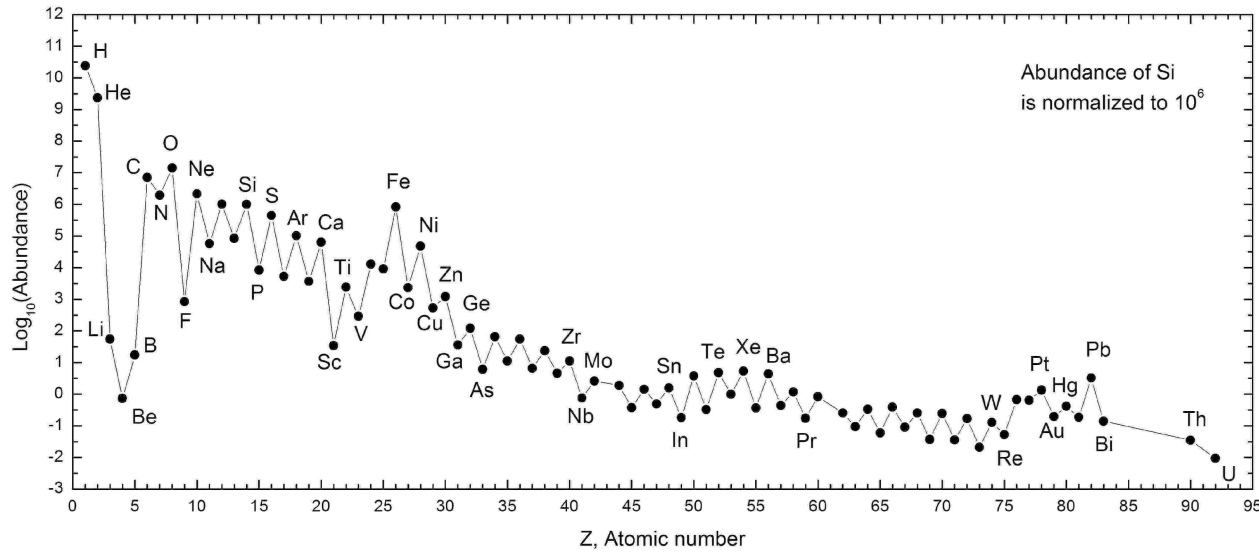
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The Potential Energy of a Positively Charged Particle as it Approaches a Nucleus.

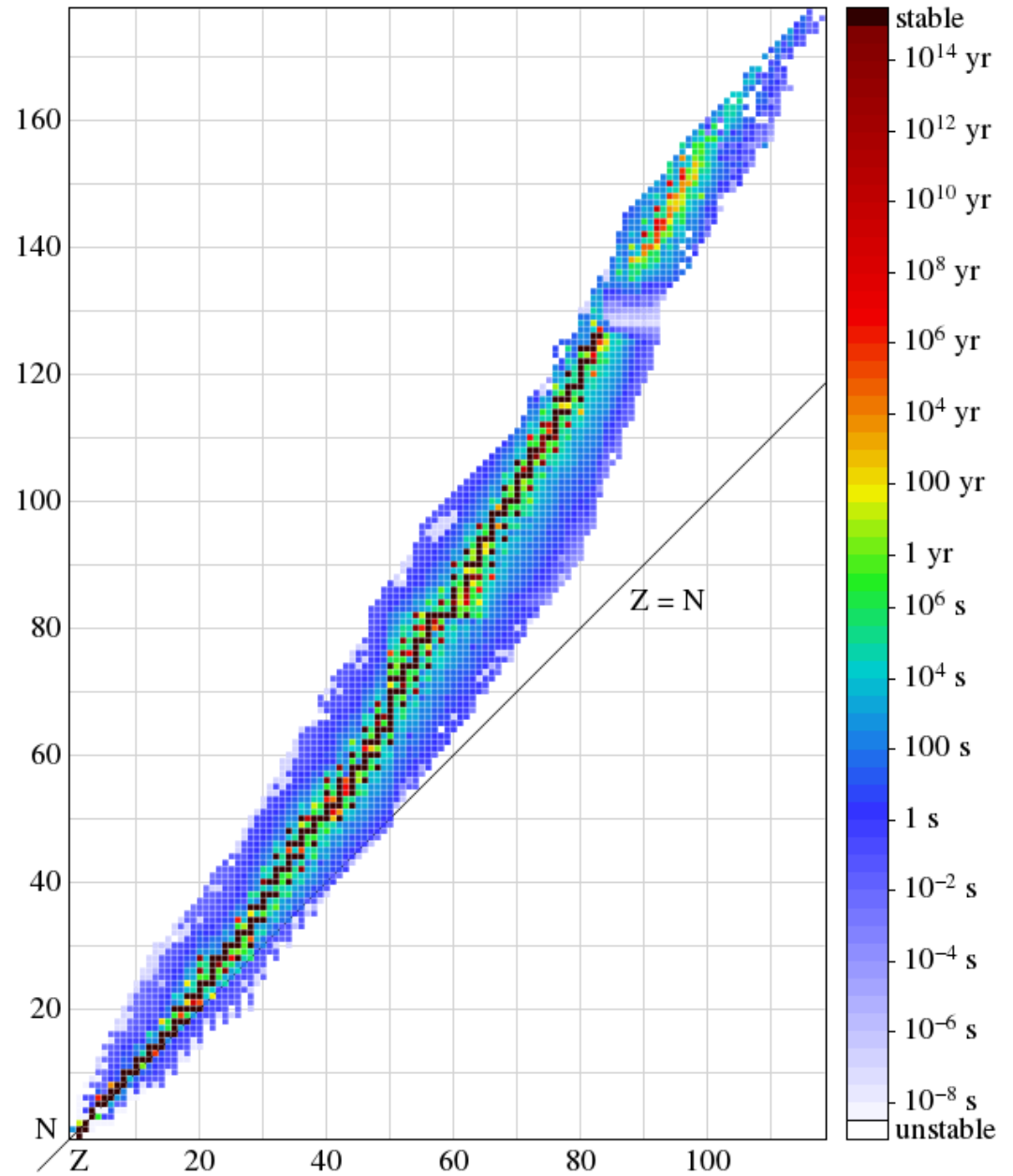
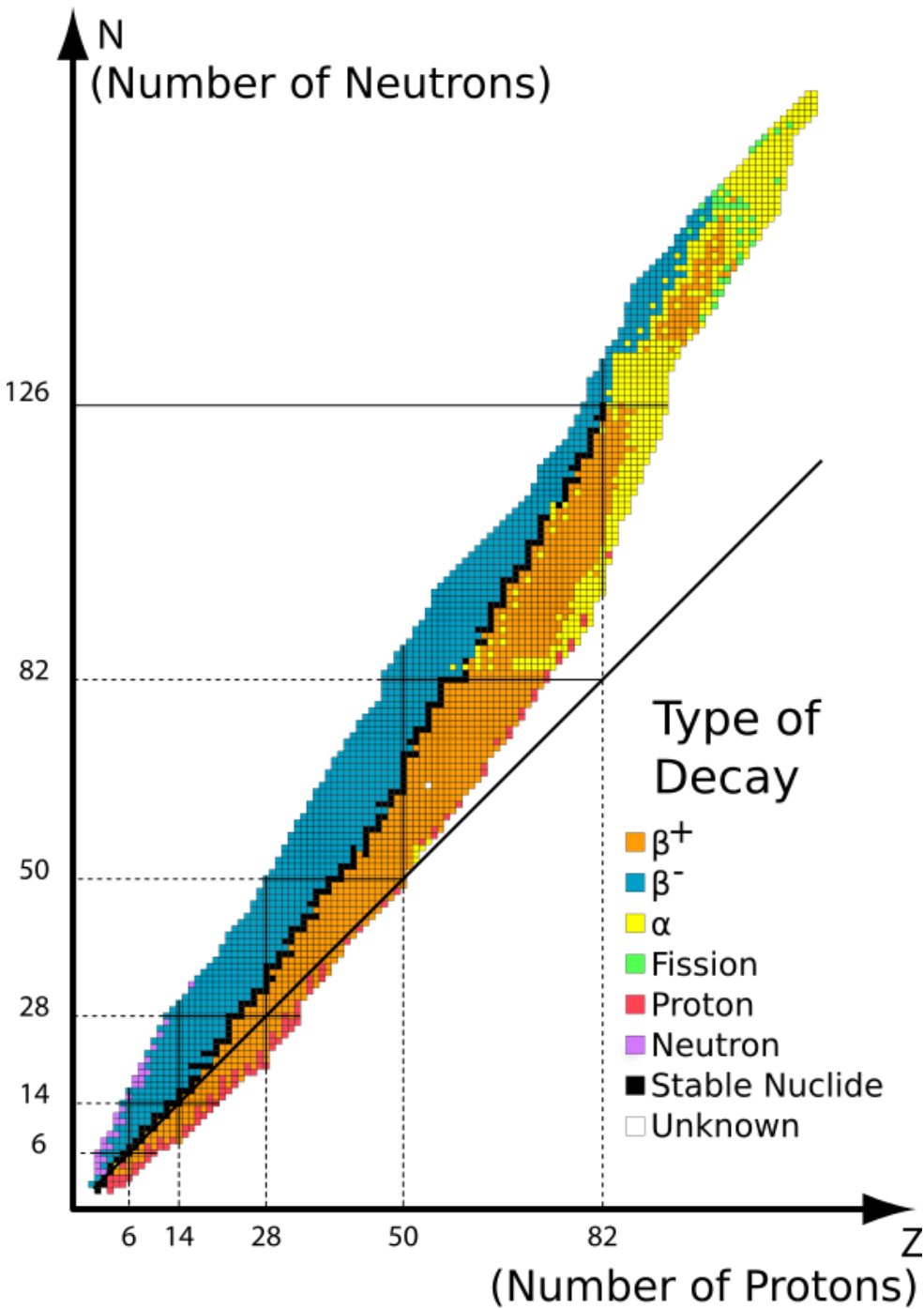


s, r, p processes

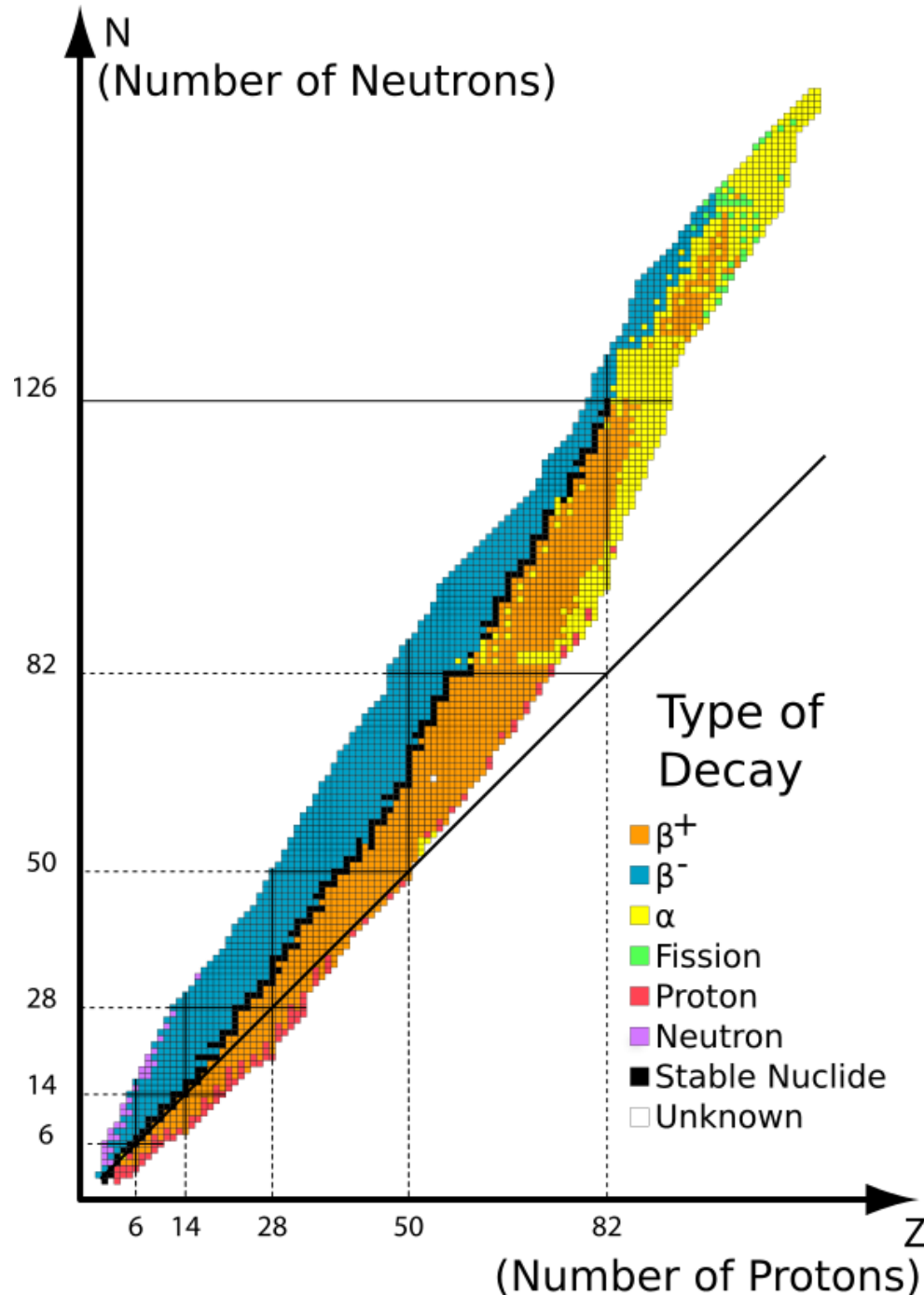


process	conditions	timescale	site
s-process (n-capture, ...)	$T \sim 0.1 \text{ GK}$ $\tau_n \sim 1\text{-}1000 \text{ yr}$, $n_n \sim 10^{7-8}/\text{cm}^3$	10^2 yr and 10^{5-6} yrs	Massive stars (weak) Low mass AGB stars (main)
r-process (n-capture, ...)	$T \sim 1\text{-}2 \text{ GK}$ $\tau_n \sim \mu\text{s}$, $n_n \sim 10^{24} / \text{cm}^3$	$\sim 1 \text{ s}$	Core collapse Supernovae Neutron Star Mergers?
p-process ((γ, n), ...)	$T \sim 2\text{-}3 \text{ GK}$	$\sim 1 \text{ s}$	Core collapse Supernovae

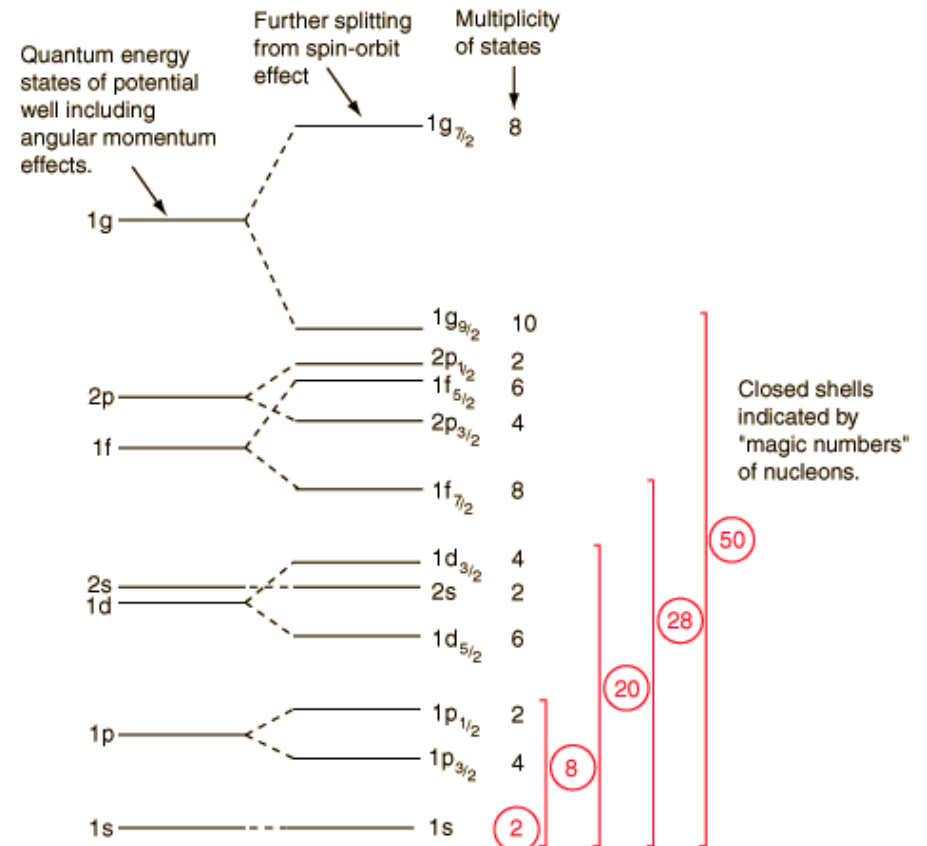
Graph of isotope stability



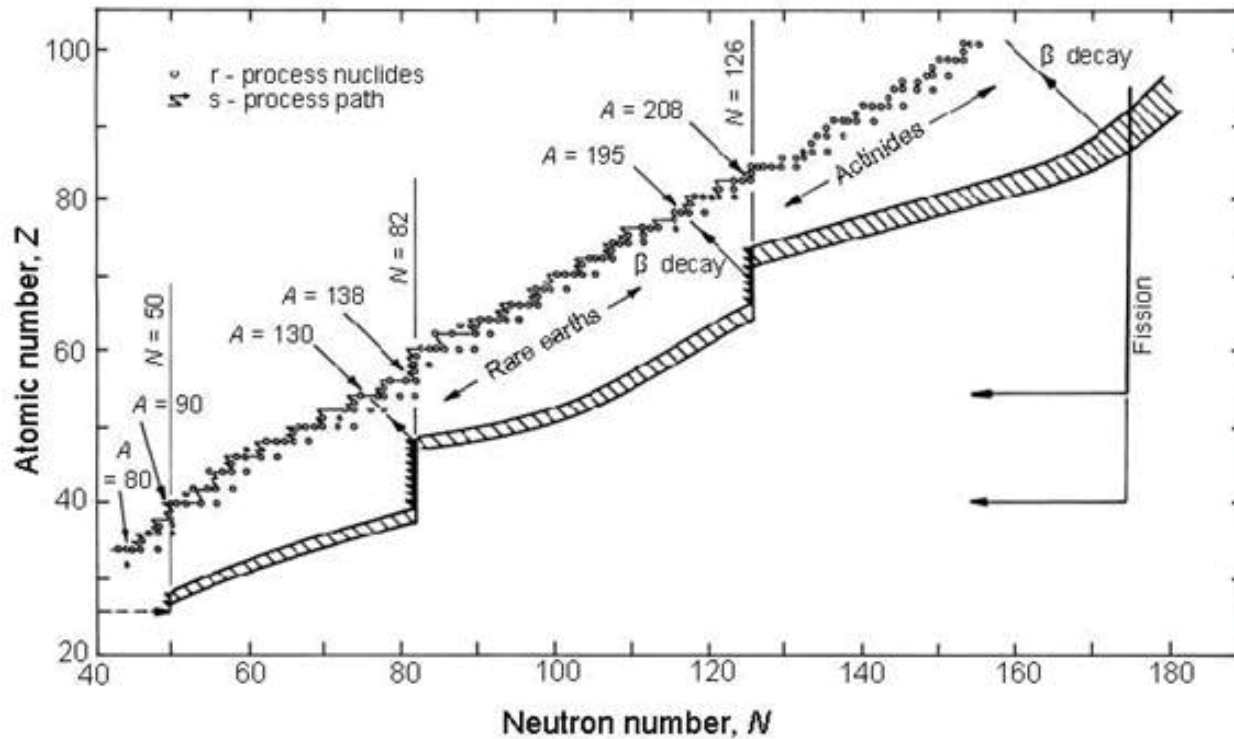
Graph of isotope stability



- $Z, N = 2, 8, 20, 28, 50, 82, 126 \dots$
- Higher binding energy per nucleon
- Doubly magic – n & p
 - ${}^4\text{He}, {}^{16}\text{O}, {}^{40}\text{Ca} \dots {}^{208}\text{Pb}$
- Closed shell model, similar to inert gases

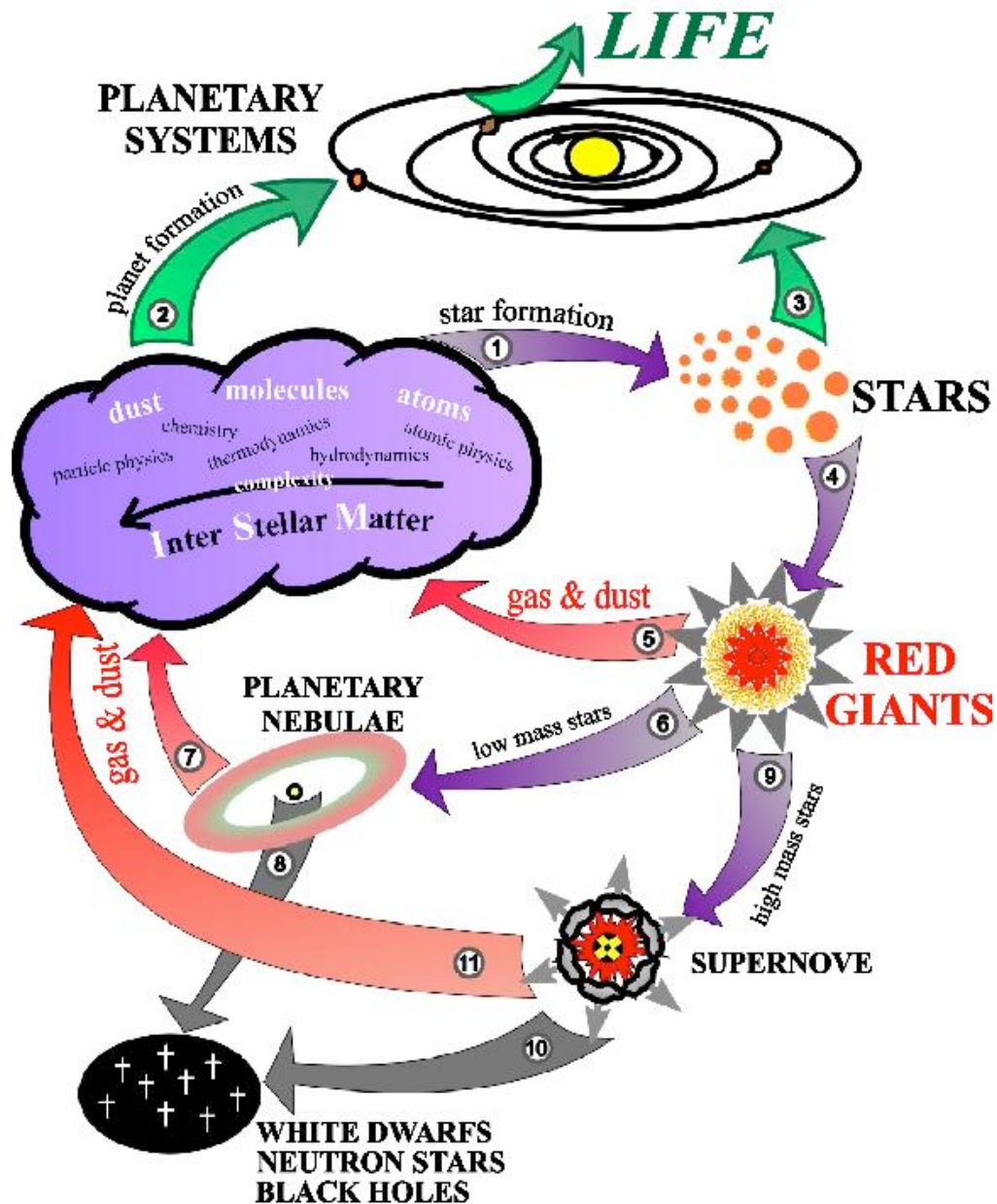


s, r processes



process	conditions	timescale	site
s-process (n-capture, ...)	$T \sim 0.1 \text{ GK}$ $\tau_n \sim 1\text{-}1000 \text{ yr}$, $n_n \sim 10^{7-8}/\text{cm}^3$	10^2 yr and 10^{5-6} yrs	Massive stars (weak) Low mass AGB stars (main)
r-process (n-capture, ...)	$T \sim 1\text{-}2 \text{ GK}$ $\tau_n \sim \mu\text{s}$, $n_n \sim 10^{24} / \text{cm}^3$	$\sim 1 \text{ s}$	Core collapse Supernovae

The great matter cycle



Initial mass (M_{sun})	Main sequence lifetime (Myr)	Total stellar lifetime (Myr)
25	6.7	7.5
15	11	13
5	78	102
2	8.7×10^2	1.2×10^3
1	9.2×10^3	1.2×10^4
0.8	2.0×10^4	3.2×10^5

Age of the galaxy $\approx 1.2 \times 10^{10}$ years;
Universe $\approx 1.37 \times 10^{10}$ years

Ultra low mass stars

– Initial masses from 0.08 to $0.8 M_{\text{Sun}}$

Low-mass stars:

– Initial masses from 0.8 to $\sim 2.25 M_{\text{Sun}}$

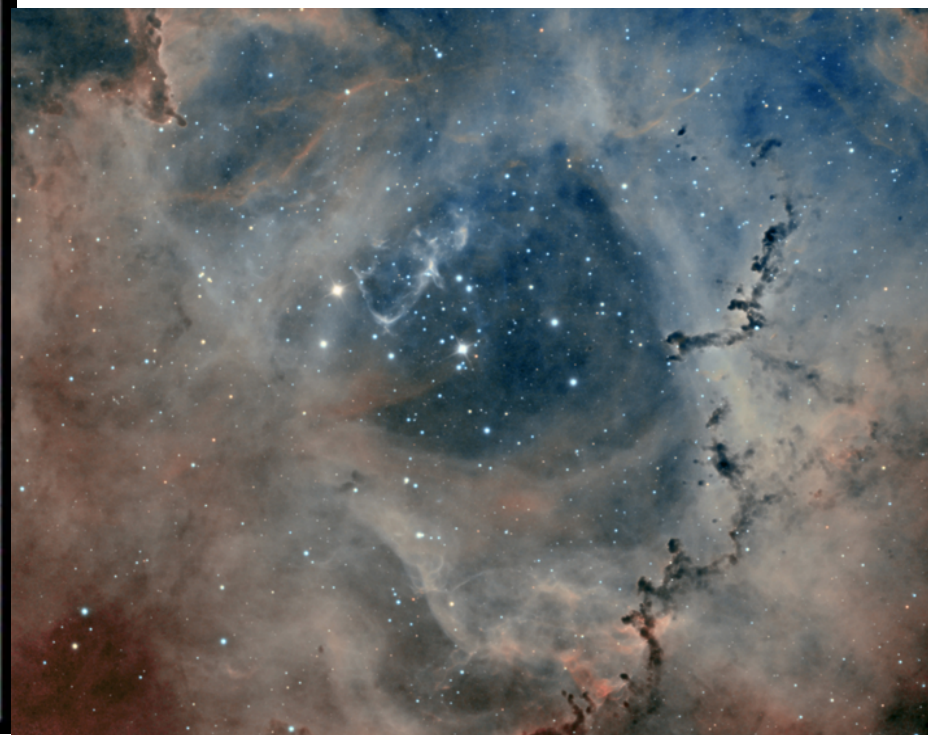
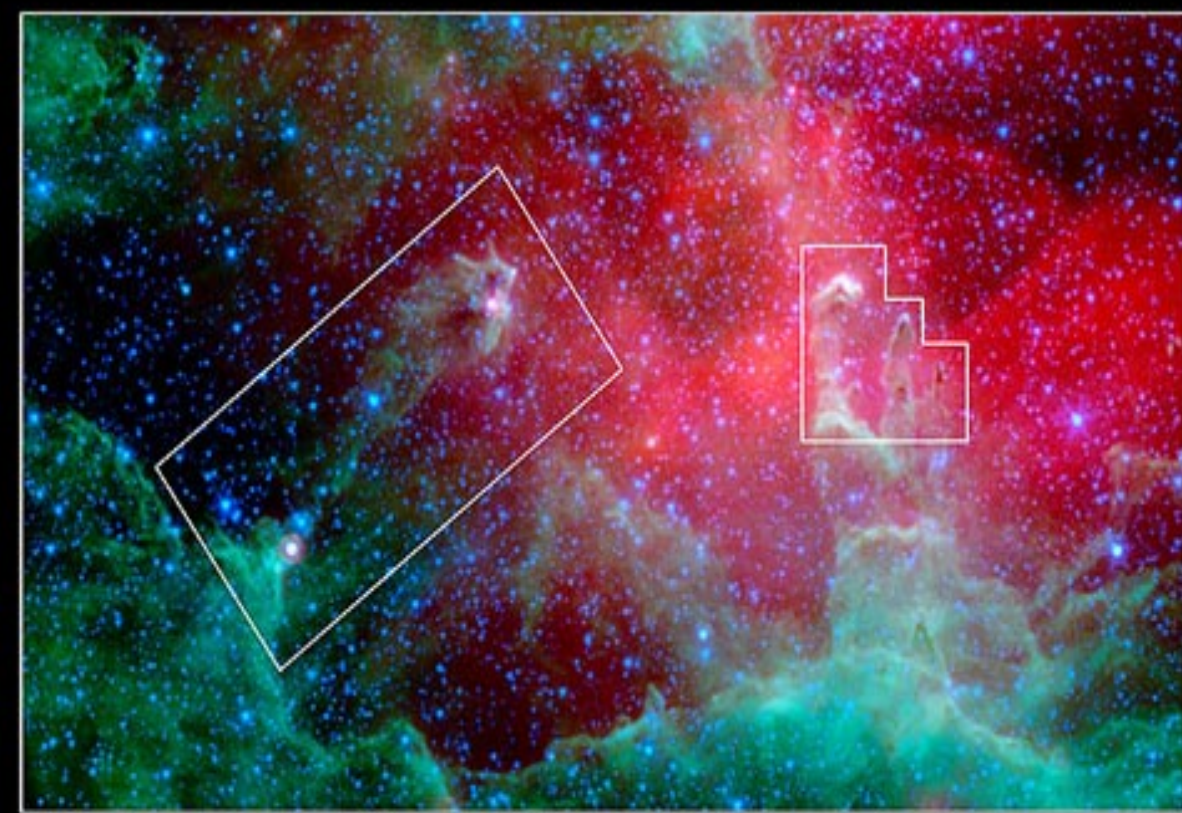
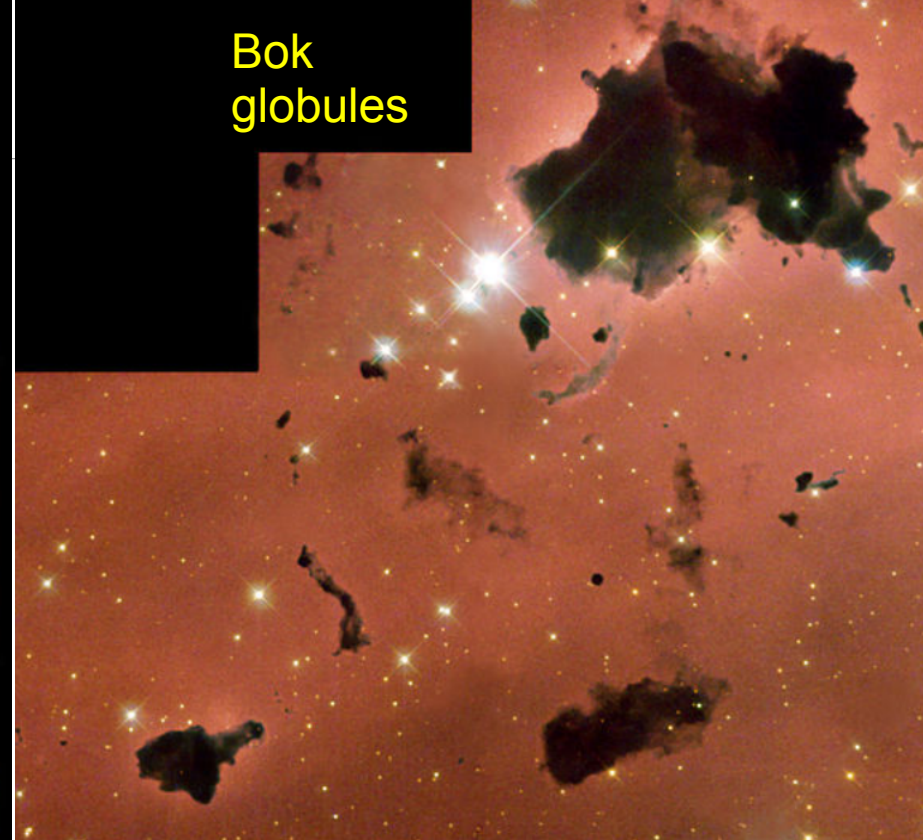
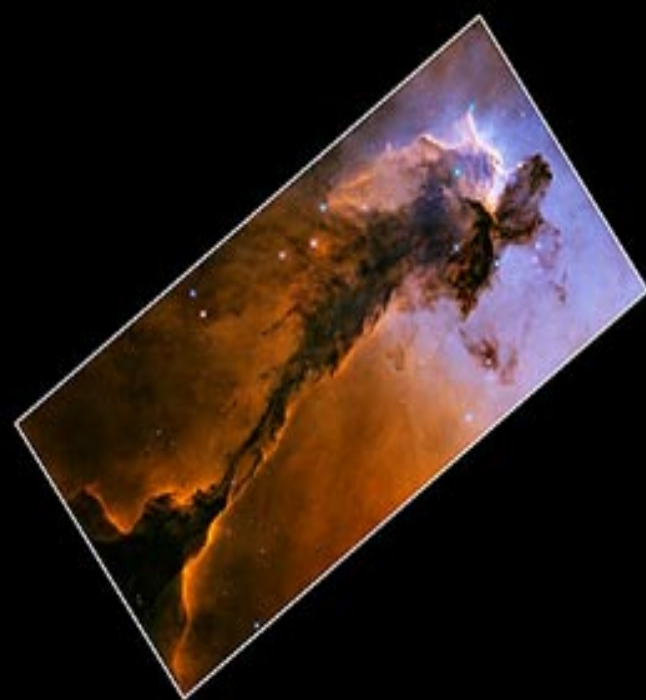
Intermediate-mass stars:

– Initial masses from ~ 2.25 to $8 M_{\text{Sun}}$

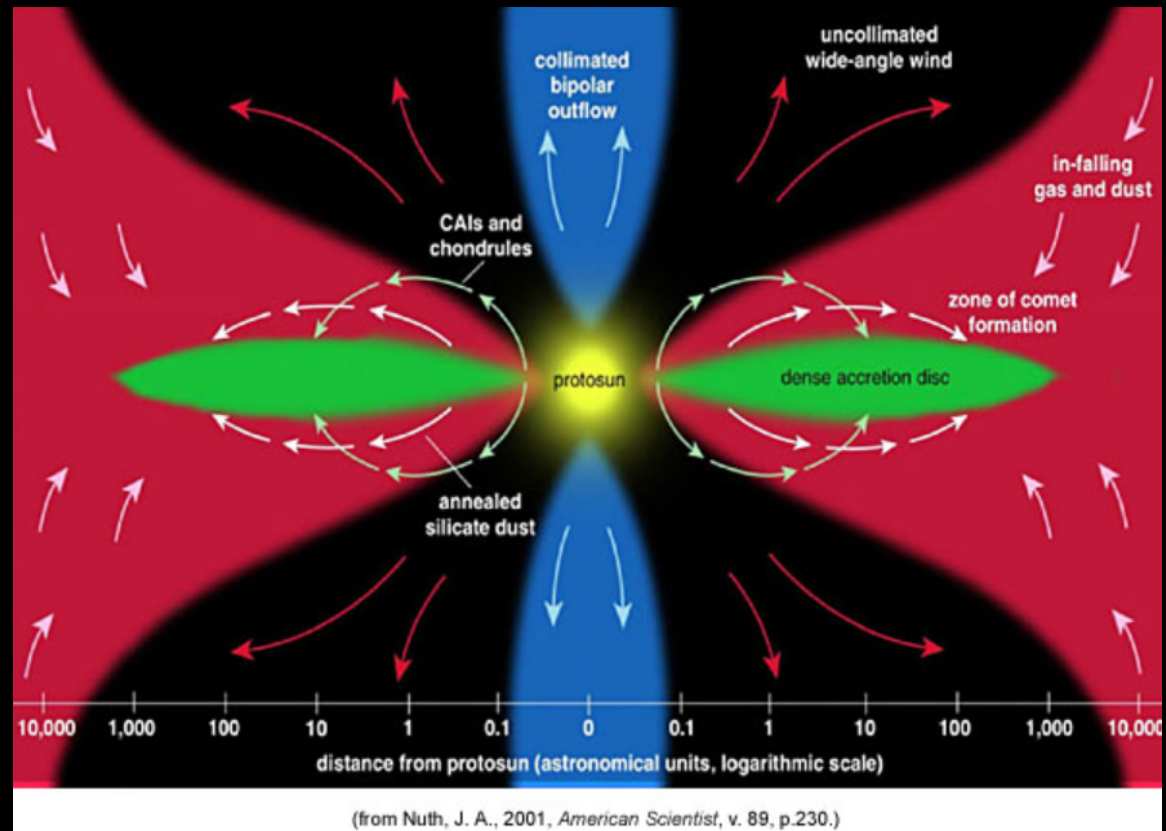
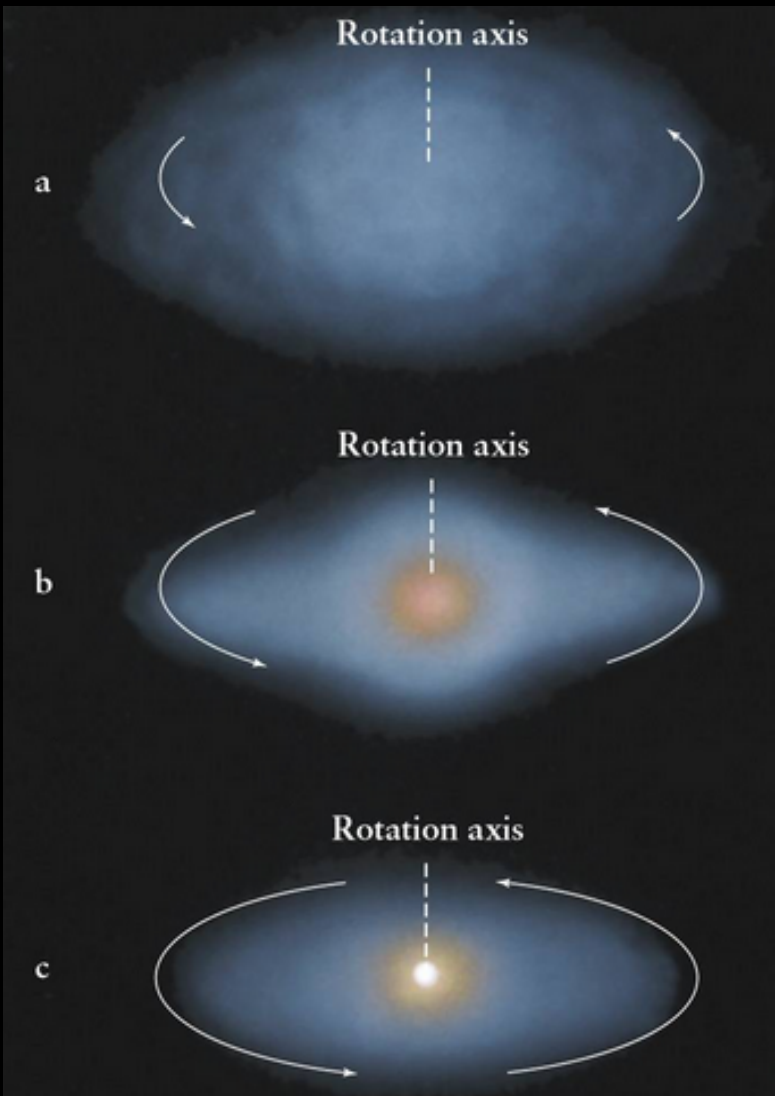
Large stars

– Initial masses above $8 M_{\text{Sun}}$





Protostar nebulae



- Proto-star is heated by a gravitational compression
- Stars below $0.08 M_{\text{Sun}}$ not form
 - Core temp does not reach required for fusion.
 - (brown dwarfs may burn deuterium)
 - ($M_{\text{Jupiter}} \sim 1/1000 M_{\text{Sun}}$)
- How can we observe proto-stars obscured by dust?
 - Infrared observations

Visible



Infrared



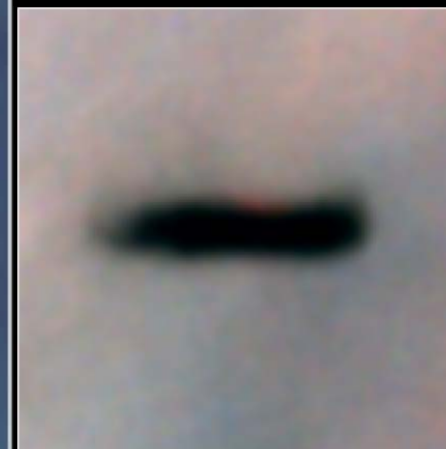
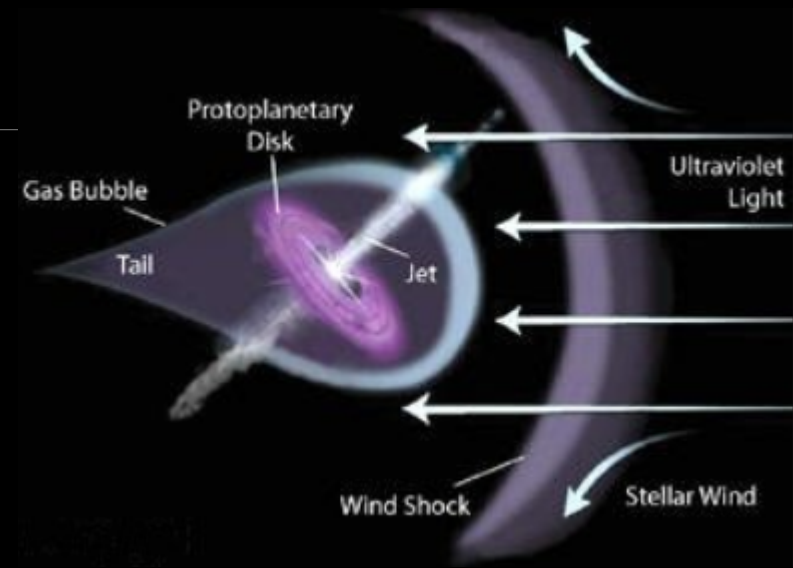
"Starless" Core L1014

Spitzer Space Telescope • IRAC • MIPS

Visible: DSS

ssc2004-20a

NASA / JPL-Caltech / N. Evans (Univ. of Texas at Austin)



**Edge-On Protoplanetary Disk
Orion Nebula**

PRC95-45c · ST ScI OPO · November 20, 1995

M. J. McCaughrean (MPIA), C. R. O'Dell (Rice University), NASA

HST · WFPC2

Protoplanetary Disks Orion Nebula

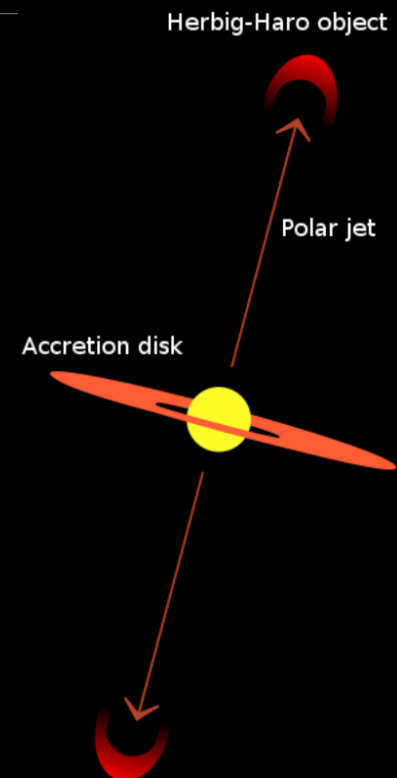
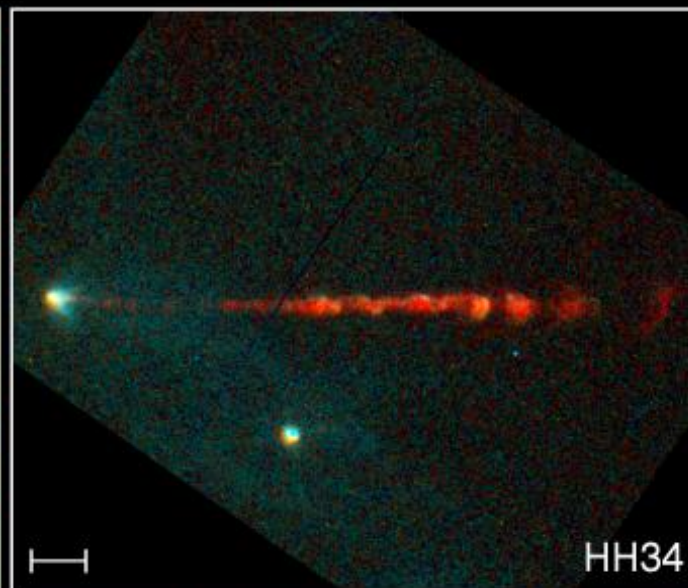
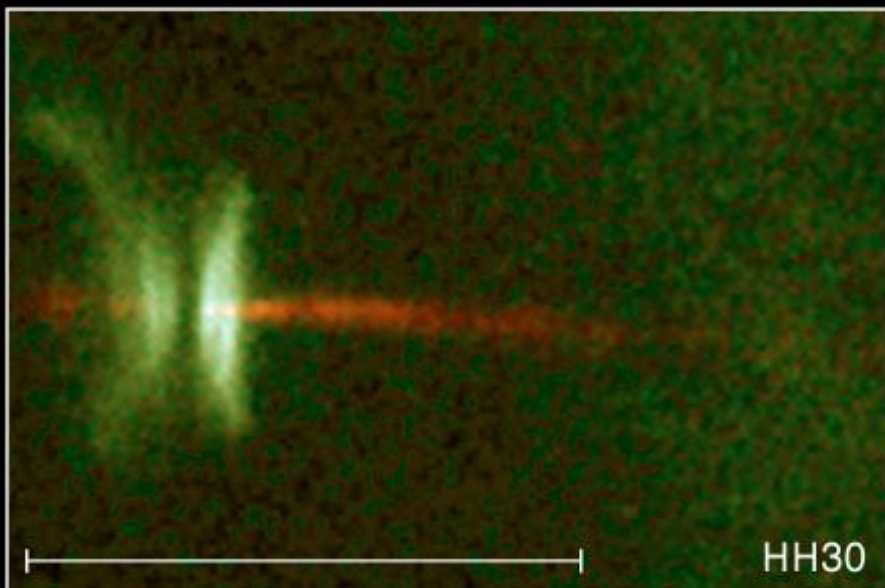
HST · WFPC2

PRC95-45b · ST ScI OPO · November 20, 1995

M. J. McCaughrean (MPIA), C. R. O'Dell (Rice University), NASA



Herbig-Haro objects

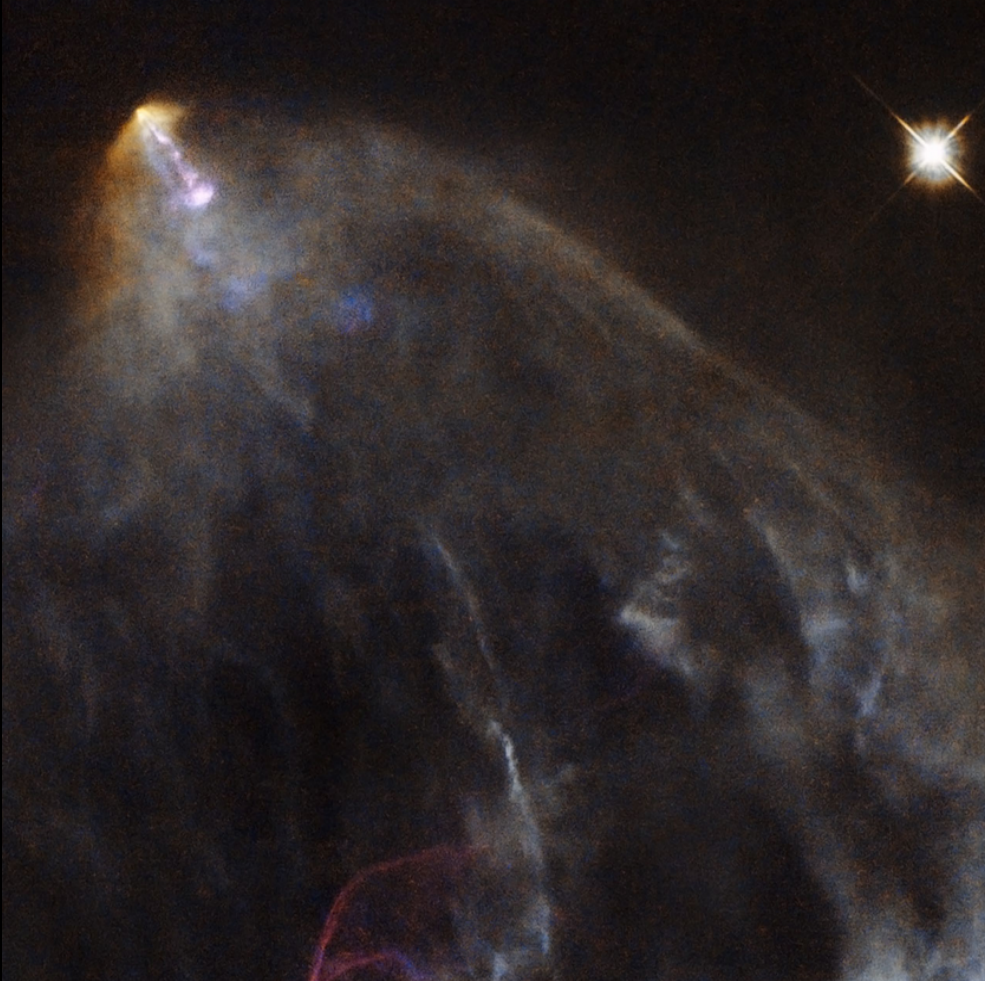


Jets from Young Stars

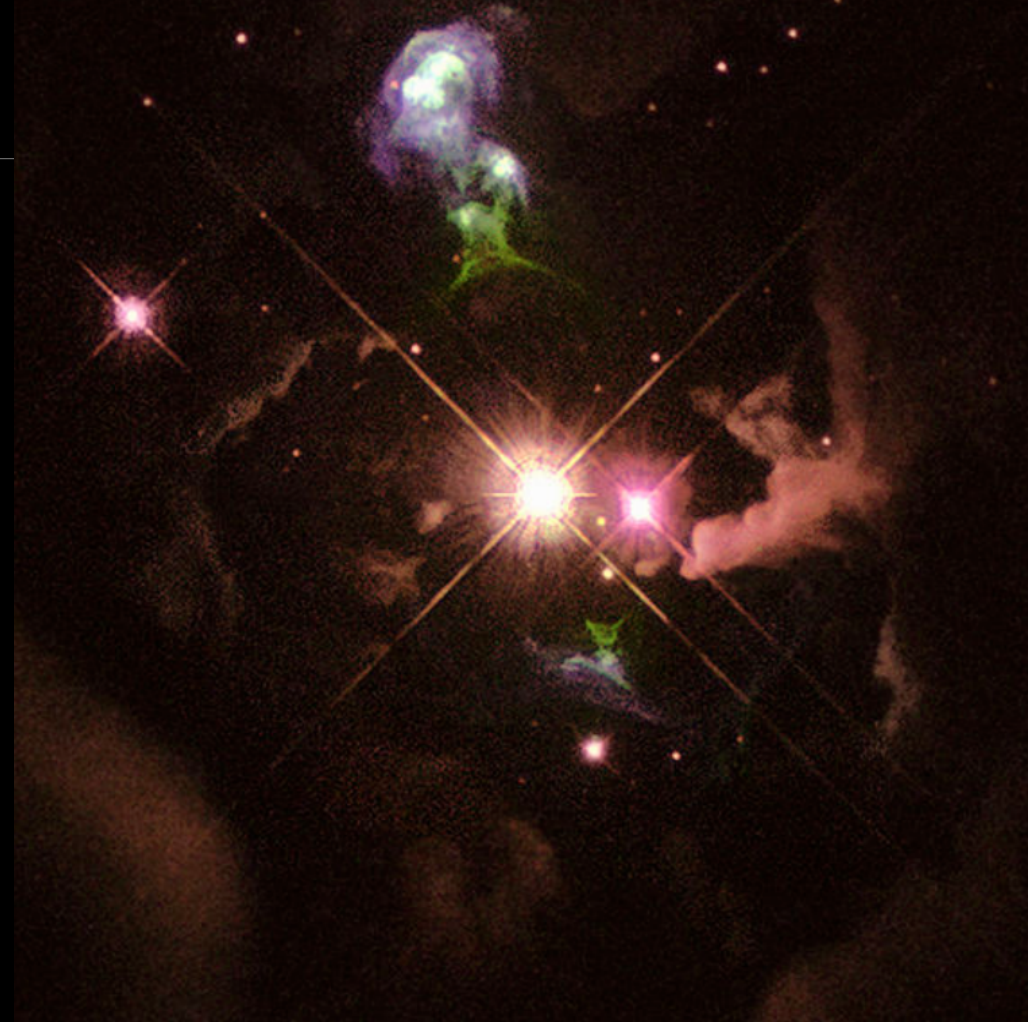
PRC95-24a · ST ScI OPO · June 6, 1995

C. Burrows (ST ScI), J. Hester (AZ State U.), J. Morse (ST ScI), NASA

HST · WFPC2

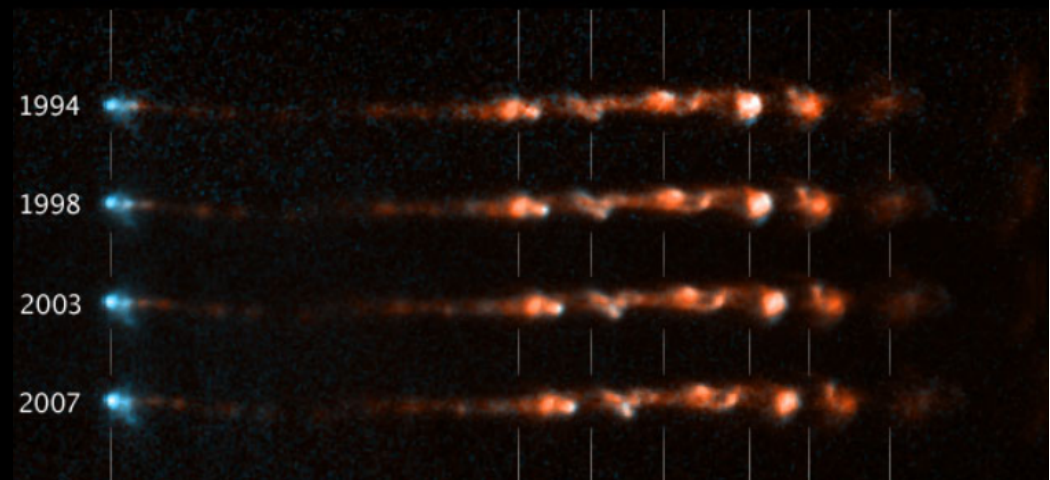


HH 151, a bright jet of glowing material trailed by an intricate, orange-hued plume of gas and dust. It is located some 460 light-years away in the constellation of Taurus , near to the young, tumultuous star HL Tau.



HH 34

Hubble Space Telescope

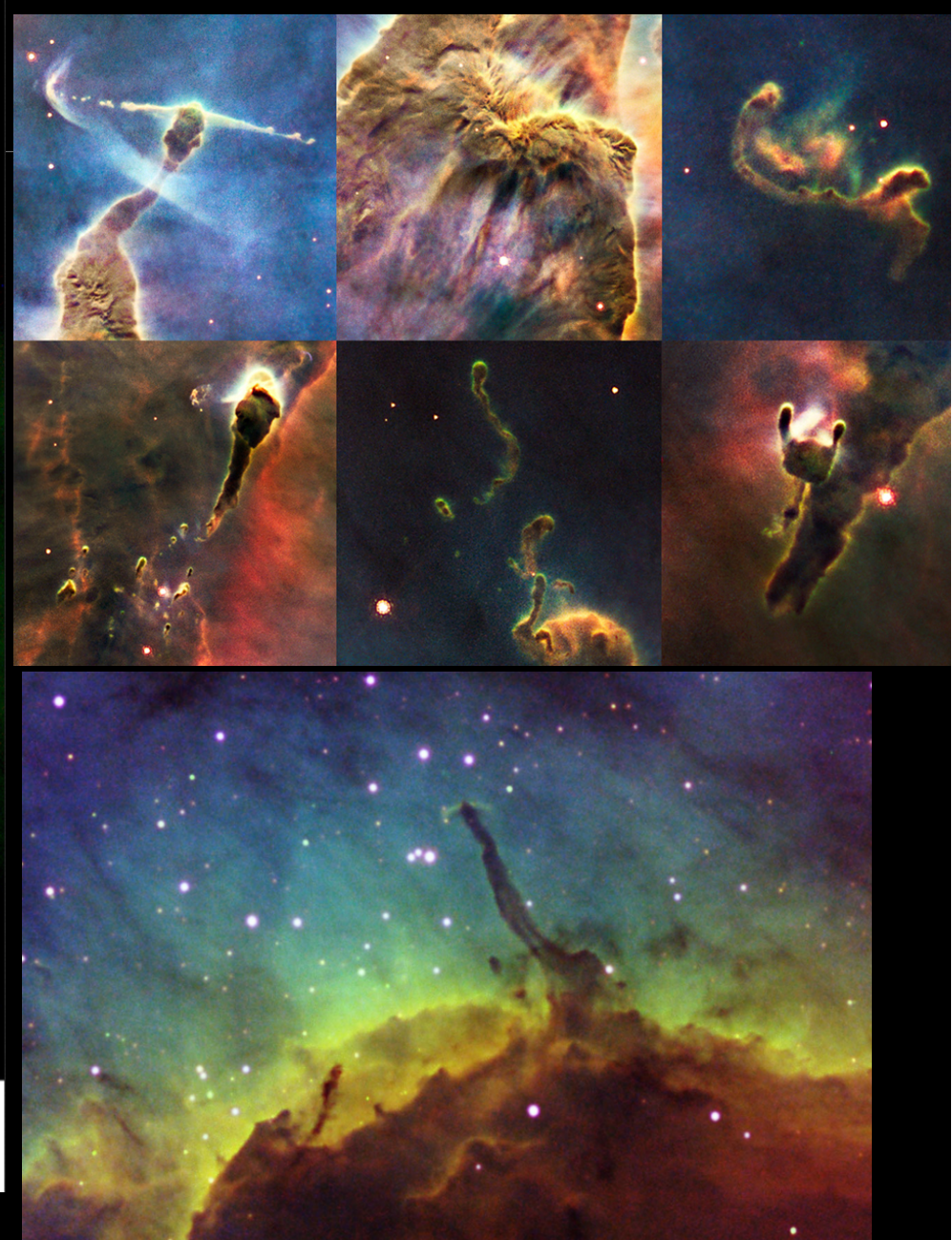




Protostar HH-34 in Orion (VLT KUEYEN + FORS2)

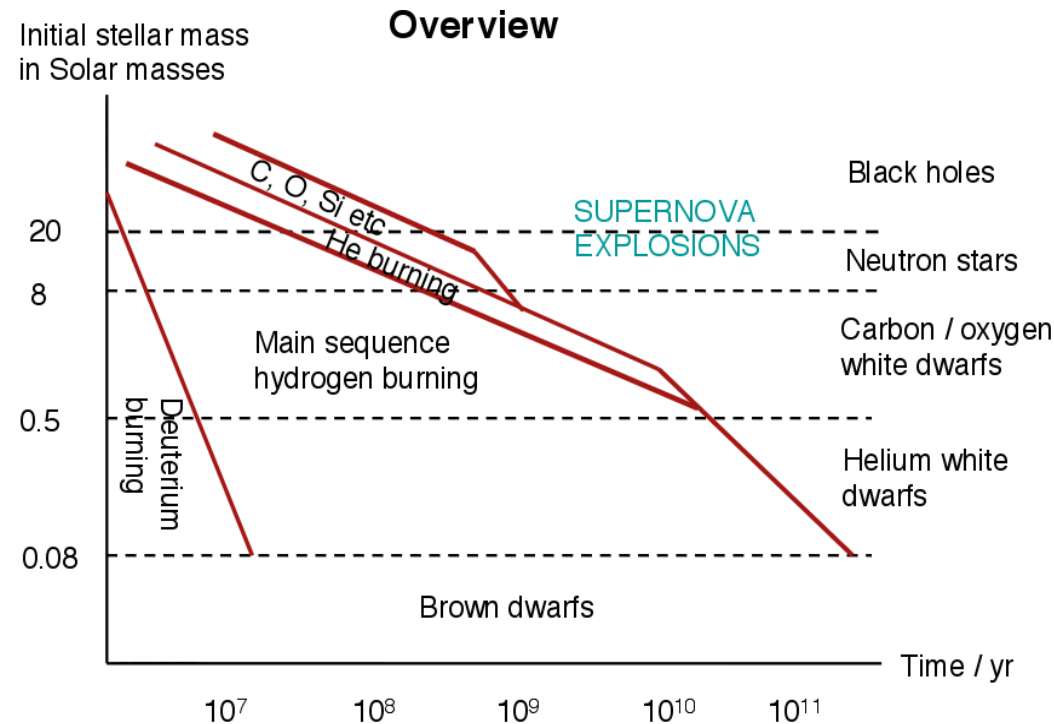
ESO PR Photo 40b/99 (17 November 1999)

© European Southern Observatory



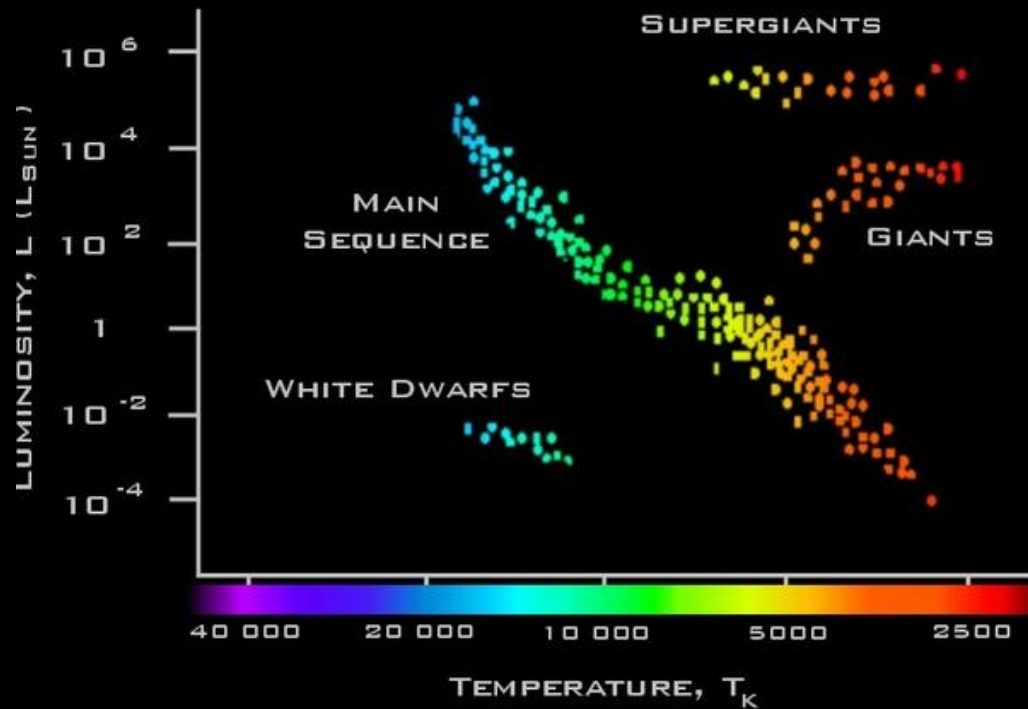
The birth of stars

- Low-mass protostars become stars slowly
 - Weaker gravity causes them to contract slowly, so they heat up gradually
 - Weaker gravity requires low-mass stars to compress their cores more to get hot enough for fusion
 - Low-mass stars have higher density!
- High-mass protostars become stars quickly
 - They contract quickly due to stronger gravity
 - Core becomes hot enough for fusion at a lower density
 - High-mass stars are less dense!

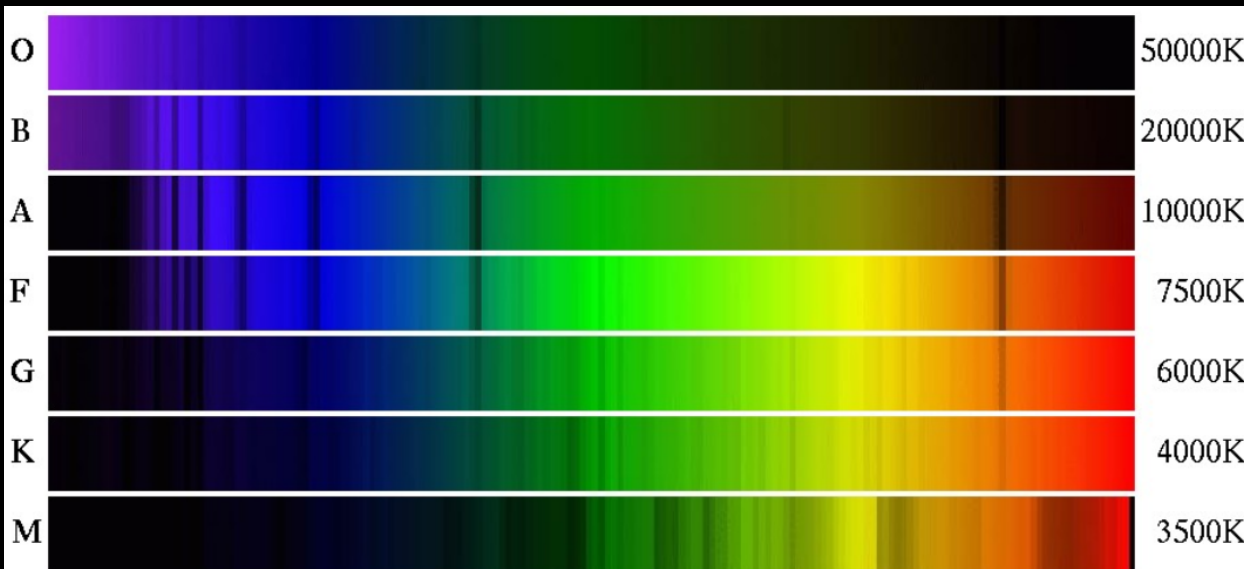


- Deuterium burning allows stars with $M > 2-3 M_{\odot}$ in the pre-main-sequence phase
 - hydrogen burning would occur while the object was still accreting matter
 - deuterium burning acts as a thermostat that stops the central temperature rising above 1 MK
 - after energy transport switches from convective to radiative, forming a radiative barrier around a deuterium exhausted core, central deuterium burning stops
 - then central temperature of the protostar can increase

Hertzsprung - Russell Diagram



- For every massive star, there are ~ 1000 intermediate mass stars and 10000 low mass stars
- About 60% of stars are born in binary systems
 - A small fraction are born in triple and quadruple systems



Class	Color	Prominent Spectral Lines	Surface Temp. (K)
O	Blue	Ionized helium, hydrogen	> 25,000 K
B	Blue-white	Neutral helium, hydrogen	11,000 – 25,000 K
A	White	Hydrogen, ionized sodium and calcium	7,500 – 11,000 K
F	White	Hydrogen, ionized and neutral sodium and calcium	6,000 – 7,500 K
G	Yellow	Neutral sodium and calcium, ionized calcium, iron, magnesium	5,000 – 6,000 K
K	Orange	Neutral calcium, iron, magnesium	3,500 – 5,000 K
M	Red	Neutral iron, magnesium, and neutral titanium oxide	< 3,500 K

Main sequence

- Stars enter the main-sequence phase when the fusion of H to He begins
- List in order from lowest to highest temp requirement: Triple alpha, CNO, proton-proton fusion
 - P-P ~ 5 million Kelvin (H->He)
 - CNO ~ 20 million Kelvin (H->He)
 - Triple alpha ~ 100 million Kelvin (He->C)
- Why are main sequence stars so stable?
 - Compression -> $T \uparrow$ -> fusion \uparrow -> $P \uparrow$ -> expansion
 - Expansion -> $T \downarrow$ -> fusion \downarrow -> $P \downarrow$ -> compression

Eulerian coordinates (r, t)	Lagrangian coordinates (m, t)
-------------------------------	---------------------------------

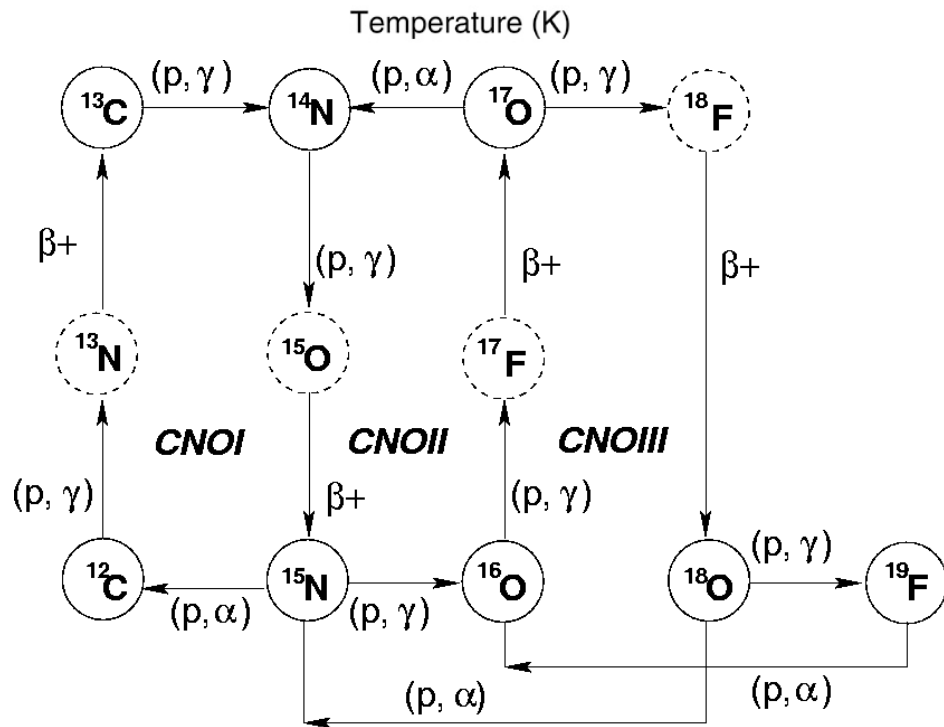
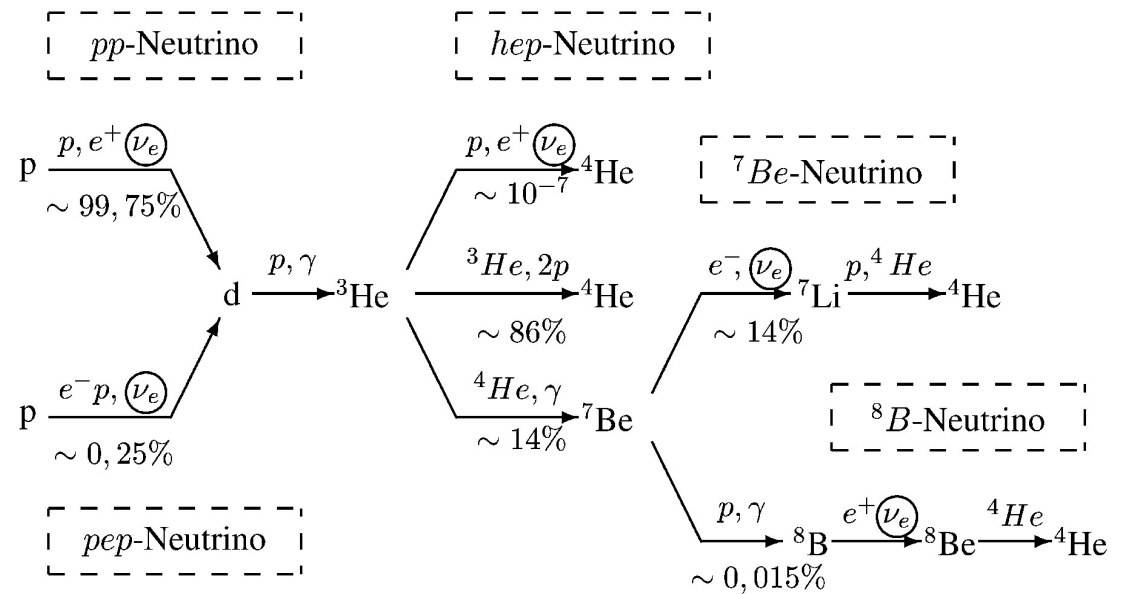
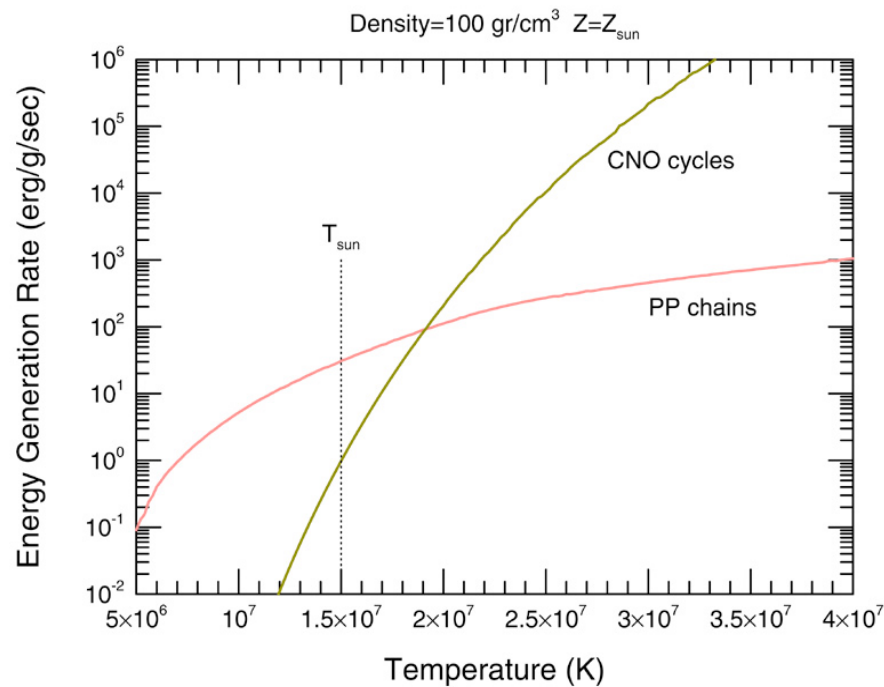
$$\frac{dm}{dr} = 4\pi r^2 \rho$$

$$\frac{dP}{dr} = -\frac{Gm\rho}{r^2}$$

$$\frac{dr}{dm} = \frac{1}{4\pi r^2 \rho}$$

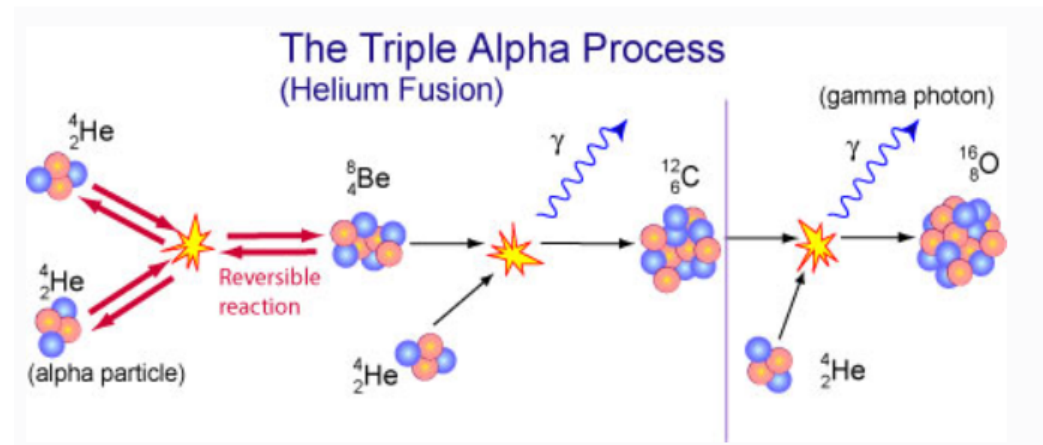
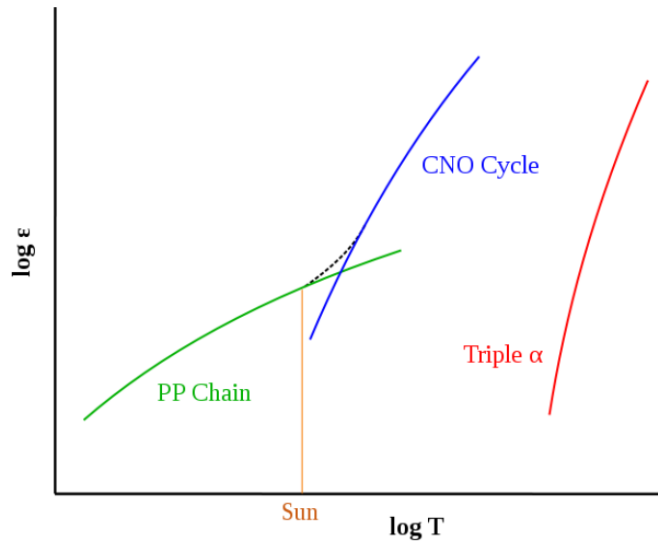
$$\frac{dP}{dm} = -\frac{Gm}{4\pi r^4}$$

pp and CNO cycles



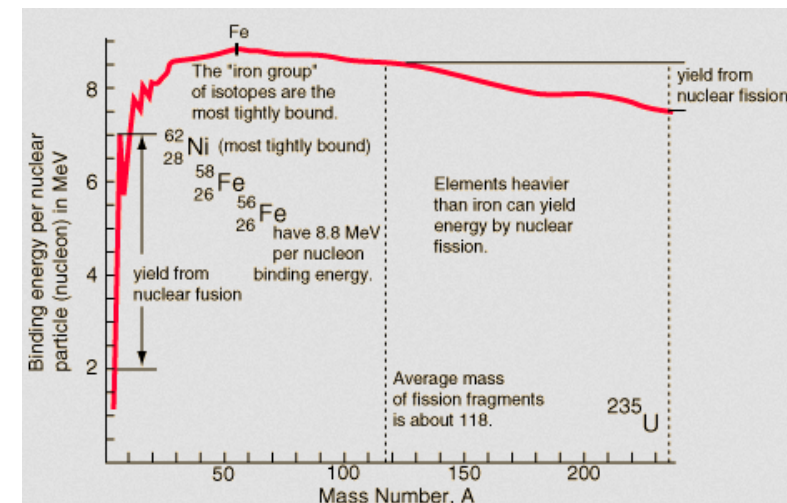
Element	Number %	Mass %
Hydrogen	92.0	73.4
Helium	7.8	25.0
Carbon	0.02	0.20
Nitrogen	0.008	0.09
Oxygen	0.06	0.8
Neon	0.01	0.16
Magnesium	0.003	0.06
Silicon	0.004	0.09
Sulfur	0.002	0.05
Iron	0.003	0.14

Triple-alpha



- $\epsilon(\text{pp}) \sim T^4$
- $\epsilon(\text{CNO}) \sim T^{17}$
- $\epsilon(\text{triple } \alpha) \sim T^{40}$

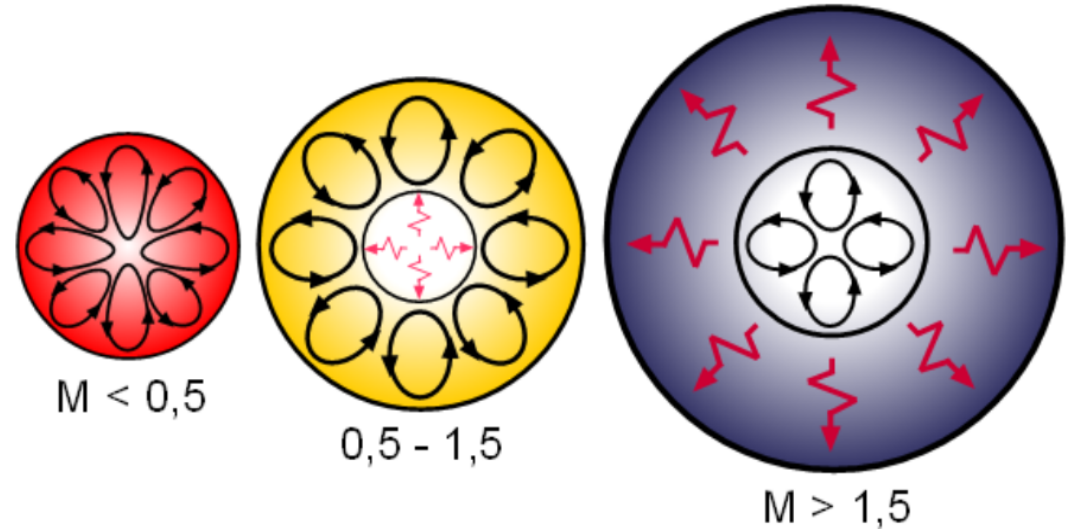
Nuclear Fuel	Process	$T_{\text{threshold}} 10^6\text{K}$	Products	Energy per nucleon (MeV)
H	PP	~4	He	6.55
H	CNO	15	He, N	6.25
He	3α	100	C, O	0.61
C	C+C	600	O, Ne, Na, Mg	0.54
O	O+O	1000	Mg, S, P, Si	~0.3
Si	Nuc eq.	3000	Co, Fe, Ni	<0.18



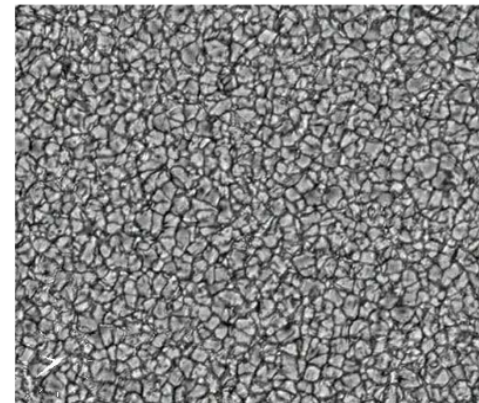
Energy transport in stars

Energy is transported by these mechanisms:

- **Radiation** - photons carry energy away from the star's center
- **Convection** - cells of hot gas move up, cells of cool gas move down (boil)
- **Conduction** - collisions between electrons can move energy outwards, relevant in degenerate matter
- **Neutrino radiation** – advanced burning stages, supernovae



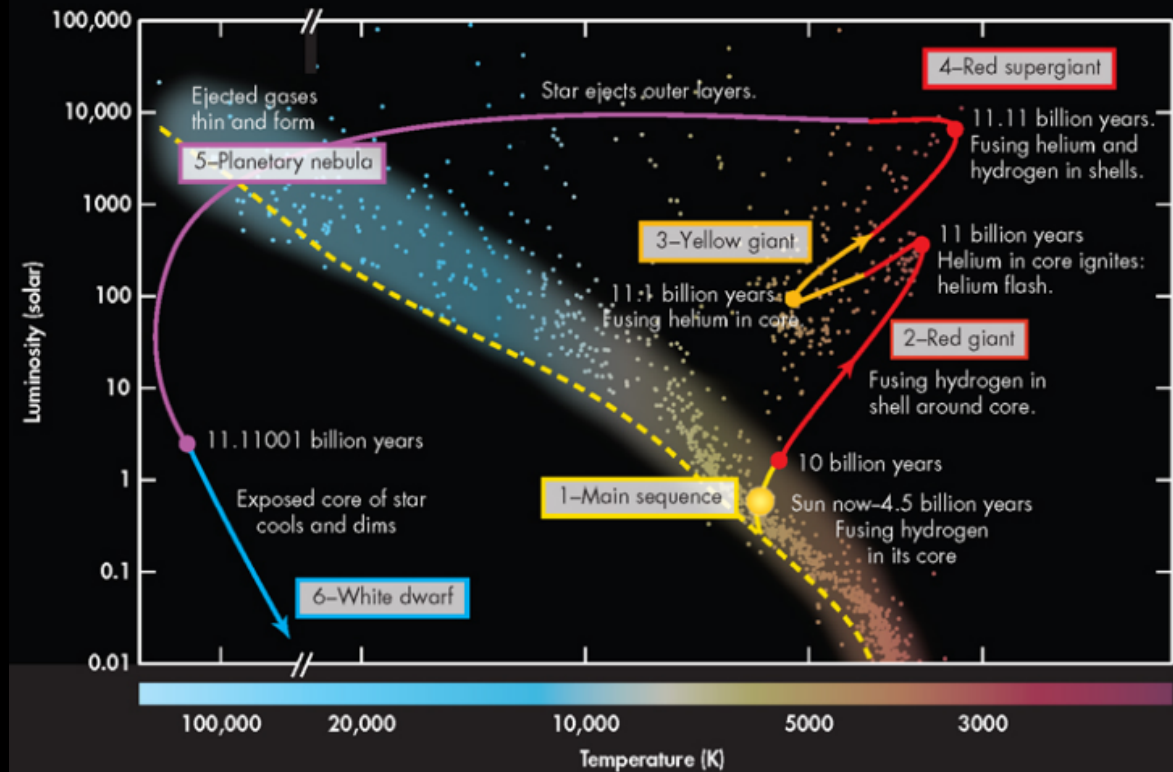
- Convection takes over once radiation becomes inefficient.
- Radiation becomes noneffective when there is a large temperature gradient. ($\sim T^4$, $\sim T^{17}$, $\sim T^{40}$)
- Efficiency of the radiative transport also depends on the opacity



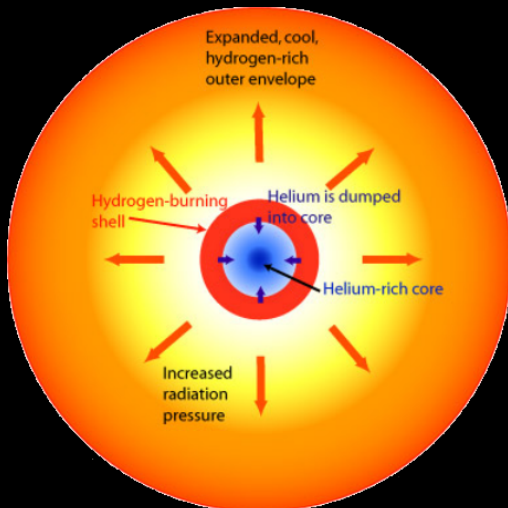
$\Delta T \simeq 100 \text{ K}$

$v \sim \text{few km/s}$

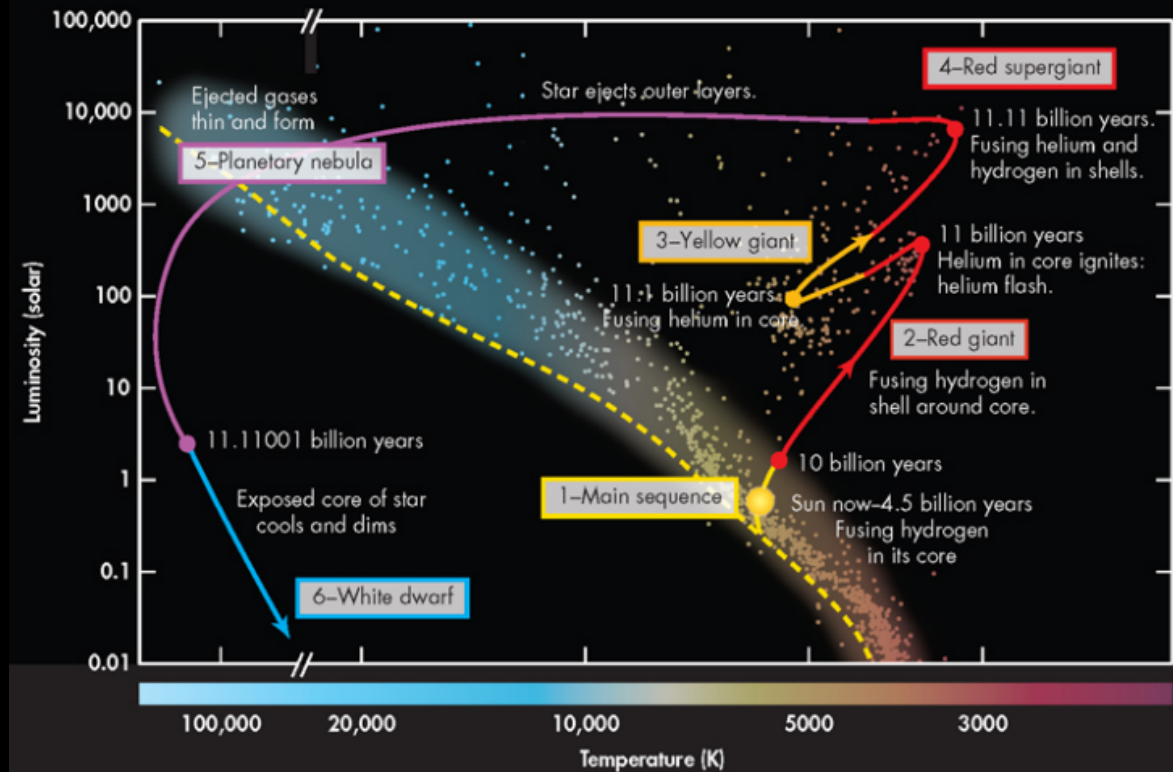
Red Giant Branch



- H in core has become depleted
- H shell burns
- As the core continues to shrink, the outer layers of the star expand and cool
- It is now a red giant, extending out as far as the orbit of Mercury
- Despite its cooler temperature, its luminosity increases enormously due to its larger size
- First dredge-up occurs: The convection in the envelope moves in when the stars is near the bottom of the RGB and "dredges up" material that has been through partial hydrogen burning by the CNO cycle and pp chains.

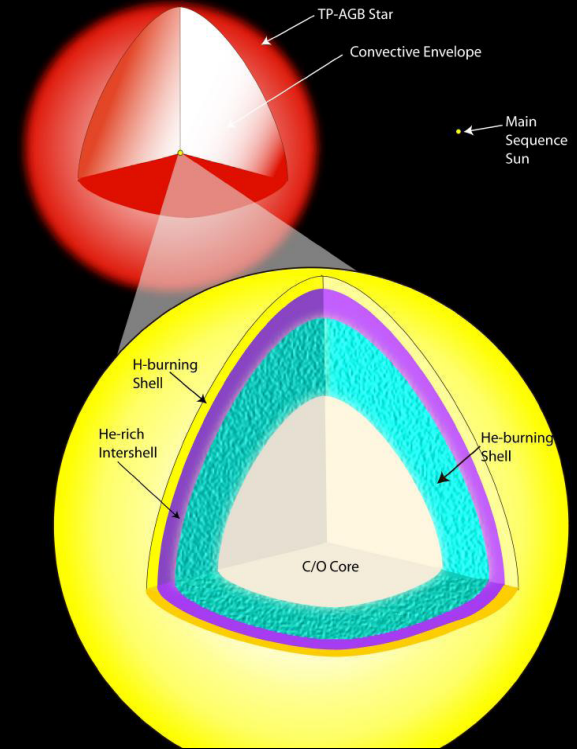
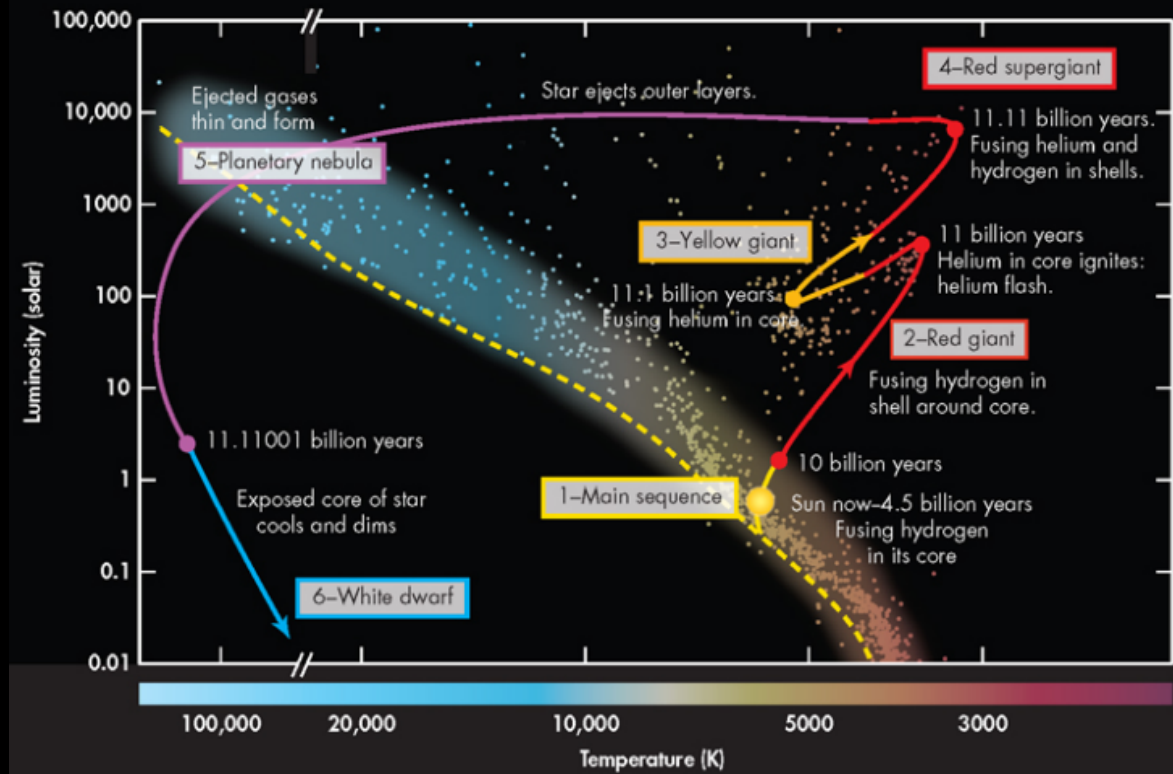


Helium flash and horizontal branch



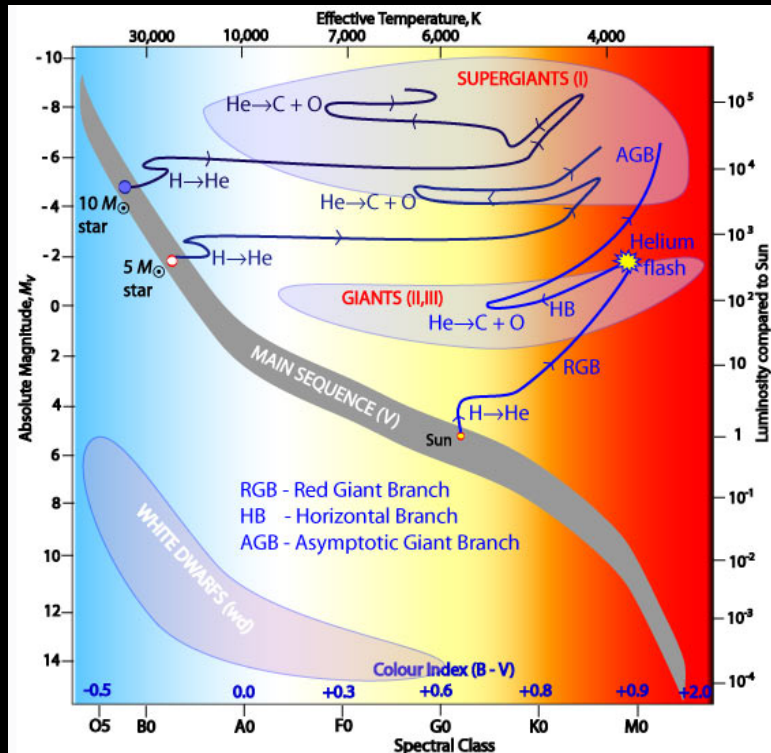
- Low-mass stars ($0.8 \sim 2.2 M_{\odot}$) have He degenerate core
- Density 10^6 g cm^{-3}
- As M_{core} increases, it shrinks
- Increase in T = increase in p , degeneracy pressure is not dependent on T
- He fusion commences at 10^8 K
- Helium begins to fuse extremely rapidly (triple alpha $\sim T^{40}$)
- H fusion in shell is extinguished by this source of energy, it takes $O(10^3)$ years to reignite
- High Z stars occupy region known as red clump
- Low Z live in the horizontal branch

Asymptotic Giant

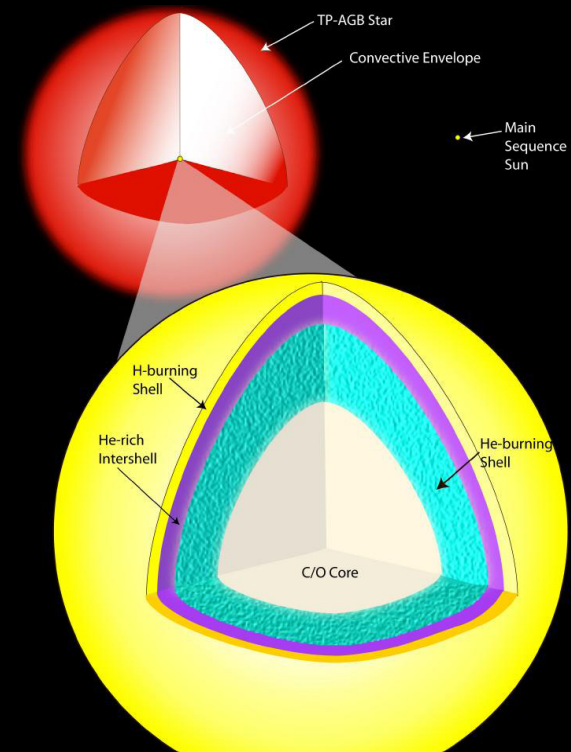


- Final nuclear burning phase for all stars with masses 0.8 to 8 M_{sun}
- After core He-burning, the C-O core contracts and star becomes a giant again
- Lasts less than 1% of the main-sequence lifetime
- Helium begins to fuse extremely rapidly (triple alpha $\sim T^{40}$)
- At the end – pulsations as the He and H burning shells get close together, mixing
- Strong stellar winds blow away outer shells (pulsations, dust)
- The core left becomes white dwarf

Masses above $2.2 M_{\odot}$



- Core non-degenerate, no helium flash
- Develop a CO core, in AGB phase shed their envelopes
- IF $M > 8 M_{\odot}$, carbon burning
- White dwarf + nebula result
- IF $M > \sim 11 M_{\odot}$, up to Fe



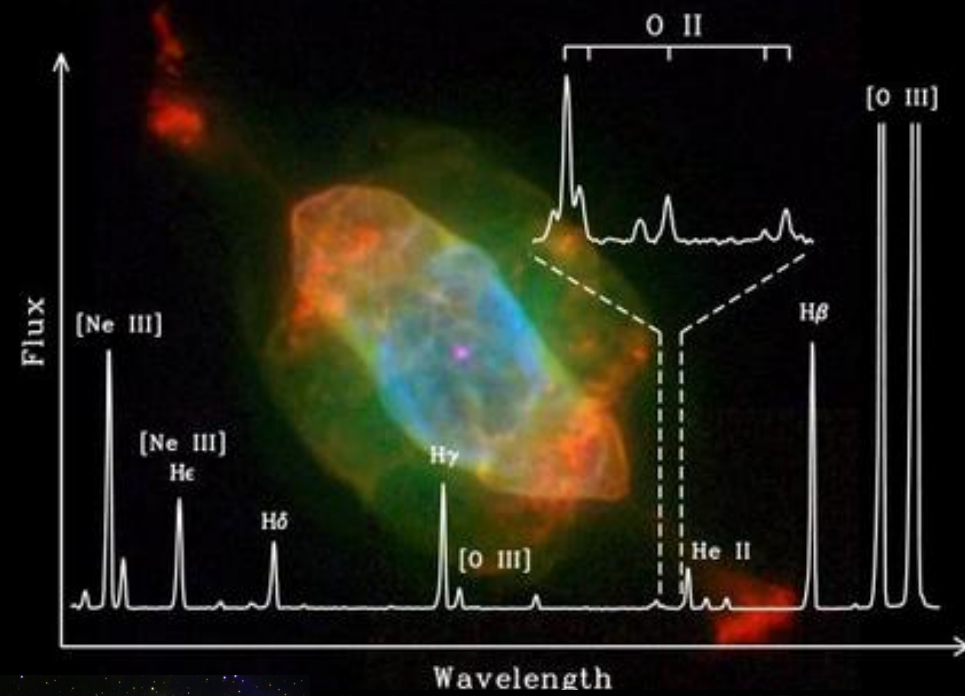
NGC 7009



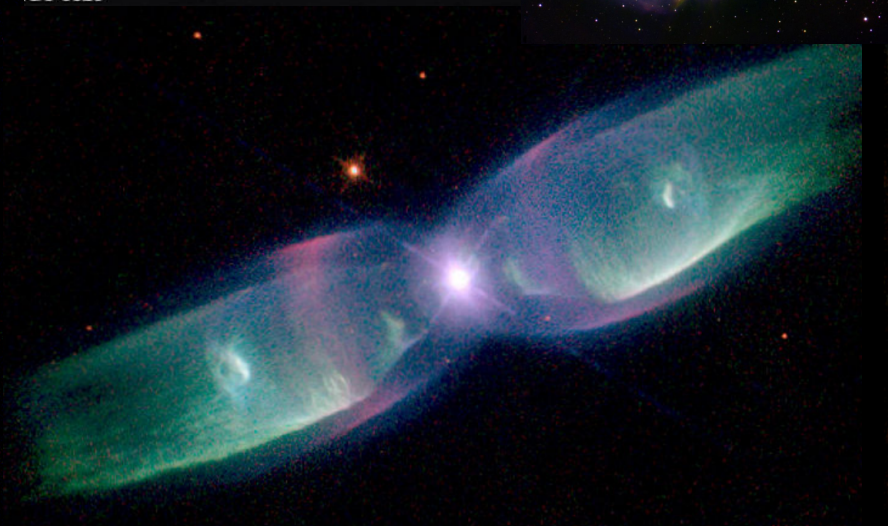
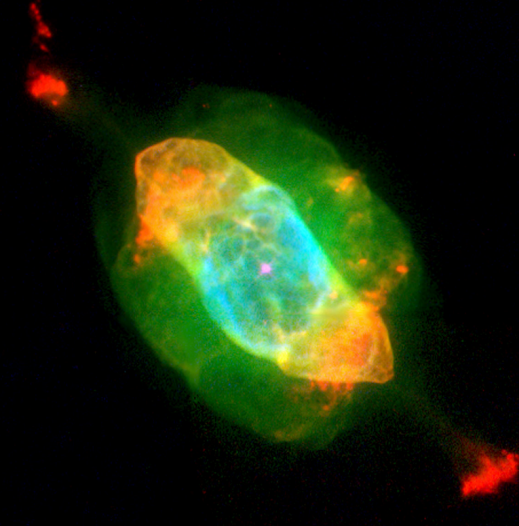
CAT'S EYE



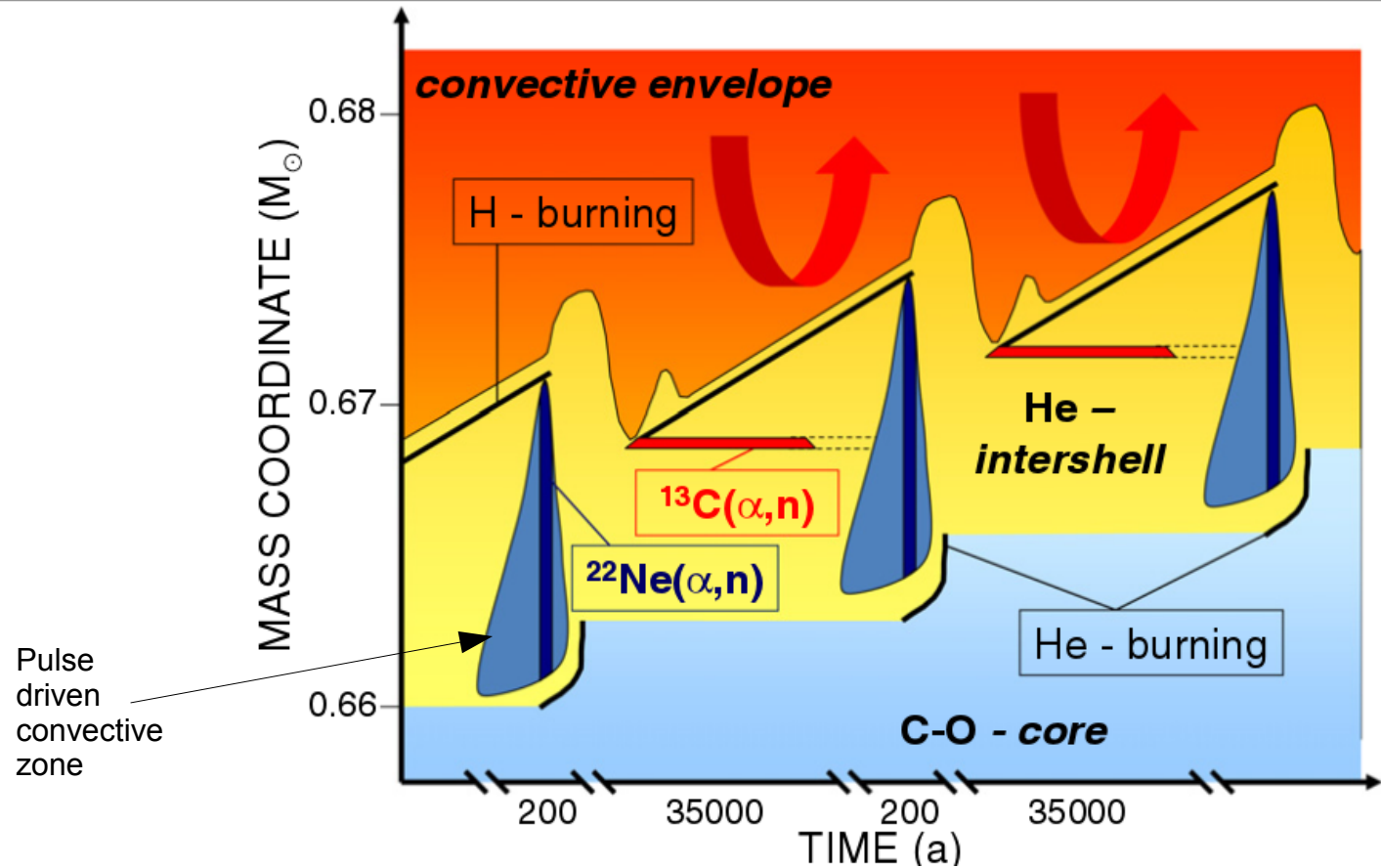
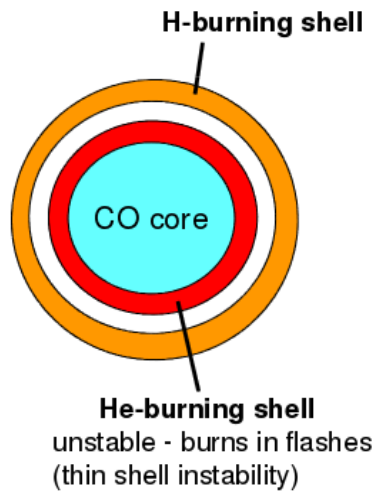
NGC 7662



NGC 6826

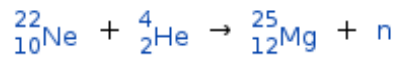
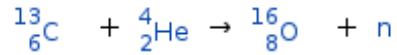


TP-AGB

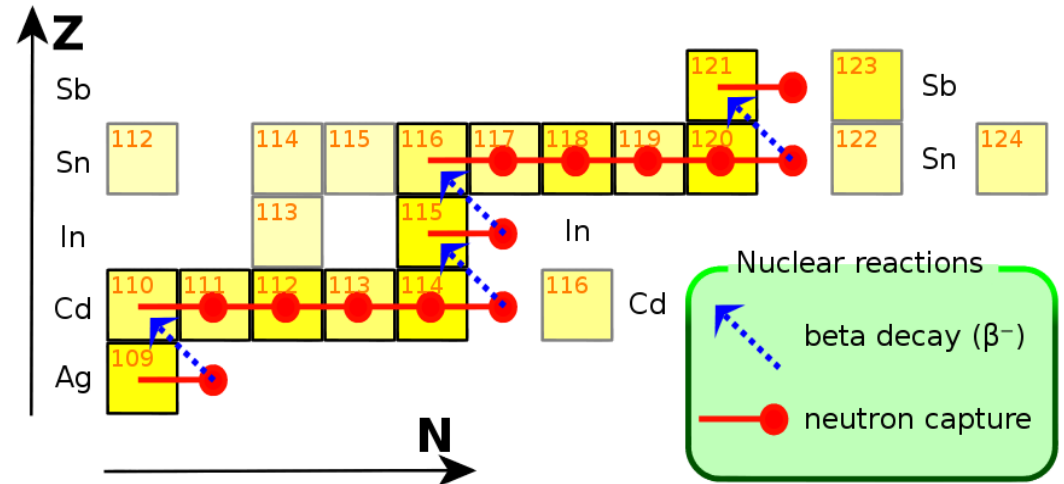
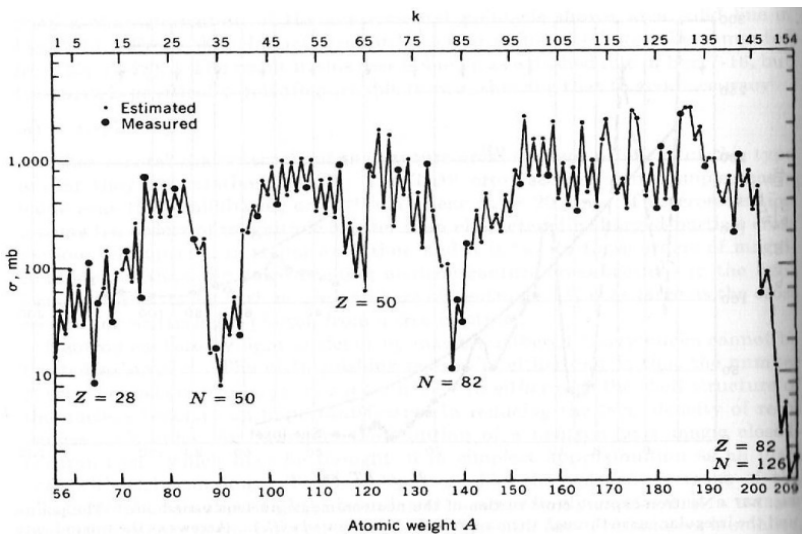


- H burning produces ashes in the intershell
- When enough He is accumulated in the intershell, He Shell Explosively fuses
- This thermonuclear runaway drives formation of a convective zone
- Expansion powered by convection kills H burning layer
- Accumulation of CO elements, protons mixed downward from convective envelope
- Regions heat during interpulse, creating ^{13}C pocket ($^{12}\text{C}(p, \gamma)^{13}\text{N}(\beta + \nu)^{13}\text{C}$)
- This carbon isotope waits until next pulse... $^{13}\text{C}(\alpha, n)^{16}\text{O}$
- Intense neutron flux $\sim 10^7 \text{ cm}^{-3}$
- This pocket is strongly enriched in newly created elements

The s-process

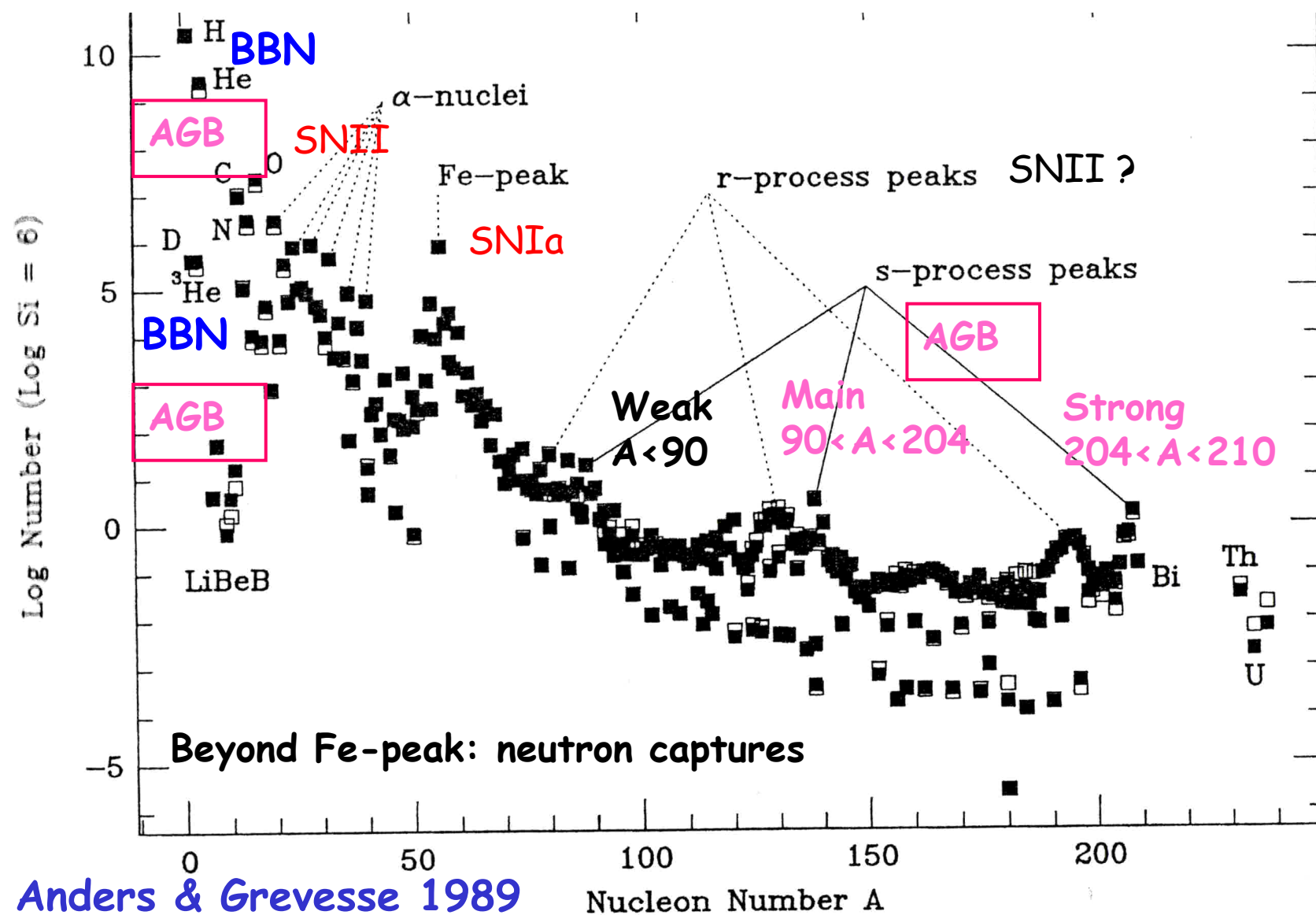


Neutron captures are not hindered by coulomb repulsion



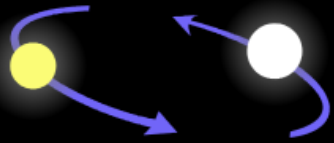
$$\frac{dN_A}{dt} = - \langle \sigma v \rangle_A n_n(t) N_A(t) + \langle \sigma v \rangle_{A-1} n_n(t) N_{A-1}(t)$$

- Neutron captures increase the mass number through (n,γ) reactions
- Time scale for n capture << Beta decay times
- The products will beta-decay until a stable (or long-lived) isotope is reached
- Nuclei with magic number of neutrons stable against neutron capture, act as bottlenecks
- The process thus terminates in ${}^{209}\text{Bi}$, the heaviest "stable" element.

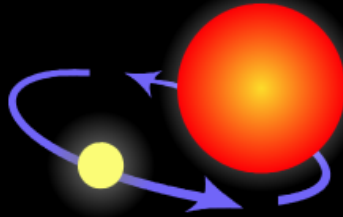


Anders & Grevesse 1989
Cameron 1982

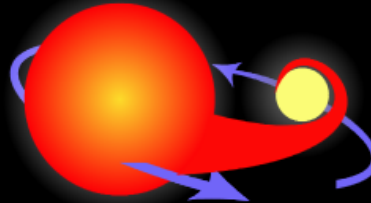
The progenitor of a Type Ia supernova



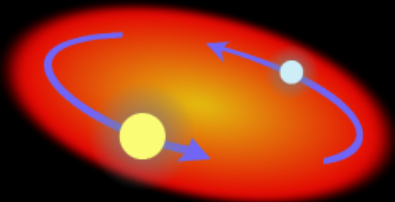
Two normal stars are in a binary pair.



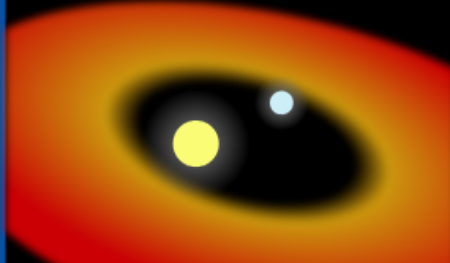
The more massive star becomes a giant...



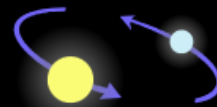
...which spills gas onto the secondary star, causing it to expand and become engulfed.



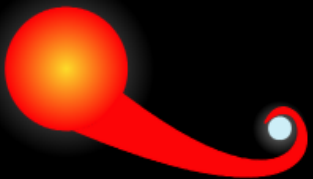
The secondary, lighter star and the core of the giant star spiral toward within a common envelope.



The common envelope is ejected, while the separation between the core and the secondary star decreases.



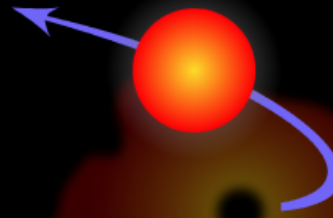
The remaining core of the giant collapses and becomes a white dwarf.



The aging companion star starts swelling, spilling gas onto the white dwarf.



The white dwarf's mass increases until it reaches a critical mass and explodes...



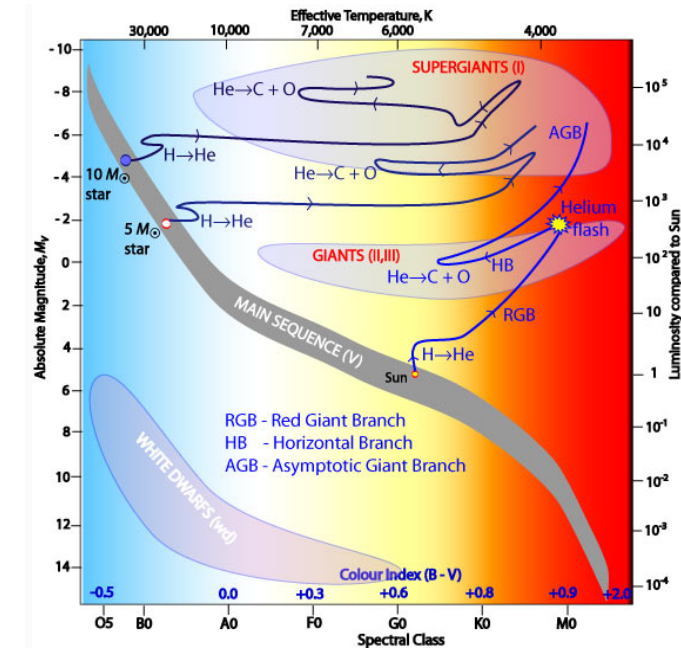
...causing the companion star to be ejected away.

Start off with a binary system

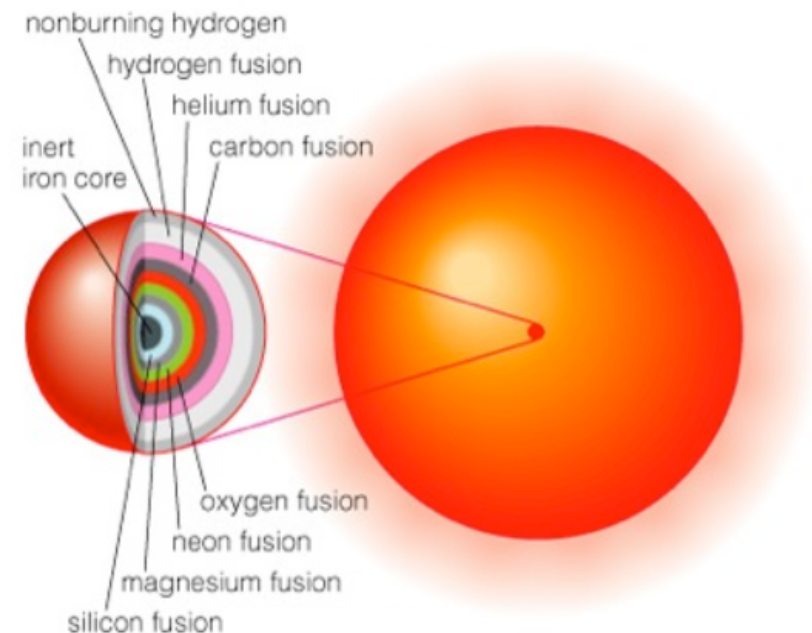
- One star comes to end of its life – forms a “white dwarf” (made of helium, or carbon/oxygen)
- If close enough, tidal forces of the white dwarf pull matter off the other star... adds to mass of white dwarf (accretion)
- If accretion pushes white dwarf over the Chandrasekhar Mass Limit ($1.4 M_{\odot}$), it starts to collapse
- Rapid compression initiates thermonuclear runaway... burn whole mass of white dwarf to iron/nickel in few seconds
- Liberated energy blows the star apart... no remnant.

Supergiant stage of heavy stars

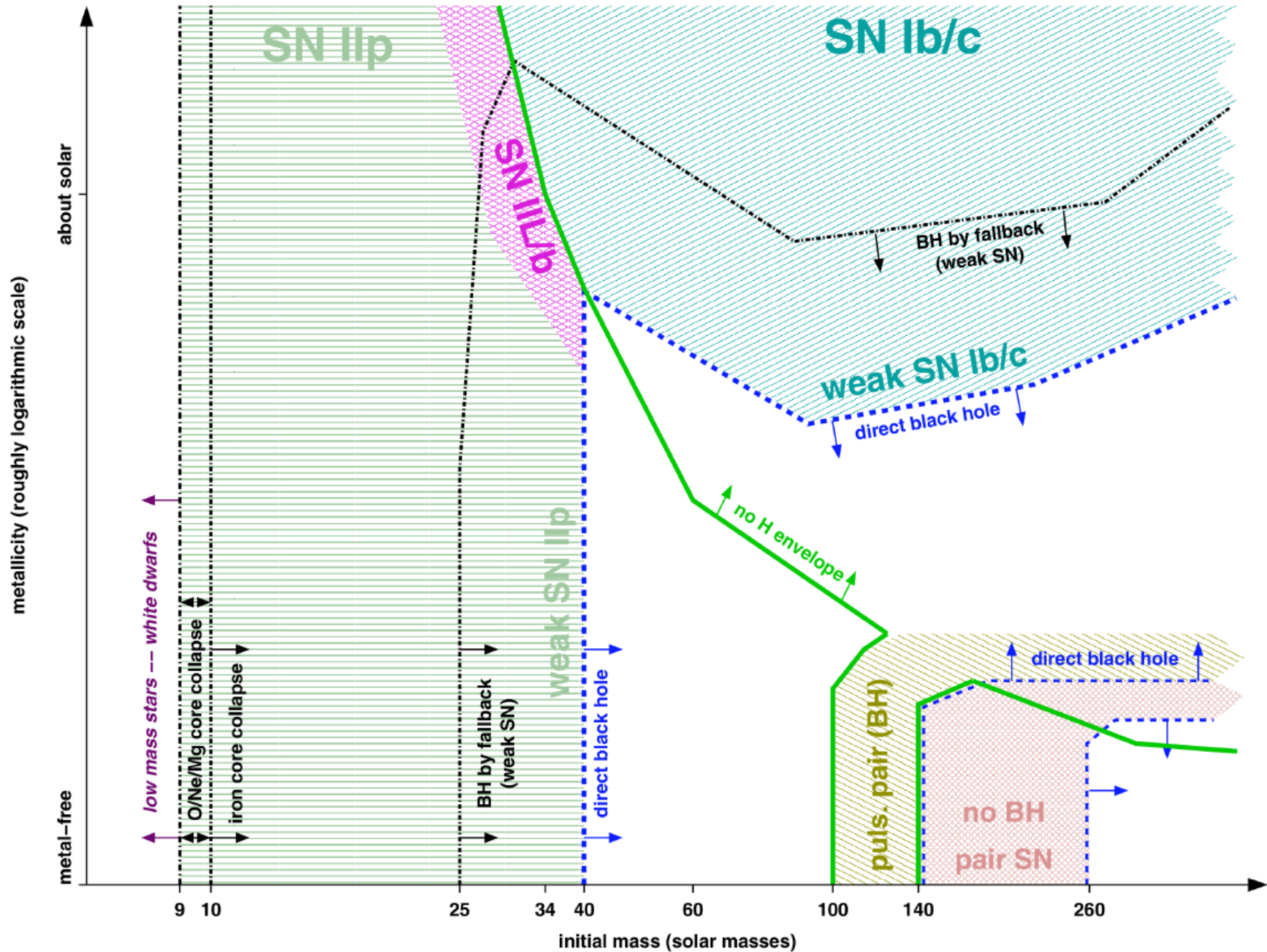
- $L \propto M^{3-4}$, $\text{Fuel} \propto M \Rightarrow \text{Lifetime} \propto 1/M^2 (M/M^3)$
- Massive stars live fast, die young
- Higher $T \Rightarrow$ CNO Cycle on MS
- No core degeneracy (higher $T \Rightarrow$ higher P) means no helium flash
- Higher $T \Rightarrow$ more nuclear burning stages (it is the high mass stars that make the elements heavier than C,N,O and even most of the C,N,O that goes into the ISM.
- Star goes through stages of H core burning, H shell burning and He core fusion as medium mass star
- Then it continues with further stages up to Iron



Nuclear Fuel	Process	$T_{\text{threshold}} 10^6 \text{K}$	Products	Energy per nucleon (MeV)
H	PP	~4	He	6.55
H	CNO	15	He, N	6.25
He	3α	100	C, O	0.61
C	C+C	600	O, Ne, Na, Mg	0.54
O	O+O	1000	Mg, S, P, Si	~0.3
Si	Nuc eq.	3000	Co, Fe, Ni	<0.18



Massive star fate

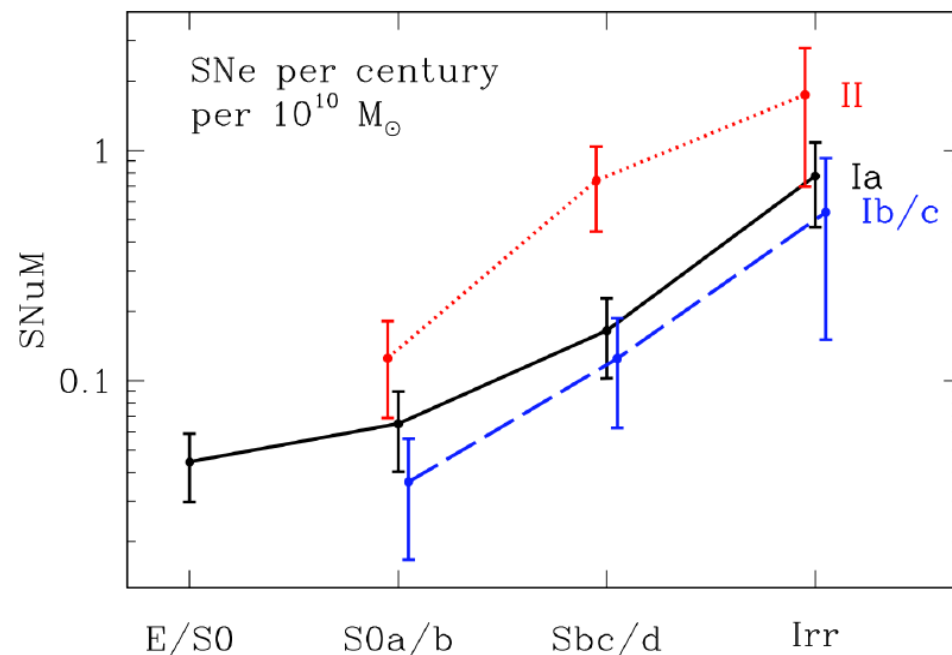


Released energy

Energetics of supernovae

Supernova	Approximate total energy (foe) ^c	Ejected Ni (solar masses)	Neutrino energy (foe)	Kinetic energy (foe)	Electromagnetic radiation (foe)
Type Ia ^{[97][98][99]}	1.5	0.4 - 0.8	0.1	1.3 - 1.4	~0.01
Core collapse ^{[100][101]}	100	(0.01) - 1	100	1	0.001 - 0.01
Hypernova	100	~1	100	1	~0.1
Pair instability ^[77]	5-100	0.5 - 50	low?	1-100	0.01 - 0.1

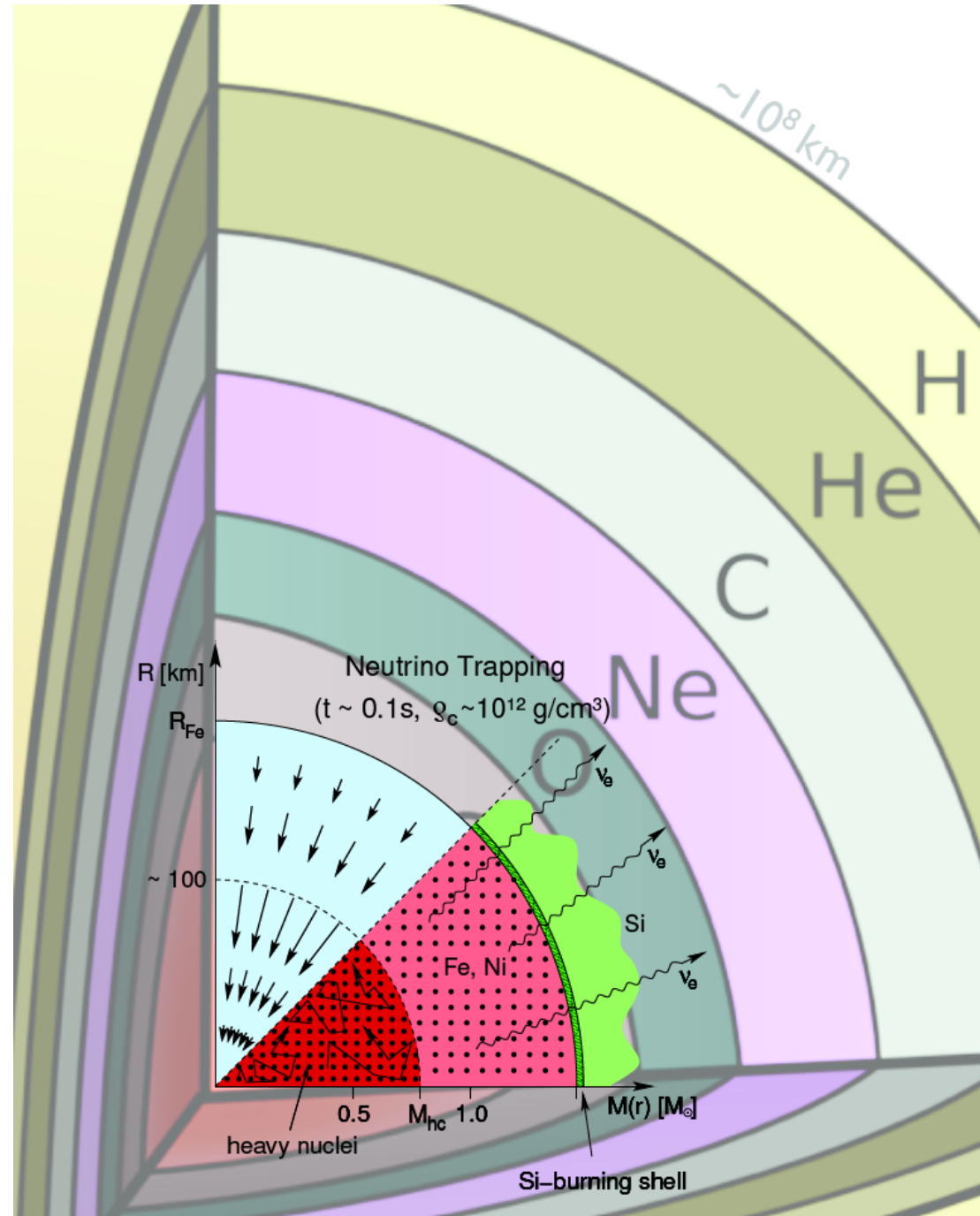
- 1 foe = 10⁵¹ ergs (~ 10⁴⁴ J)
- Sun will produce 1.2 foe over its lifetime

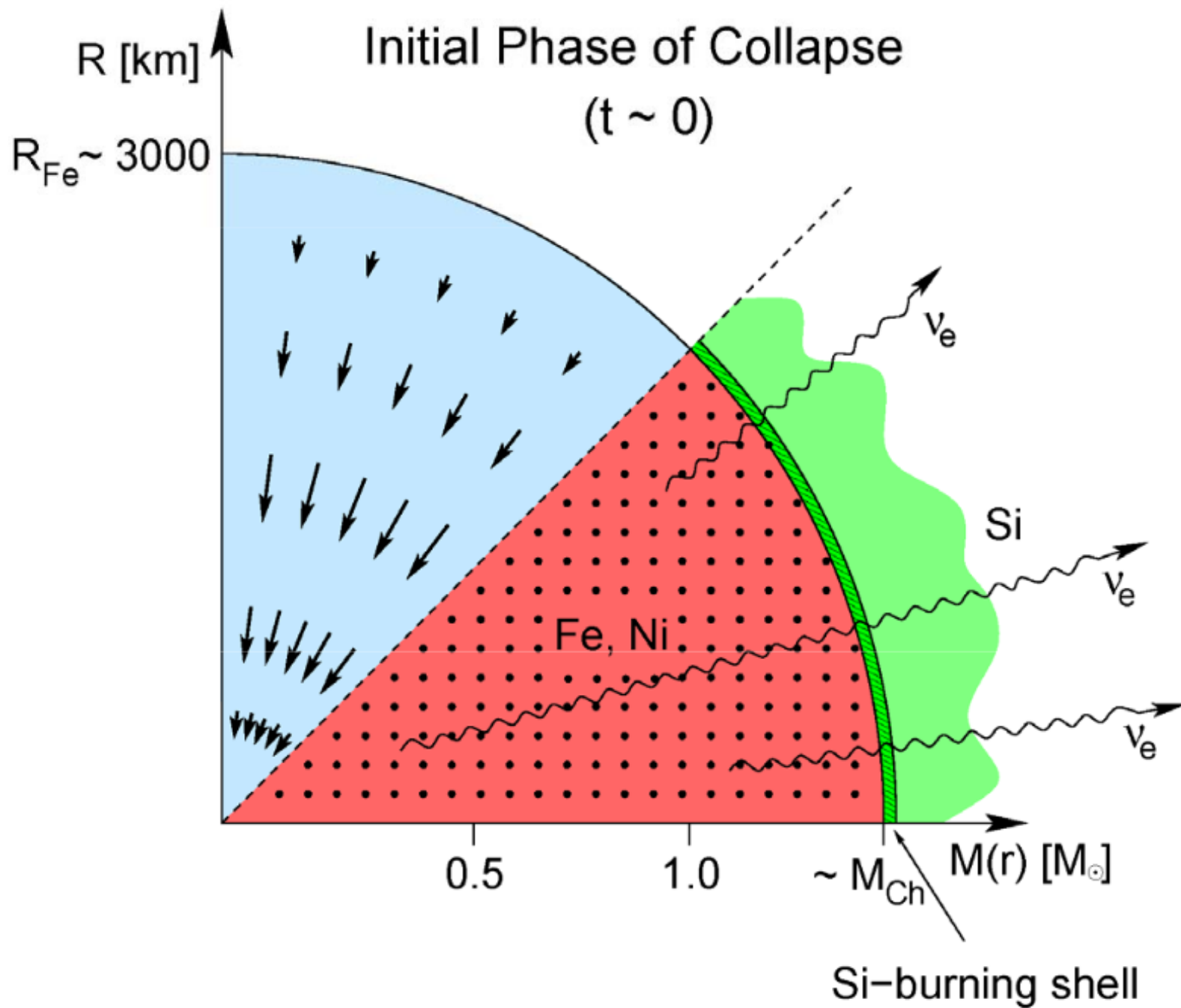


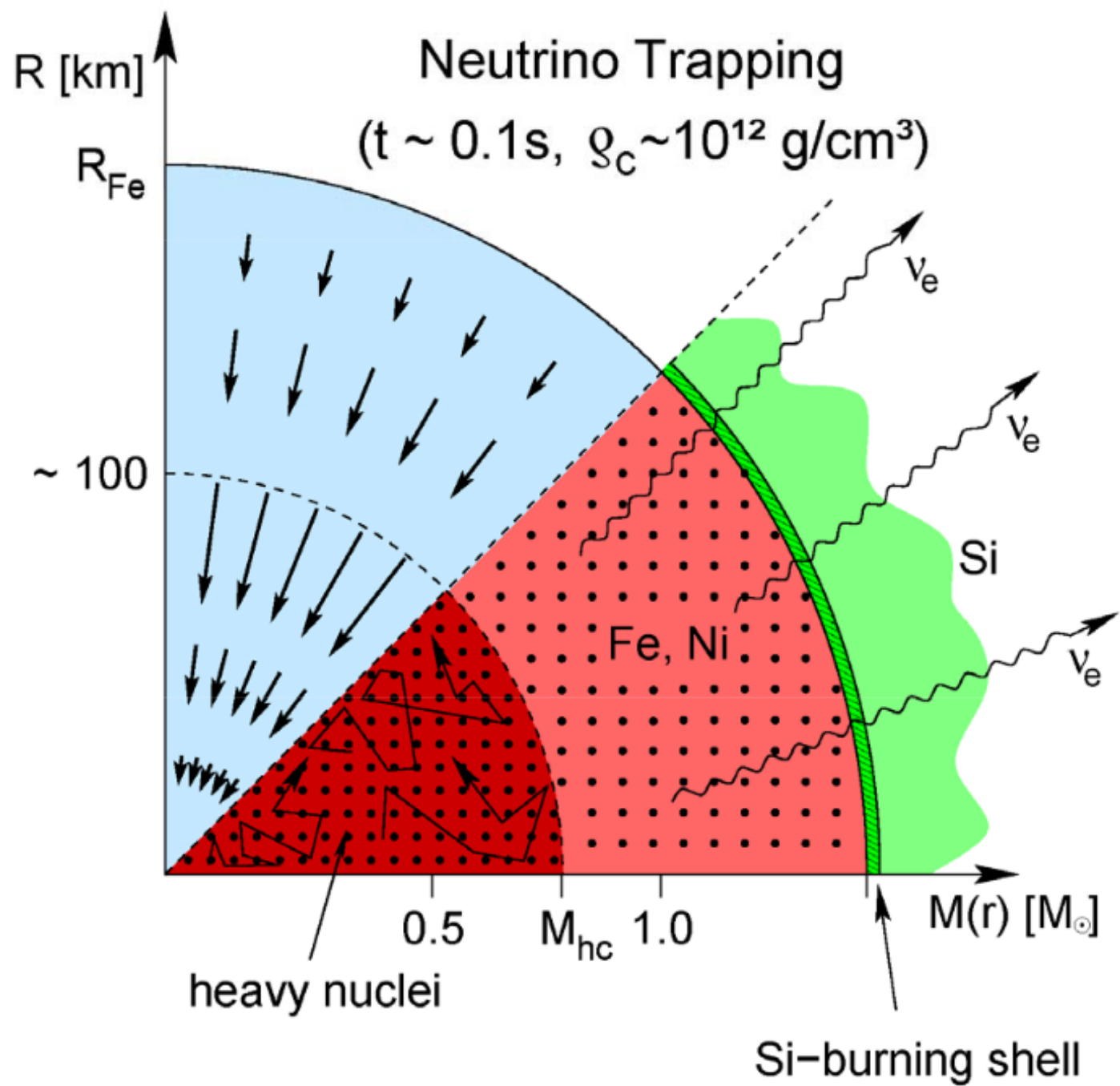
- Energy density of the core is equivalent to 1MT TNT per cubic micron
 - 99% of energy released is in the form of neutrinos
- ~1% is in the KE of the exploding matter
- ~0.01% is in light – and that's enough to make it as bright as an entire galaxy.

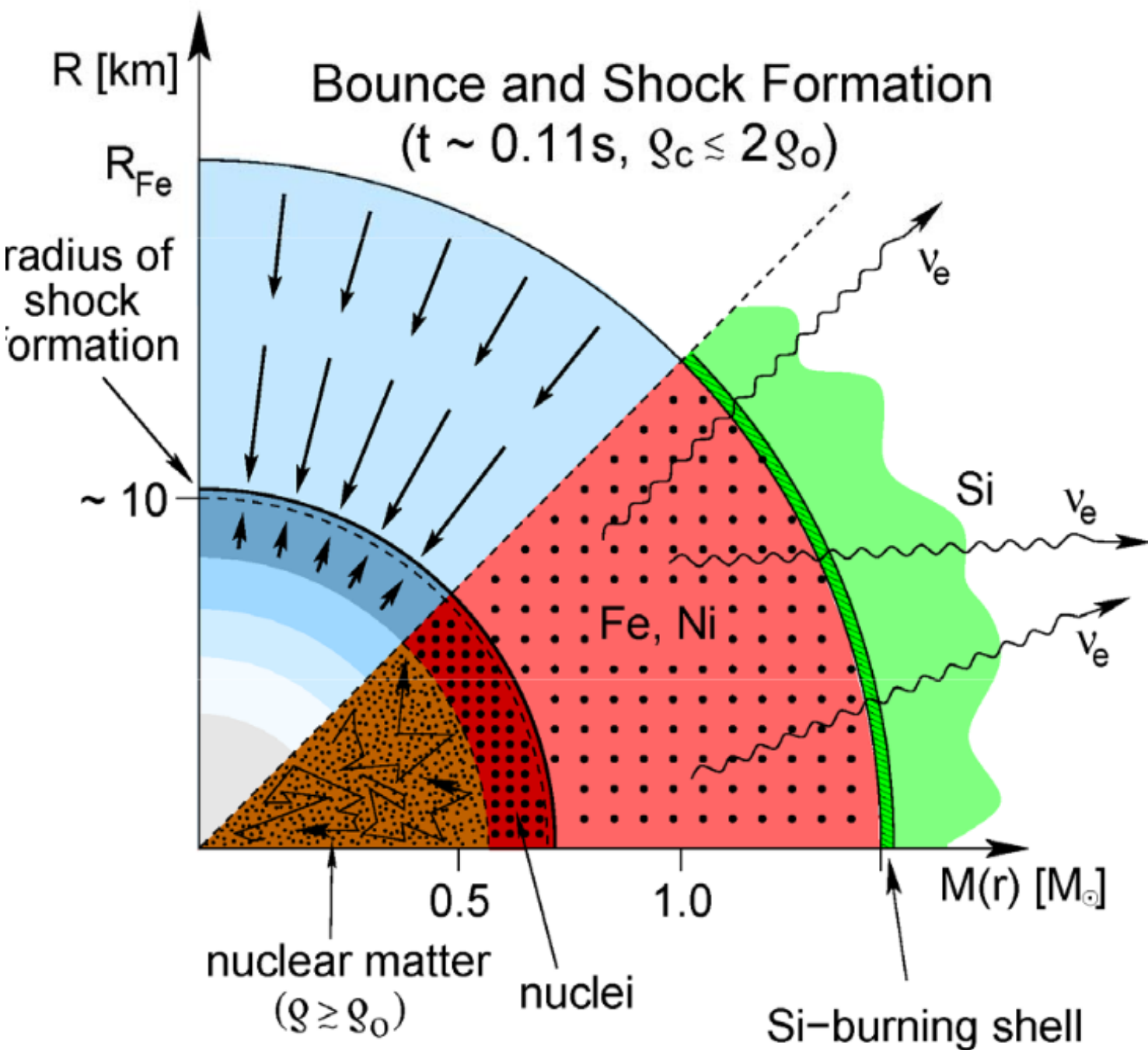
Upper bound on supernovae in Milky Way
13 supernovae per century 90% CL.

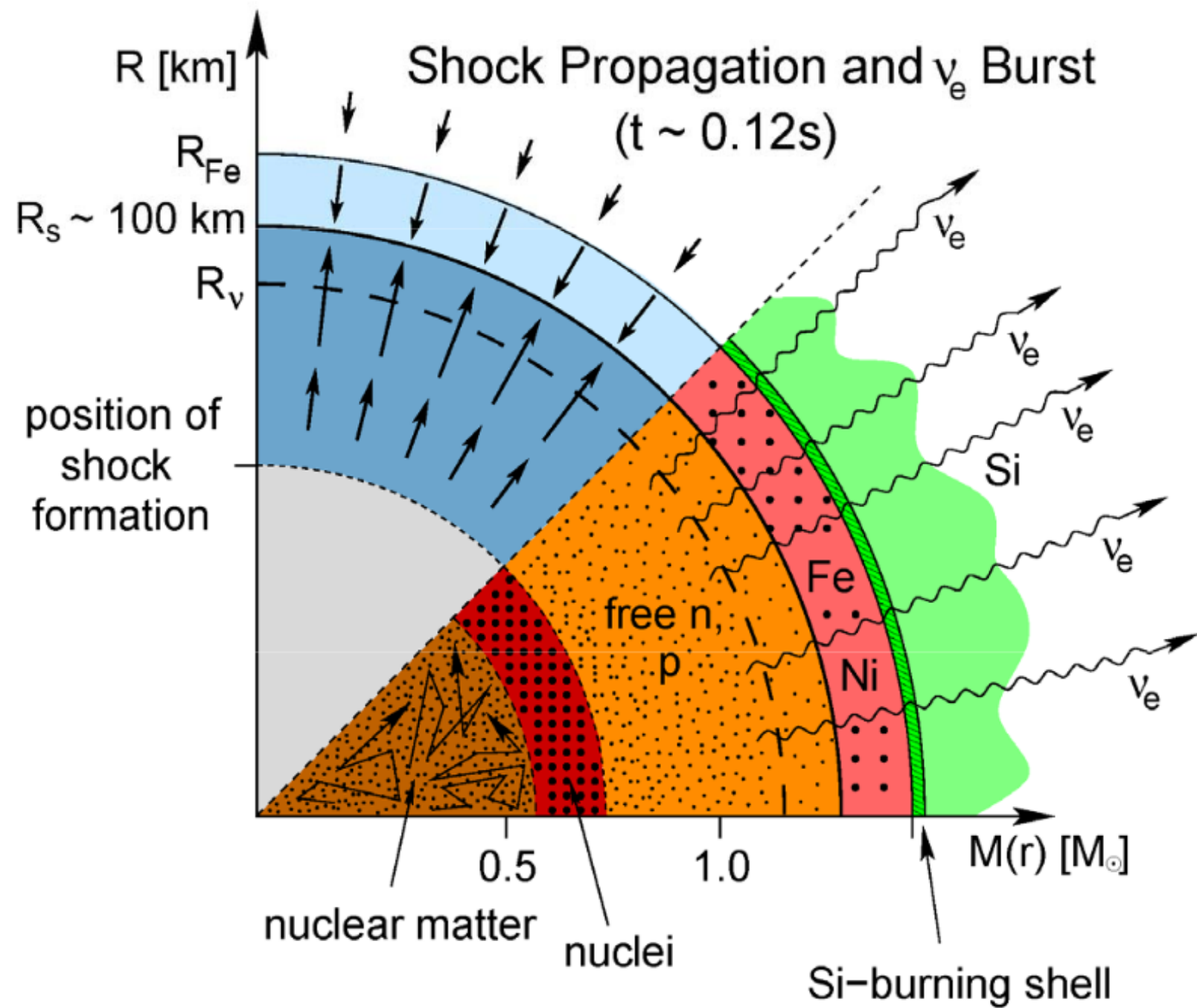
- Pressure is dominated by degenerate electrons
- Si burning at core surface increases Fe-core mass
- Chandrasekhar mass limit
- Collapse starts: gravitational energy is transformed into internal energy: $E \sim 10^{53}$ erg
- Most of the internal energy escapes in form of neutrinos
- $\rho_{\text{crit}} \sim 10_{12} \text{ g/cm}^3$ neutrinos are trapped

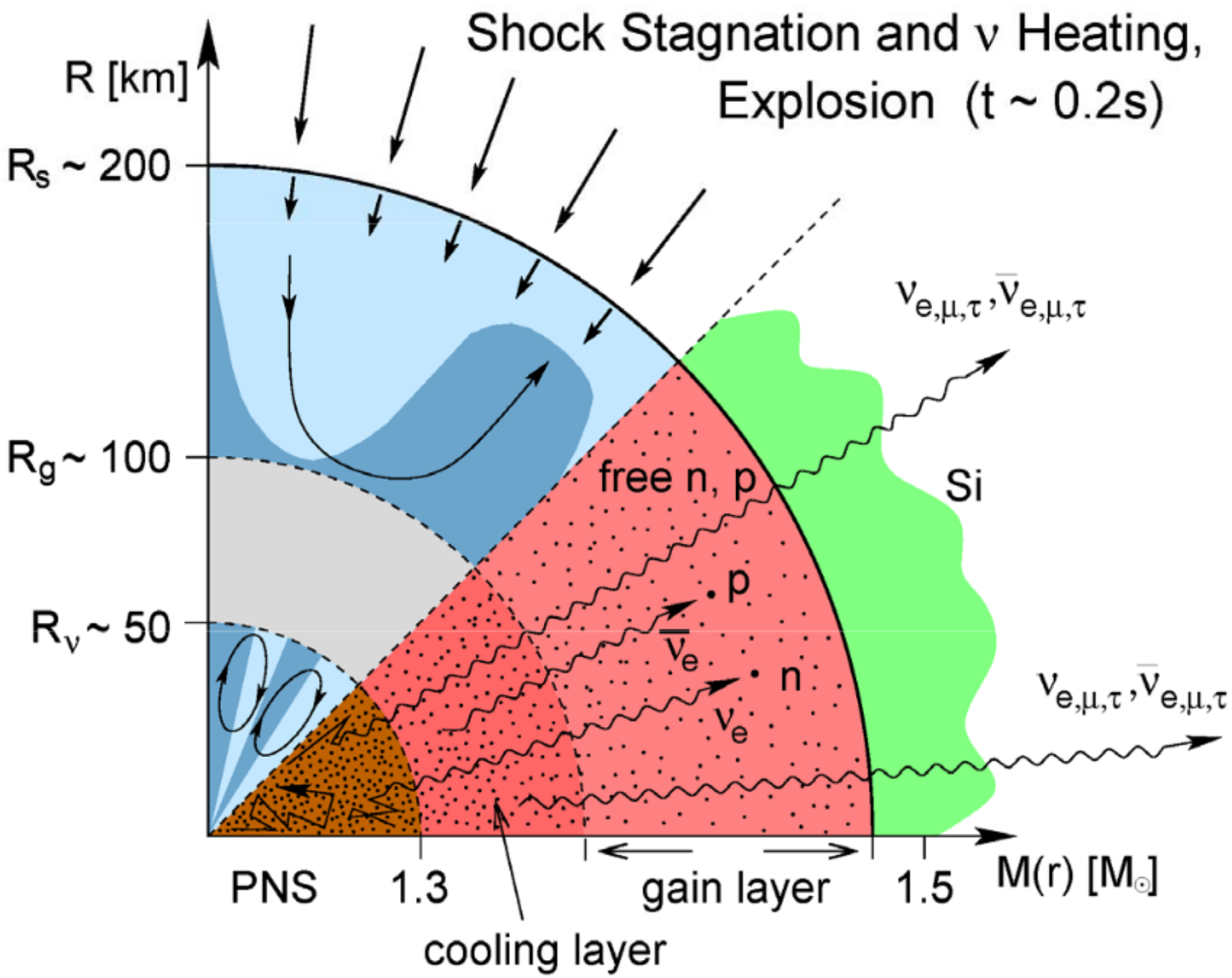


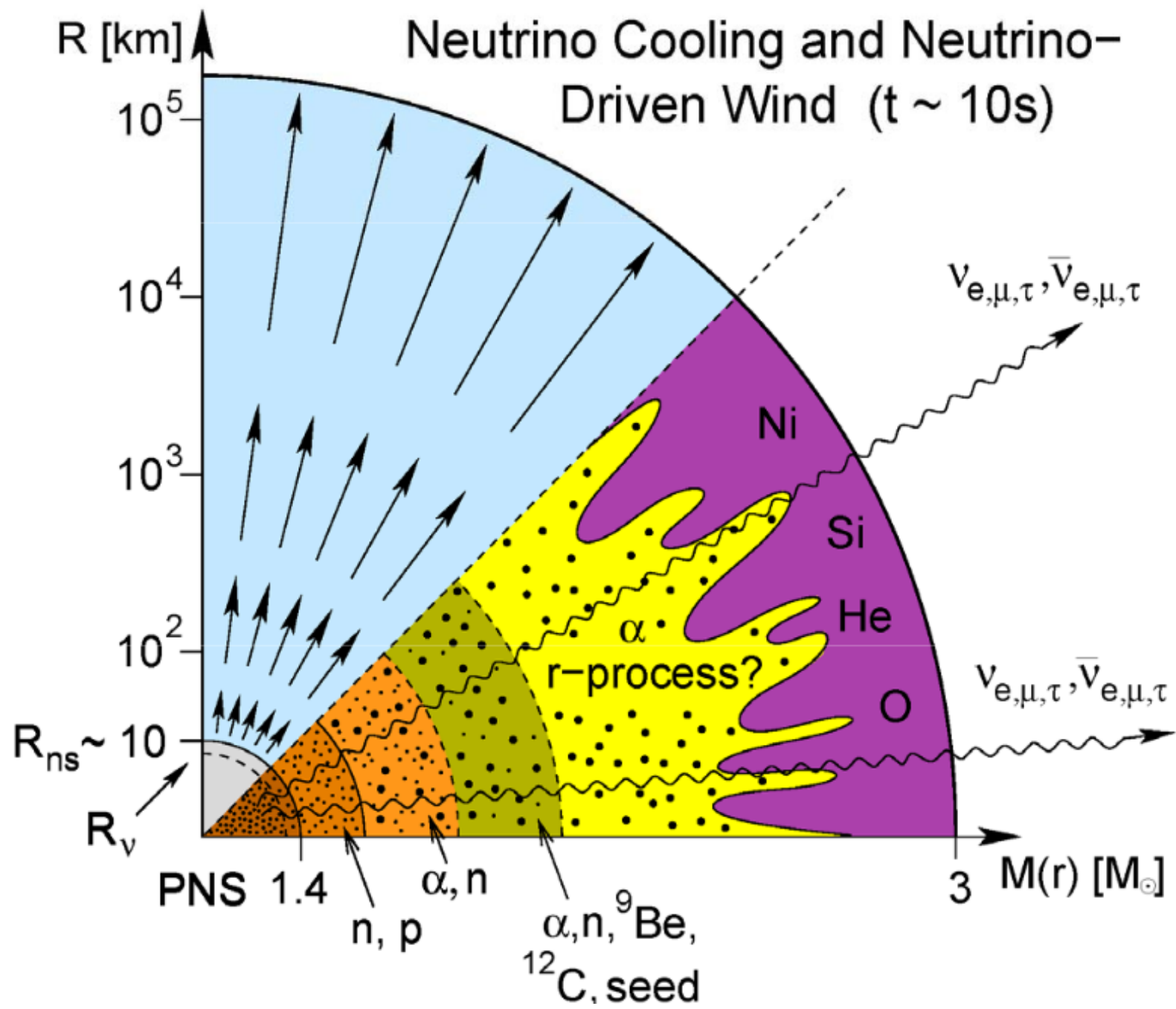




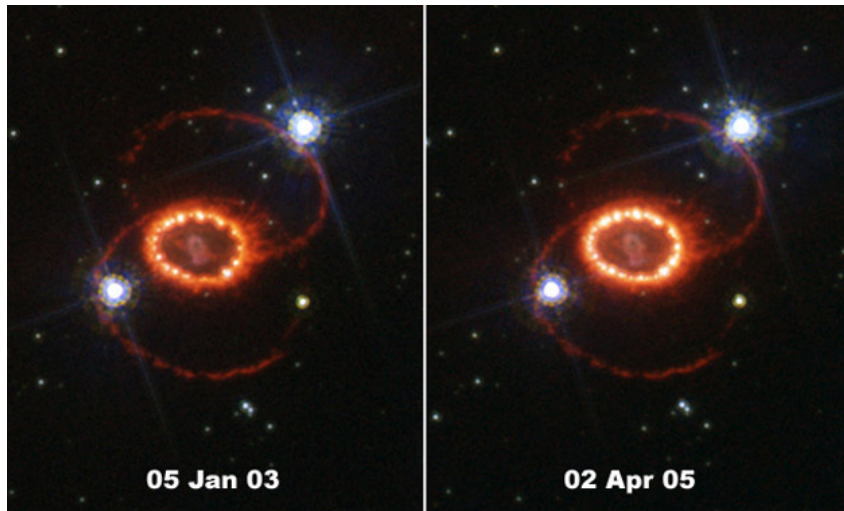
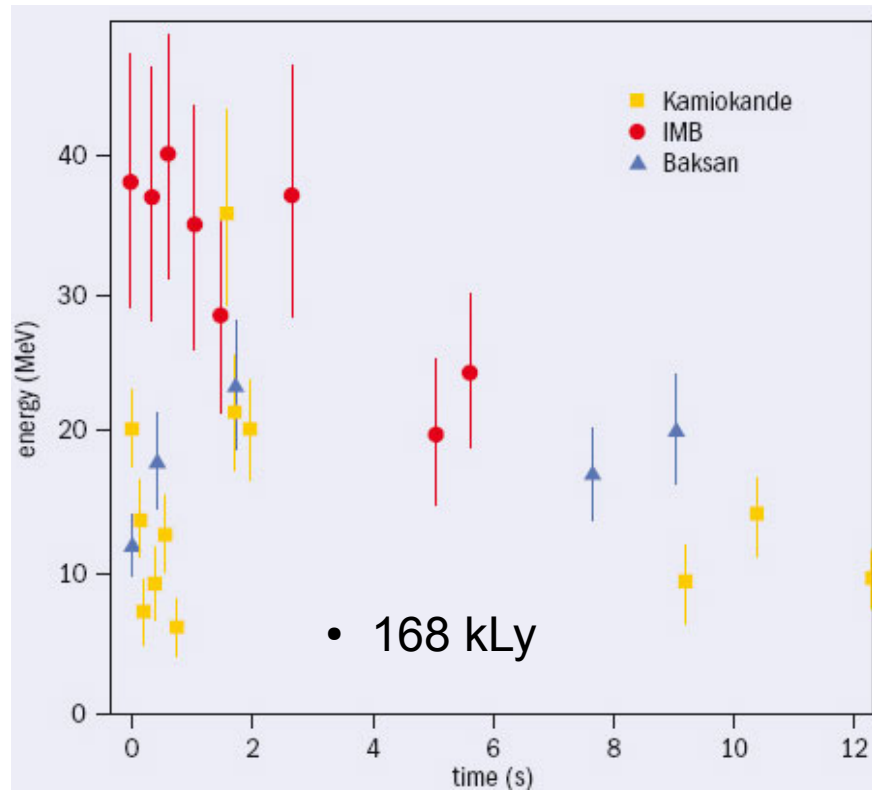








Sn1987a



The r-process

- Occurs in core collapse supernovae
- Responsible for about half elements with $Z > 26$
- Neutron density up to 10^{30} ncm^{-3}

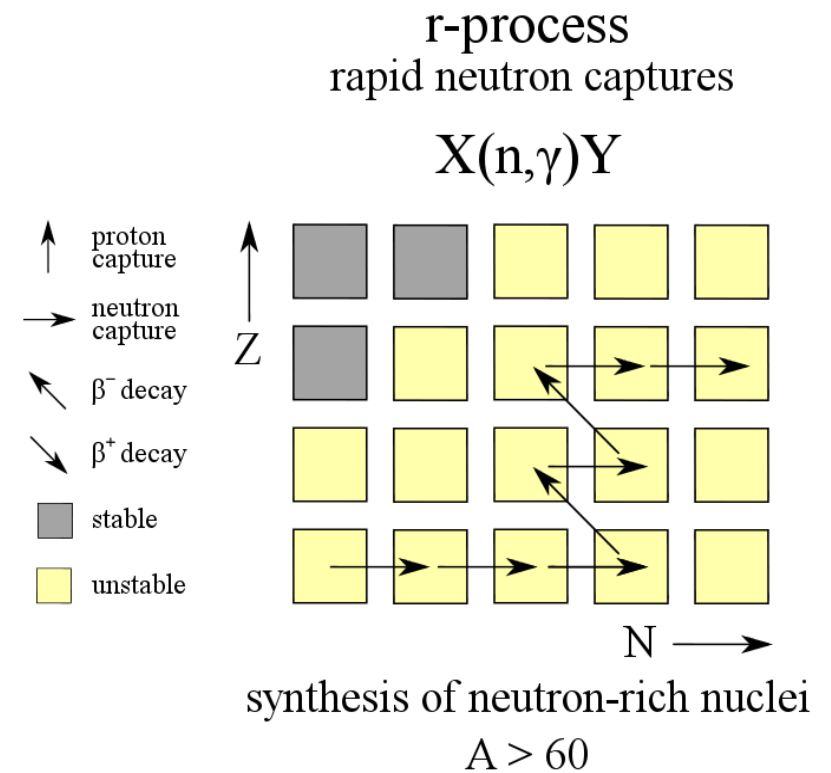
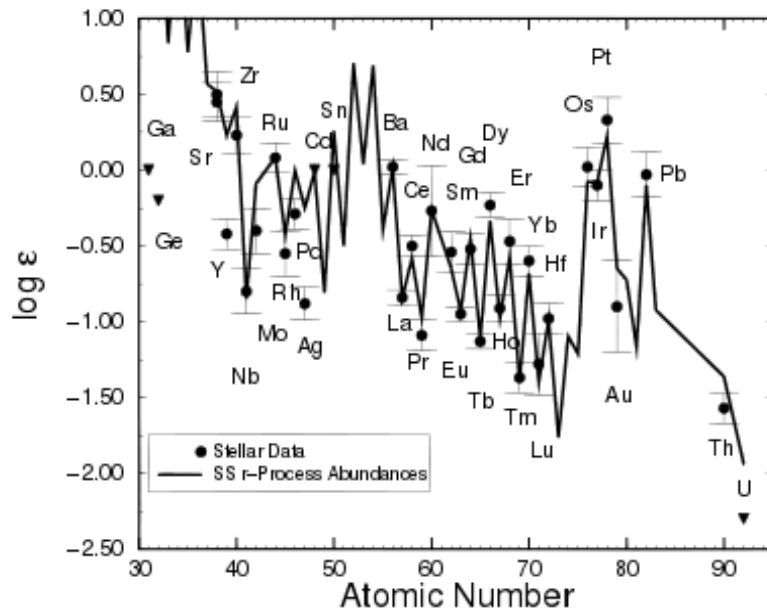
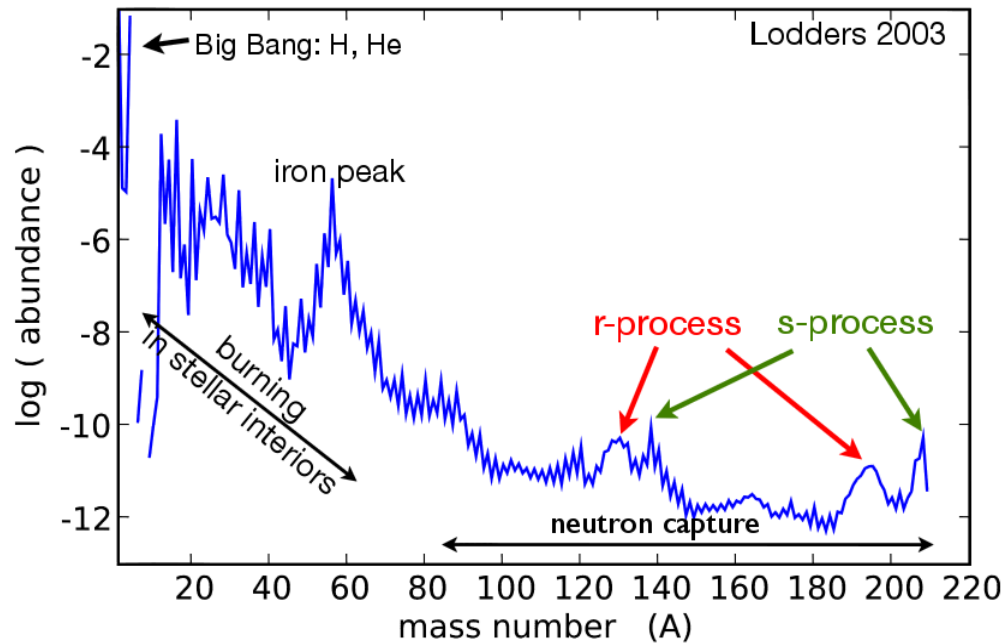


Fig. 1.2. Comparison of the observed n -capture abundances in CS 22892-052 from Sneden et al. (2003) and the solar system r -process abundance curve. Upper limits indicated by inverted triangles.

s,r-processes



- Solar system abundances produced by r and s processes. Abundance peaks are caused by minimal n-capture rates at magic numbers (corresponding to full neutron shells).
- Because the r-process carries nuclei farther from the valley of stability than does the s-process, it encounters each closed shell at slightly lower mass number.
- Hence the r-process peaks are offset to lower A. The two processes really have contributed about equally to the solar system's inventory of heavy elements. (Adapted from ref.)

s-process: slow neutron capture

r-process: rapid neutron capture

